

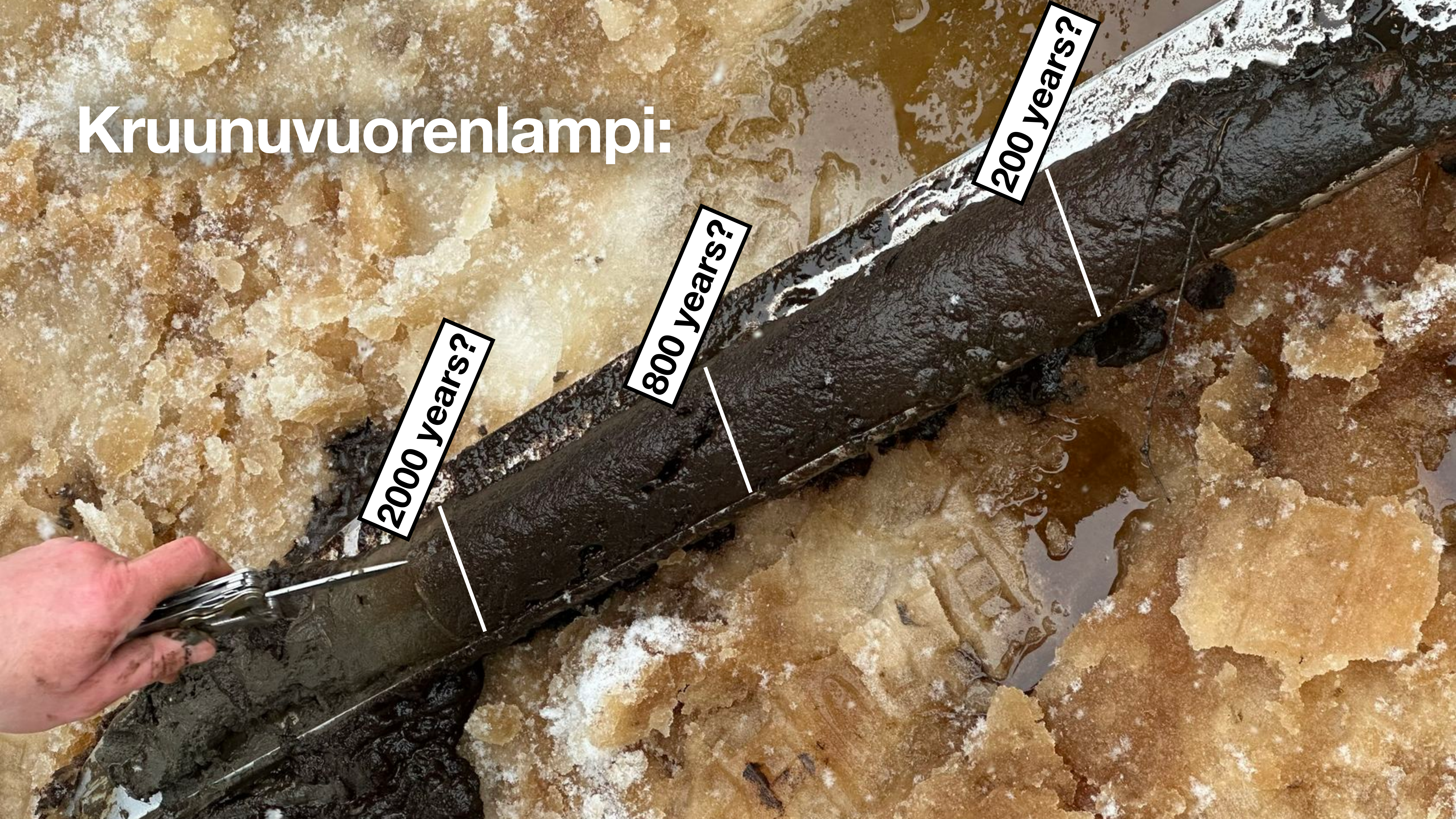
Sediment Dating

Past Environmental Change ECGS-068

Ludwig Strötz
ludwig.strotz@helsinki.fi

02.02.2026

Kruunuvuorenlampi:



Law of Superposition

- **Newer material stacks atop older material** (if undisturbed)



Nicolaus Steno (1677)

younger



older





Djúpivogur, Iceland



Karwendel, Austria

Geochronological Methods

Absolute

Geochemical



^{210}Pb



^{137}Cs , ^{241}Am



^{14}C

Other
radioactive
isotopes

Physical



Varves



Luminescence

Dendrochronology

Corals

Lichen in Rocks

Relative



Paleomagnetism



Tephra



Fossils

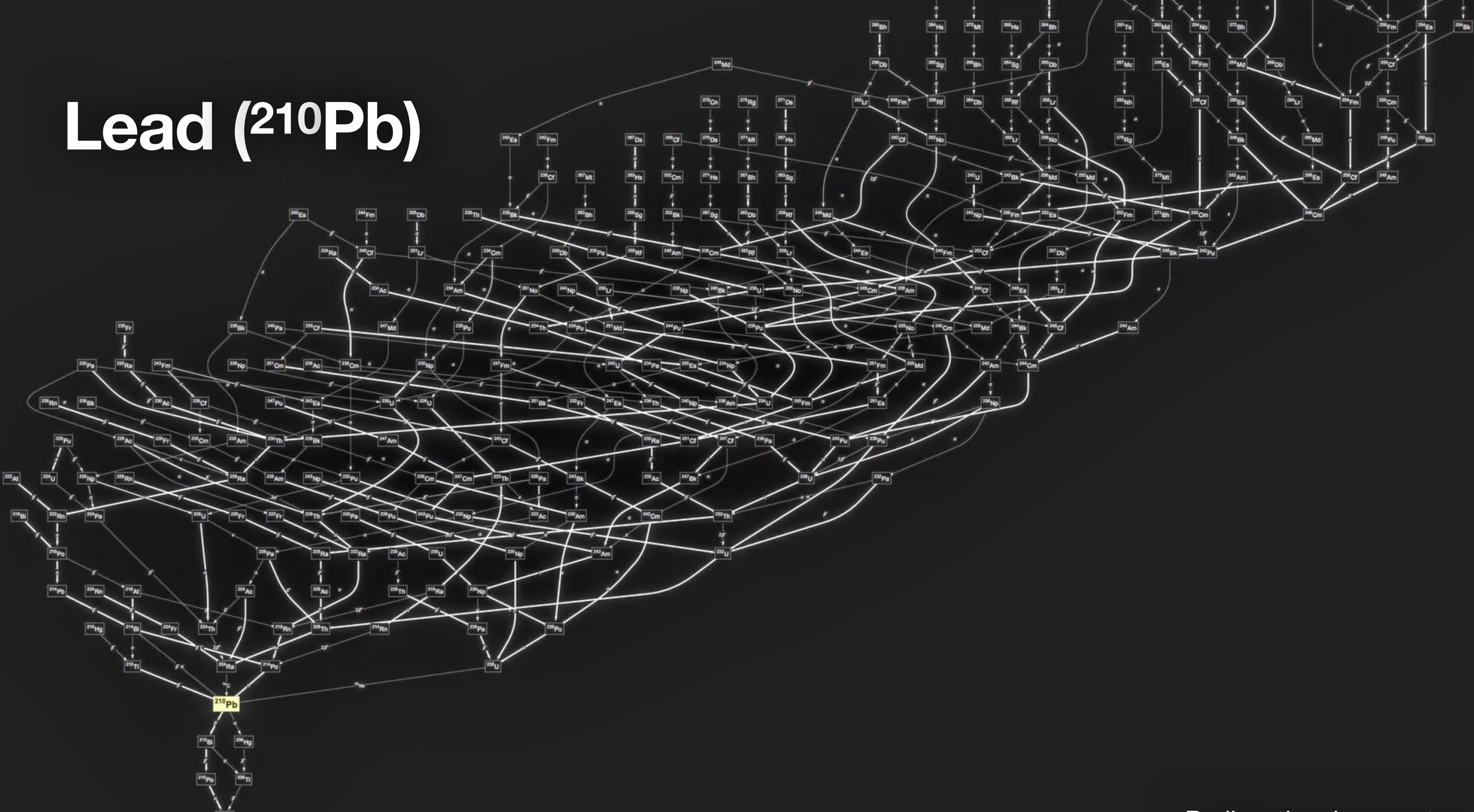


$\delta^{18}\text{O}$



SCPs

Lead (^{210}Pb)

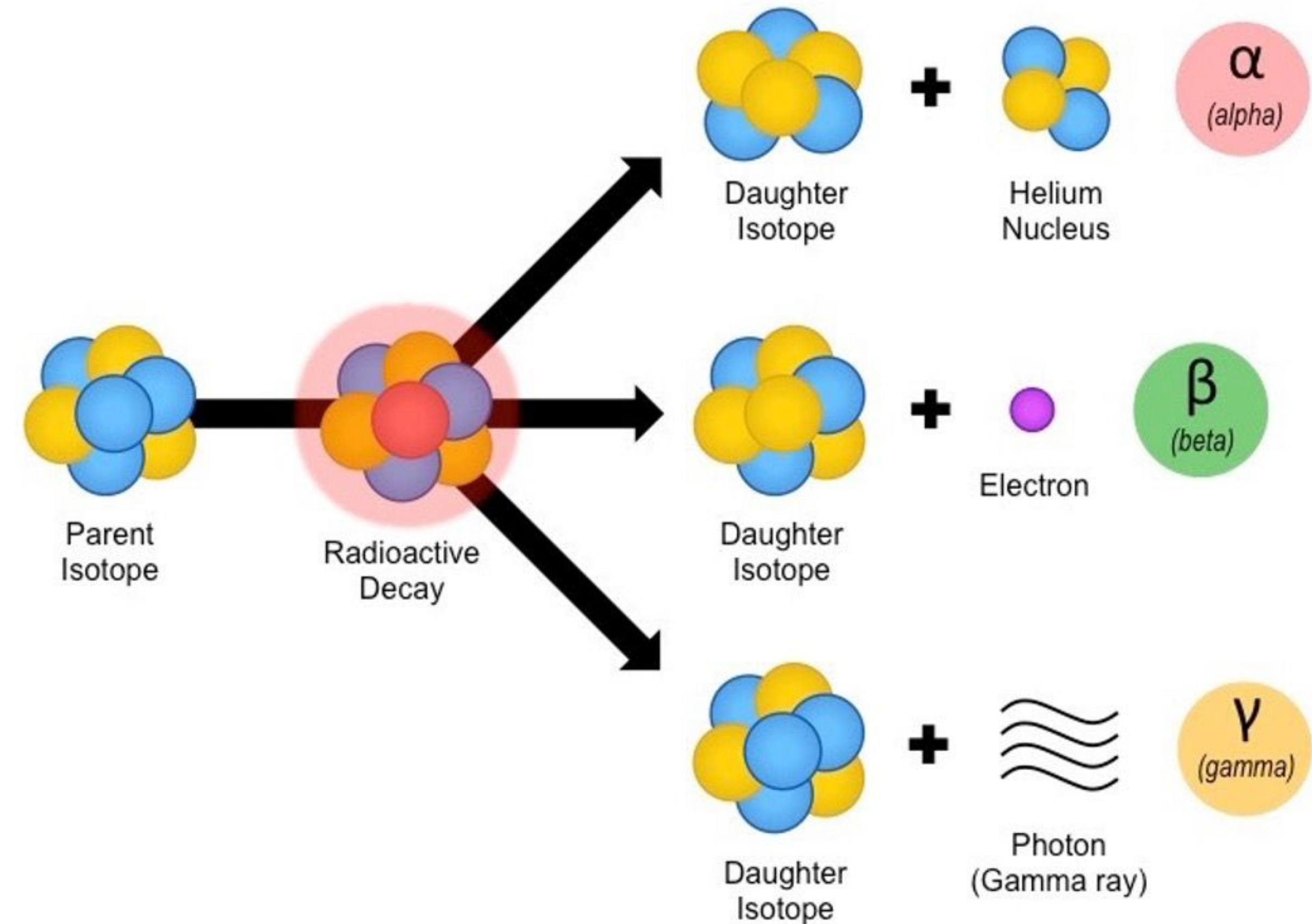


Radioactive decay network



Radioactive Isotopes

- Isotopes are either **stable** or **radioactive**
- Unstable isotopes decay, forming new elements, until they reach stability
- **Radioactivity is a random process**, what we calculate is the probability that a nuclide will decay in a given time



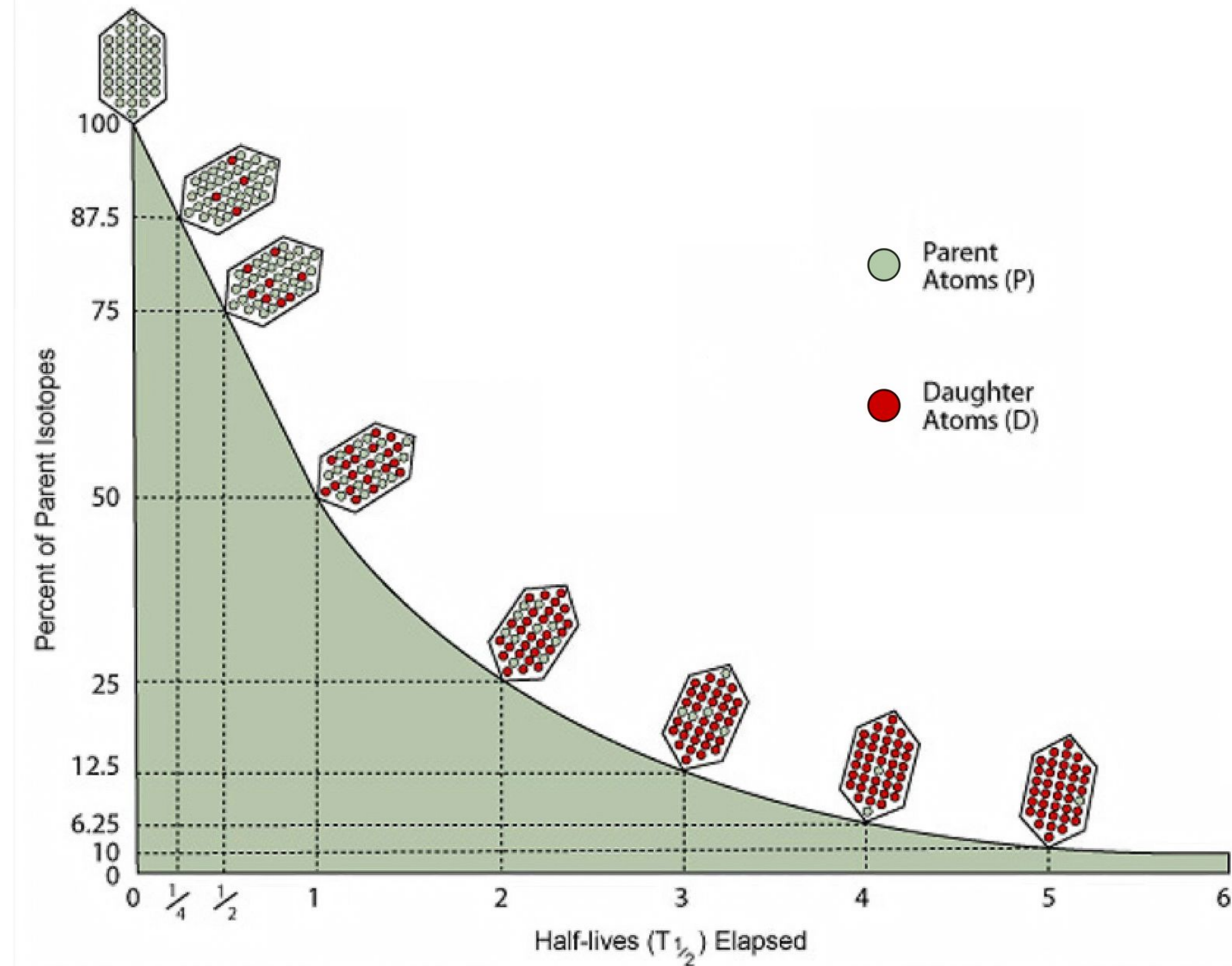


Radioactive Decay

- The **half-life** determines the **speed of radioactive decay**

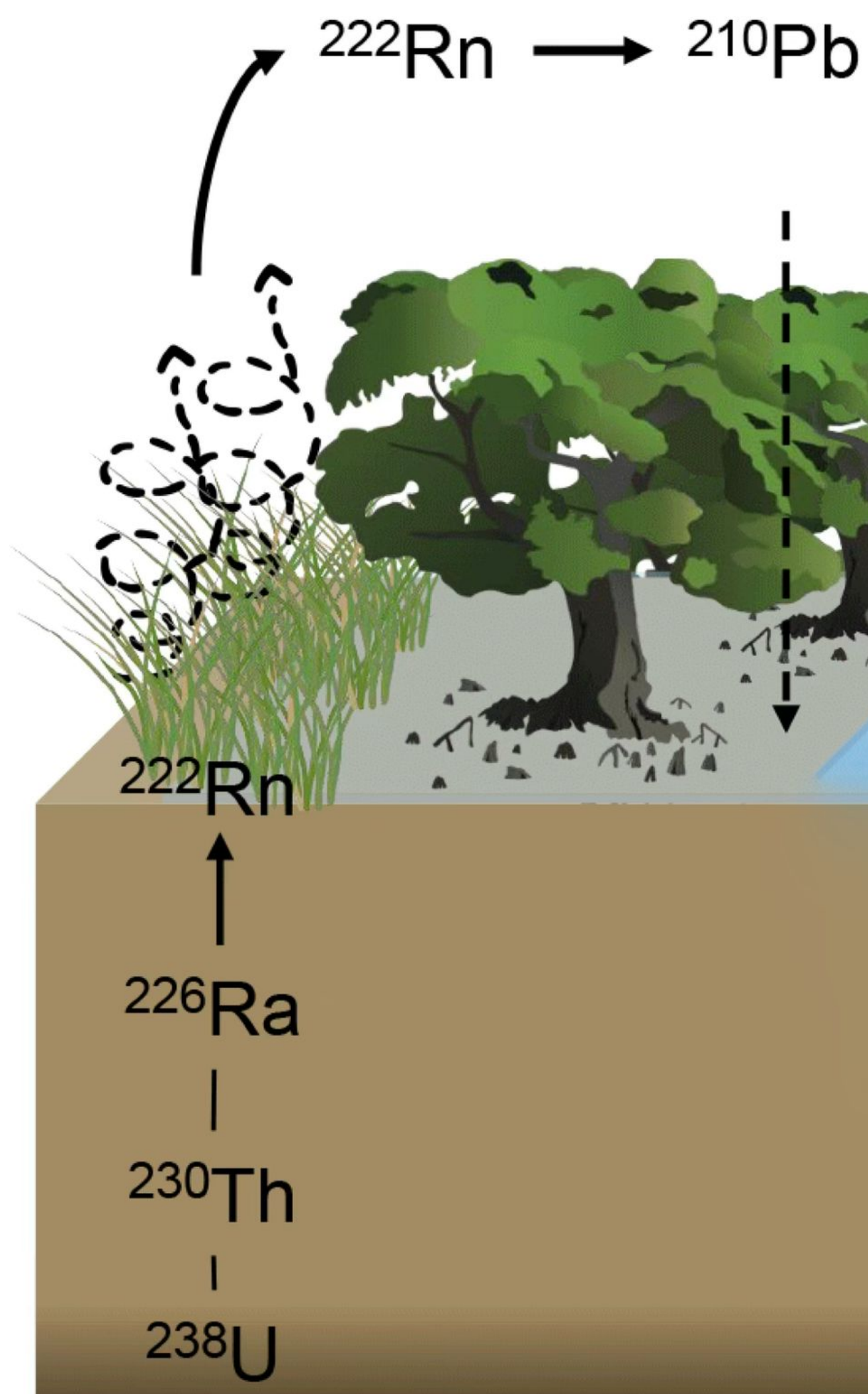
$$N = N_0 e^{-\lambda t}$$

$$t_{1/2} = \frac{\ln 2}{\lambda}$$





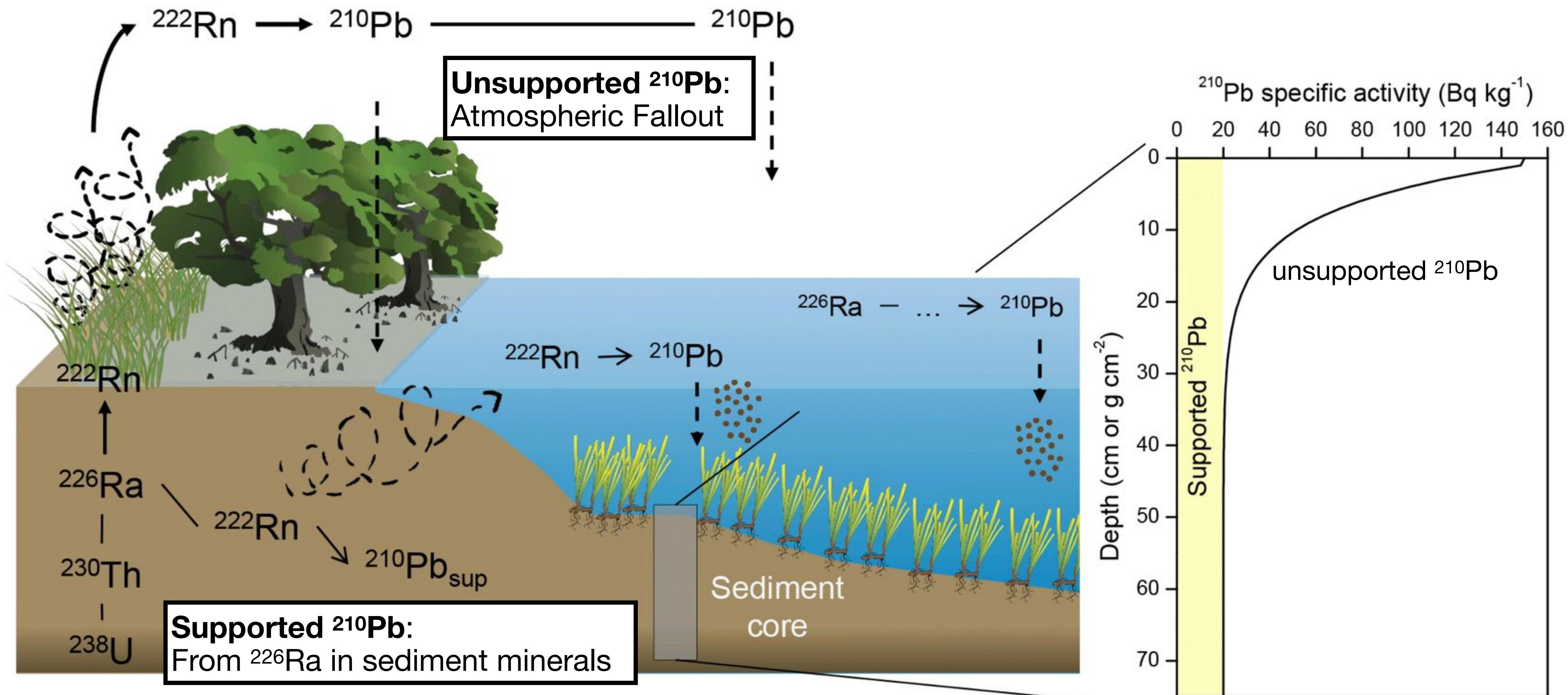
^{238}U Decay Series



type of radiation	nuclide	half-life
	uranium—238	4.5×10^9 years
α	thorium—234	24.5 days
β	protactinium—234	1.14 minutes
β	uranium—234	2.33×10^5 years
α	thorium—230	8.3×10^4 years
α	radium—226	1590 years
α	radon—222	3.825 days
α	polonium—218	3.05 minutes
α	lead—214	26.8 minutes
β	bismuth—214	19.7 minutes
β	polonium—214	1.5×10^{-4} seconds
α	lead—210	22 years
β	bismuth—210	5 days
β	polonium—210	140 days
α	lead—206	stable

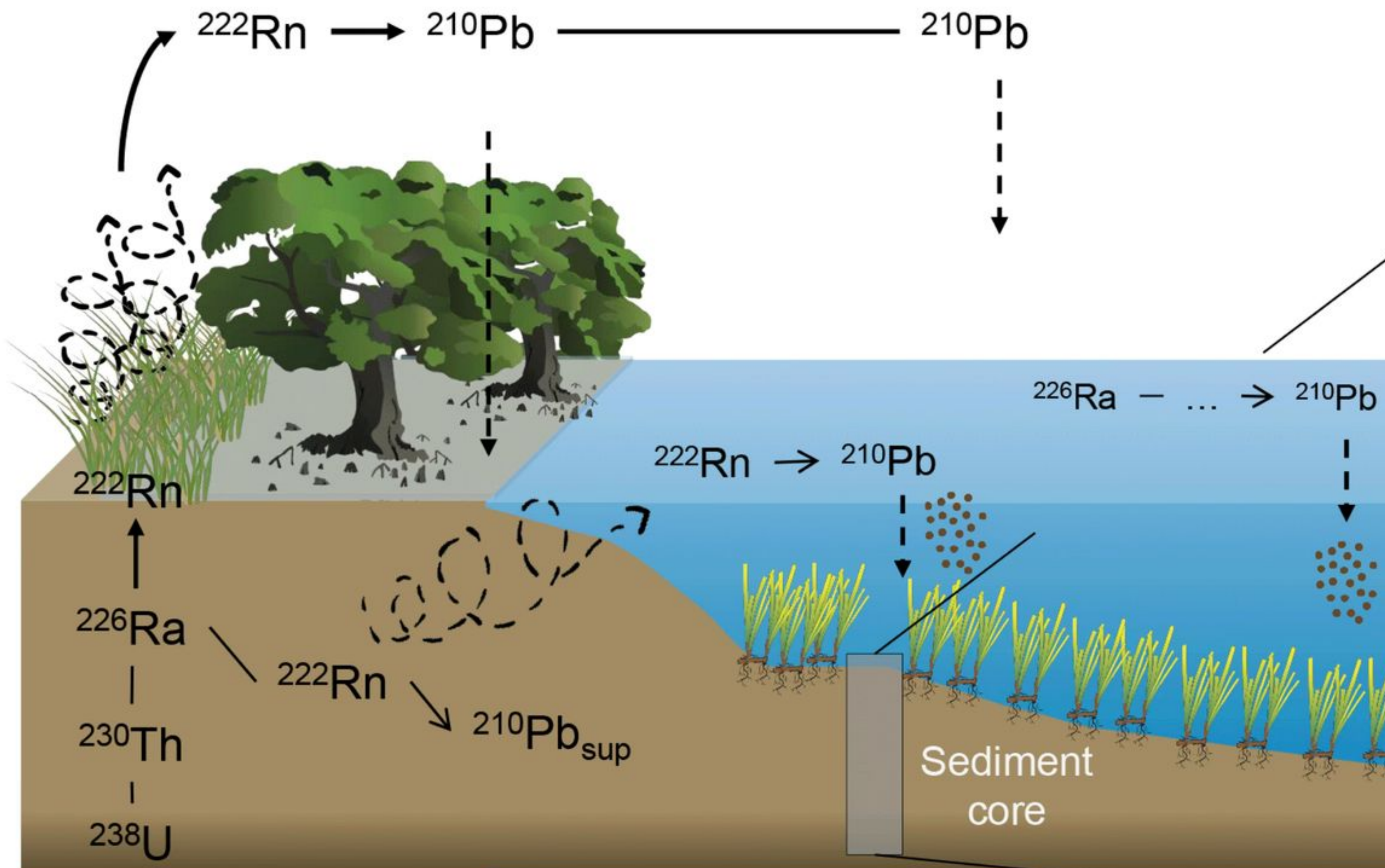


Lead (^{210}Pb)





Lead (^{210}Pb)

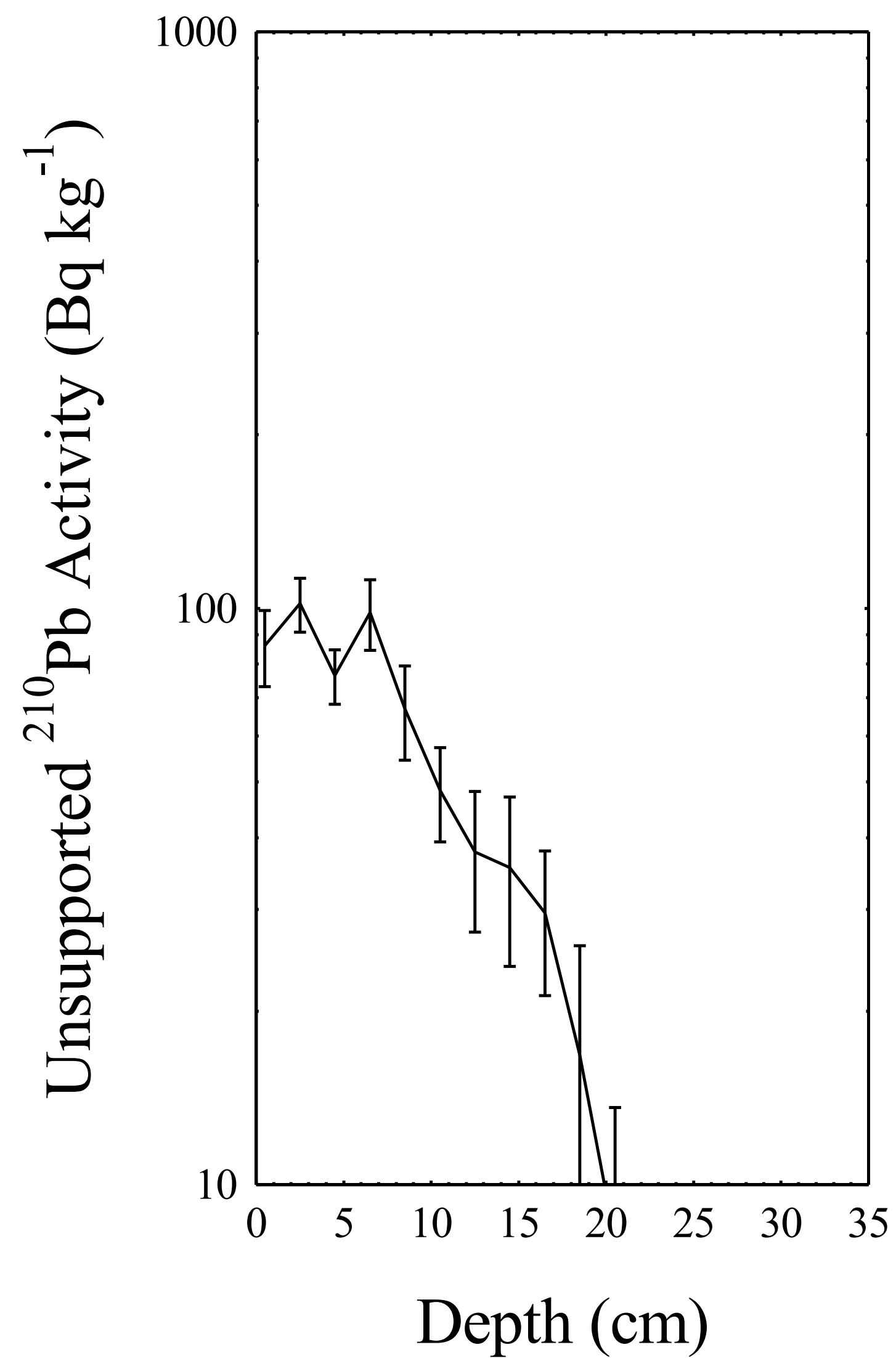
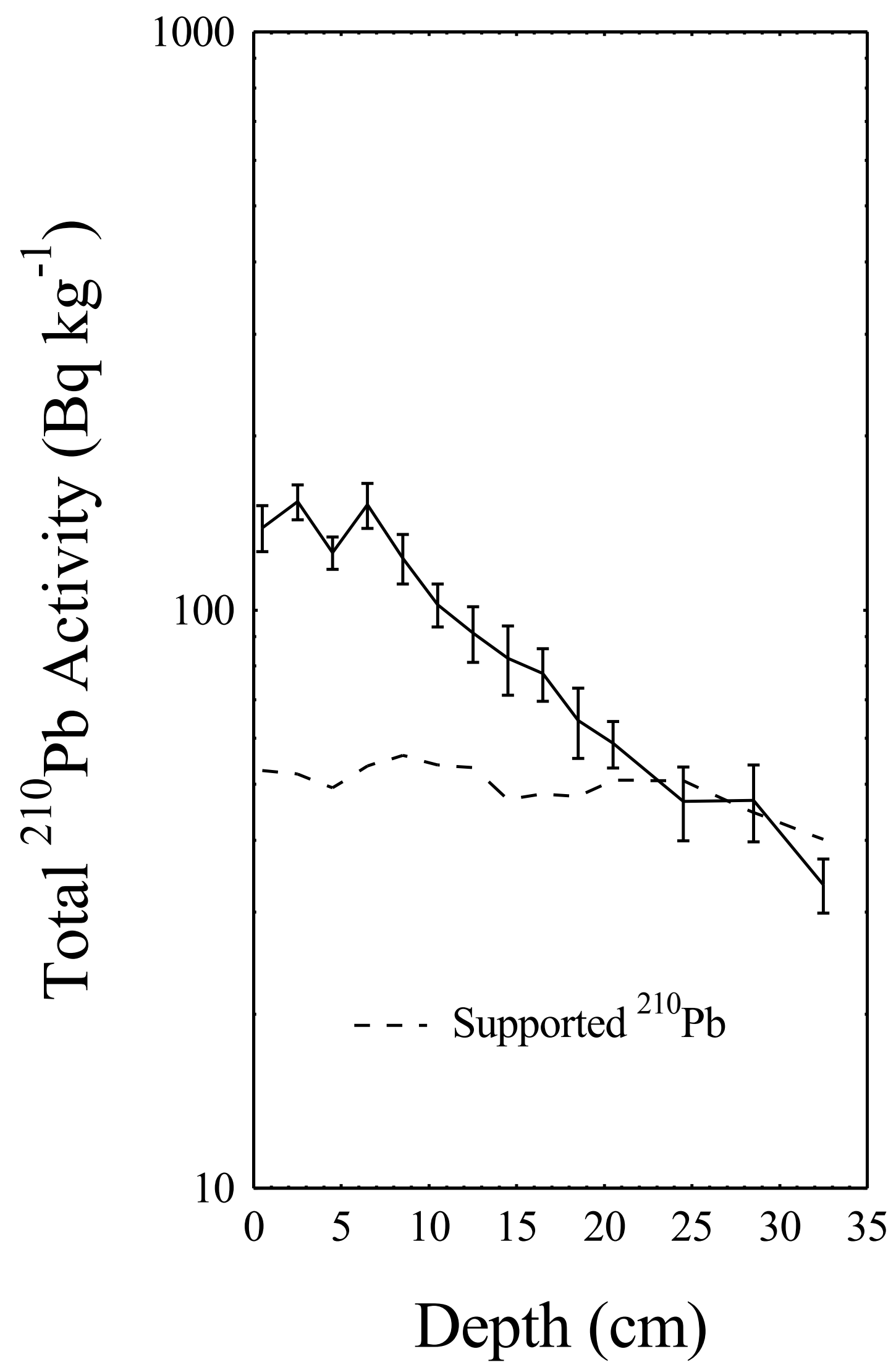


Assumptions:

- **Constant ^{210}Pb deposition** from atmosphere
- **^{210}Pb delay in water is constant**
- **No redistribution** in sediments



Lead (^{210}Pb)

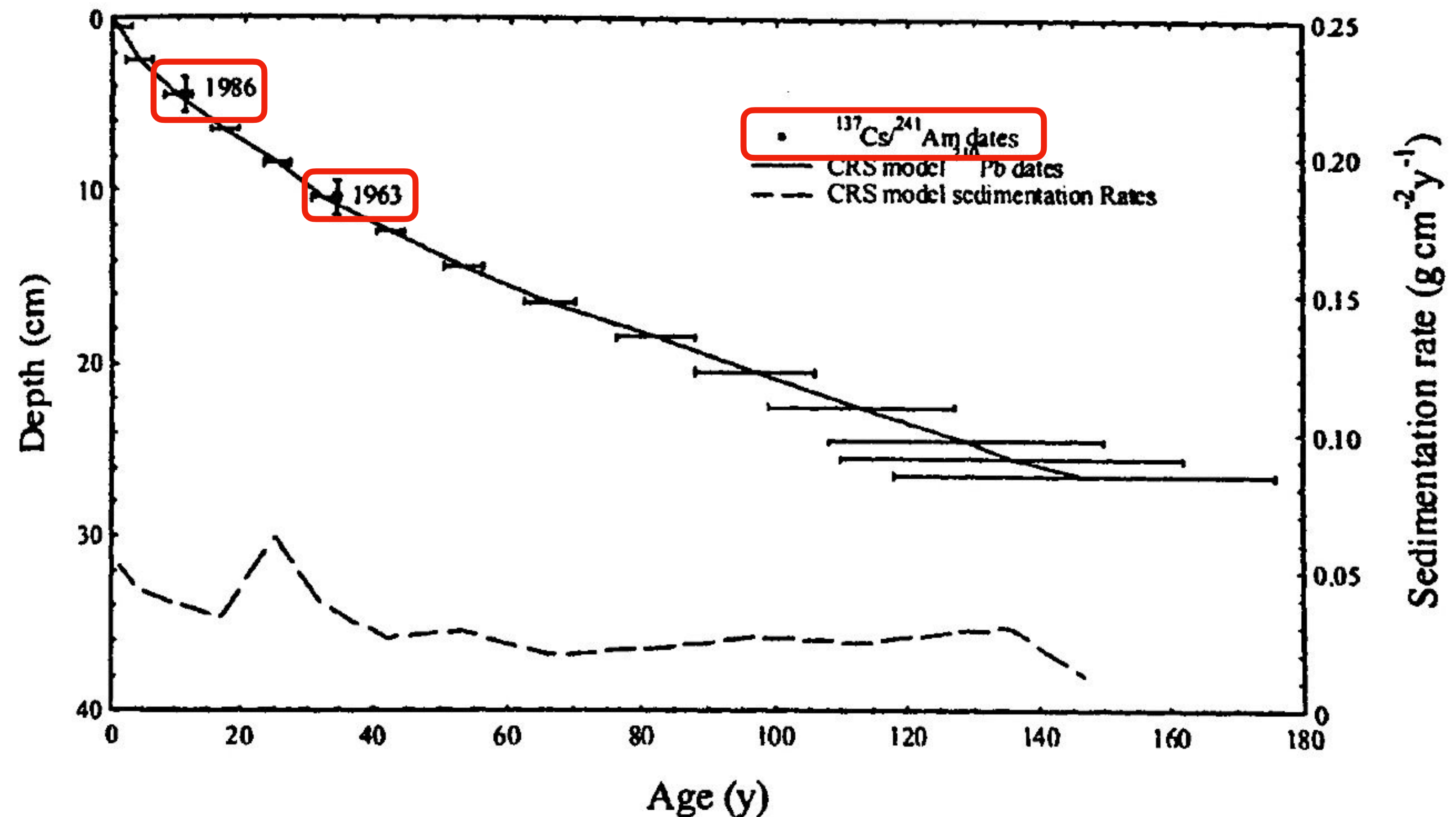




Lead (^{210}Pb)

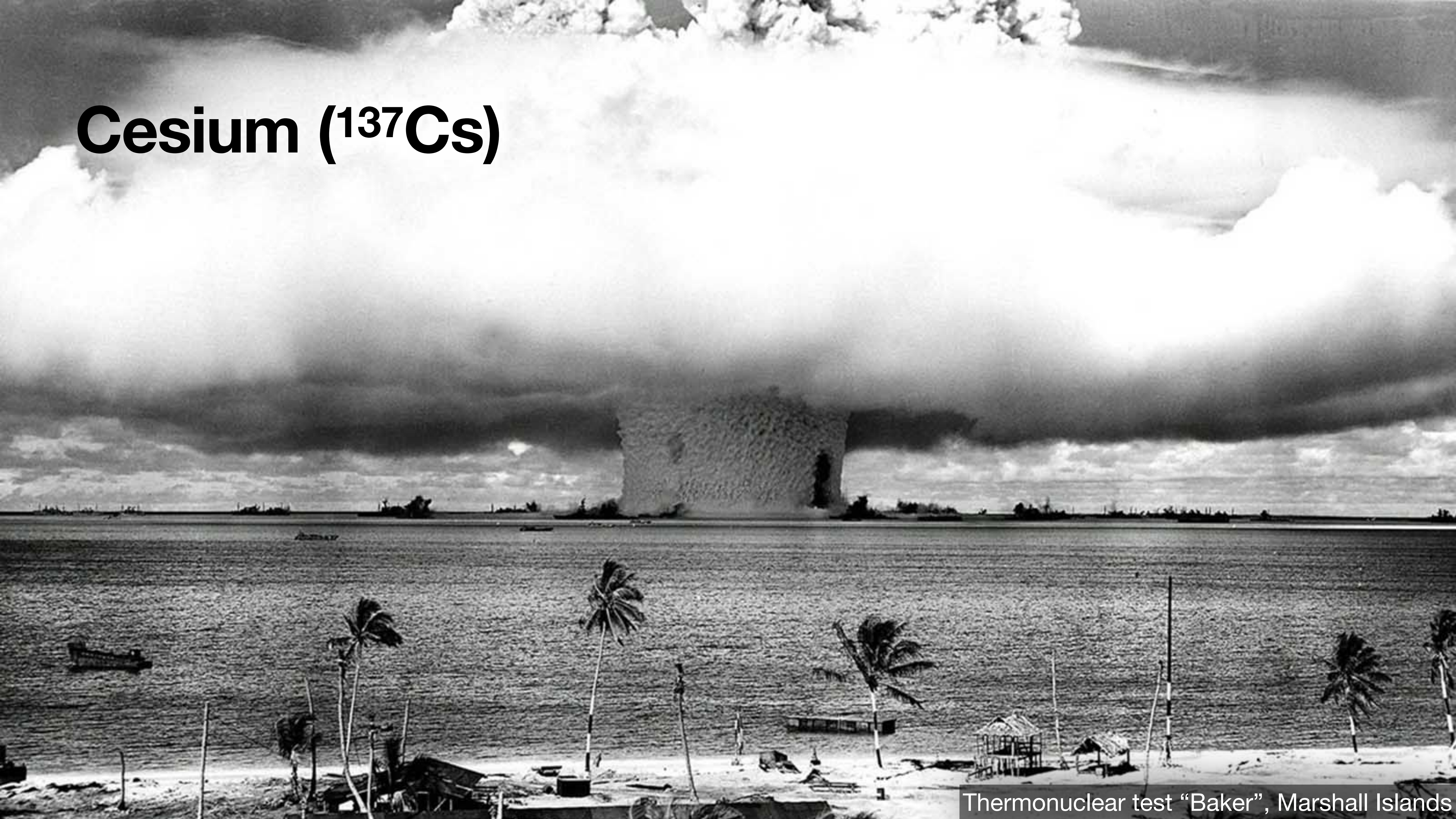
Models:

- Constant Rate of Supply of Sediment Influx (**CRS**)
- Constant Initial Concentration (**CIC**)



Appleby (2001)

Cesium (^{137}Cs)

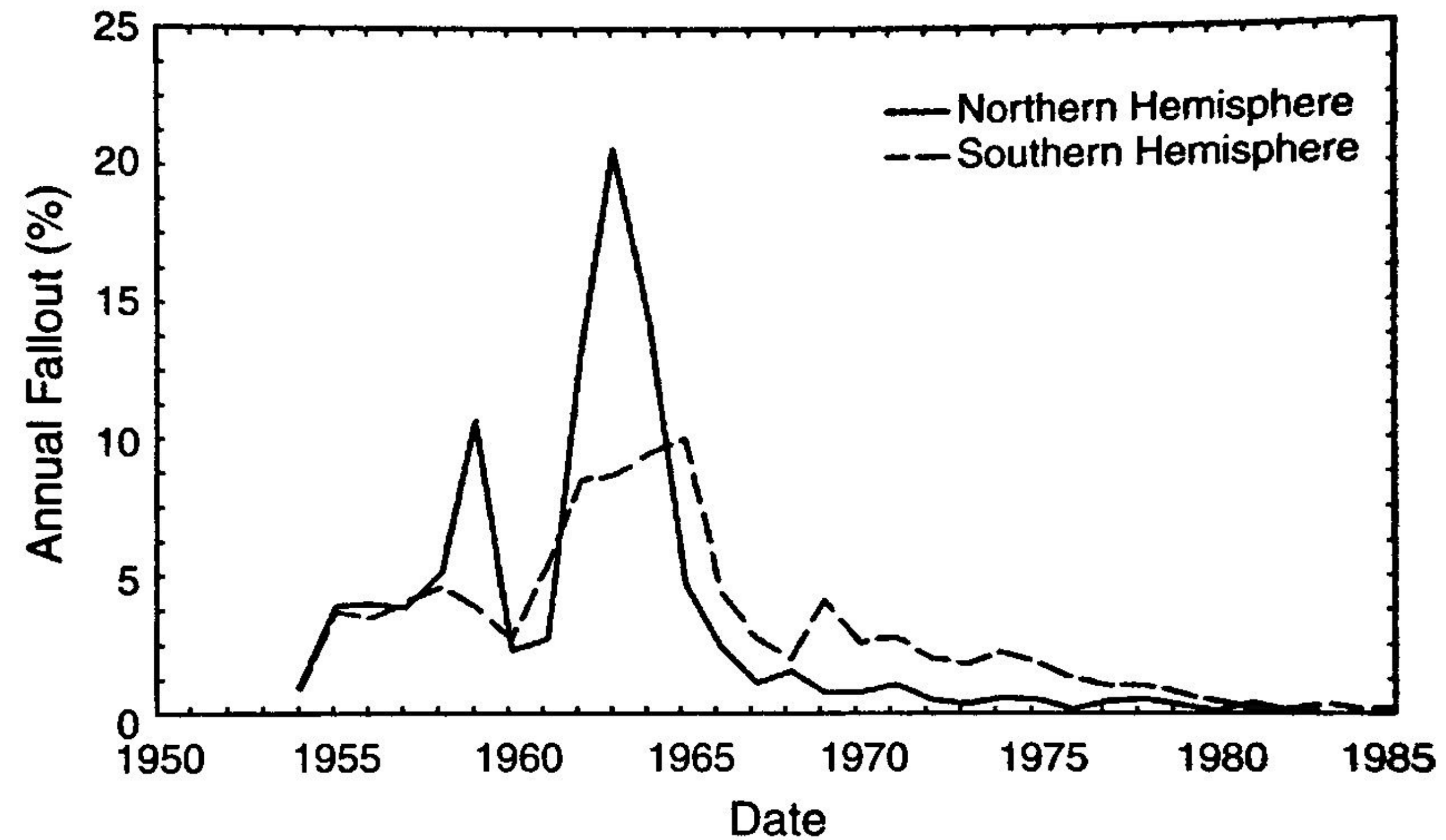


Thermonuclear test "Baker", Marshall Islands



Cesium (^{137}Cs)

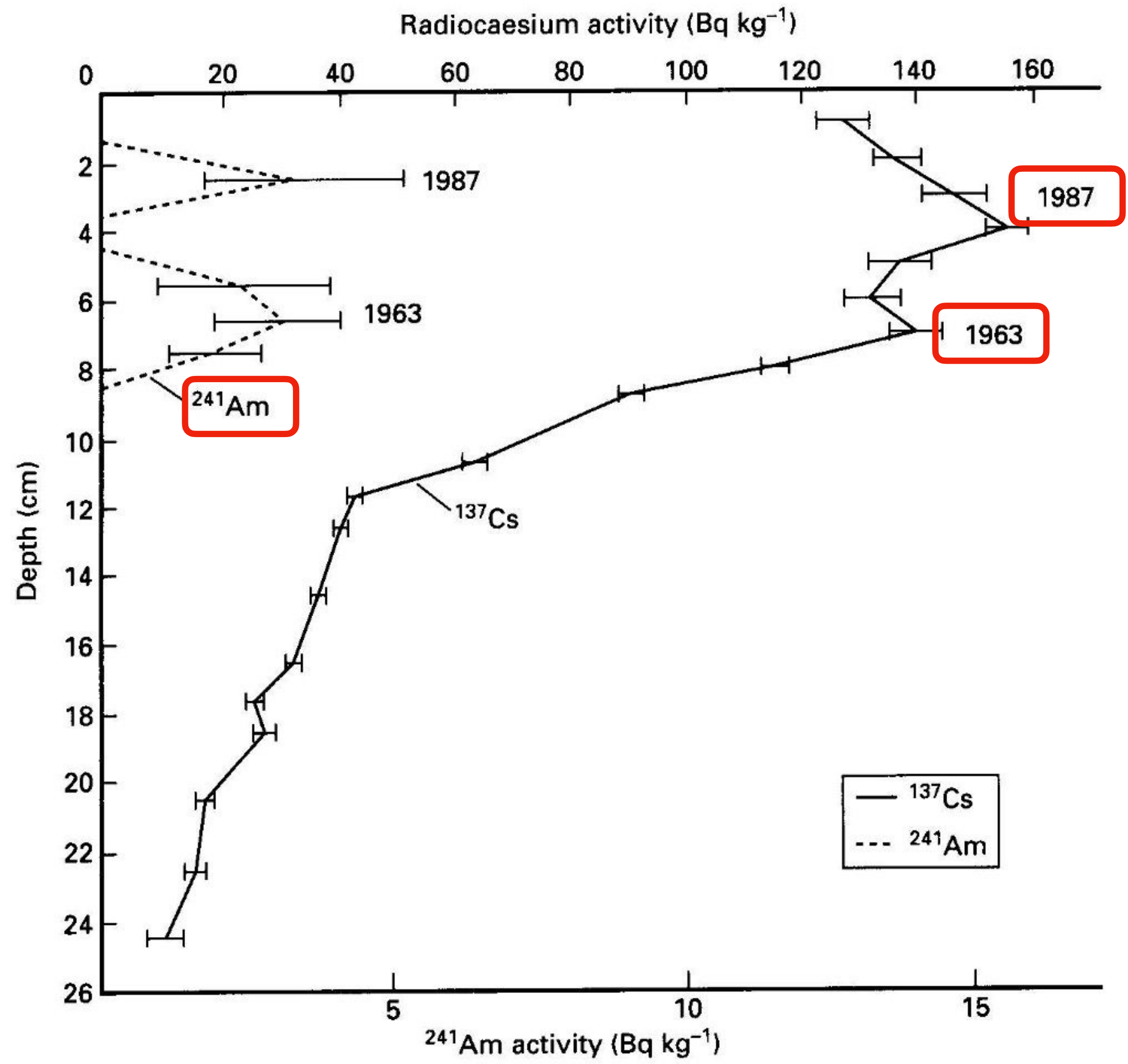
- Result of fission in nuclear reactors
- Marks **nuklear testing / disasters**
- **Complements ^{210}Pb dating**
- Unstable in sediments → migration



Appleby (2001)

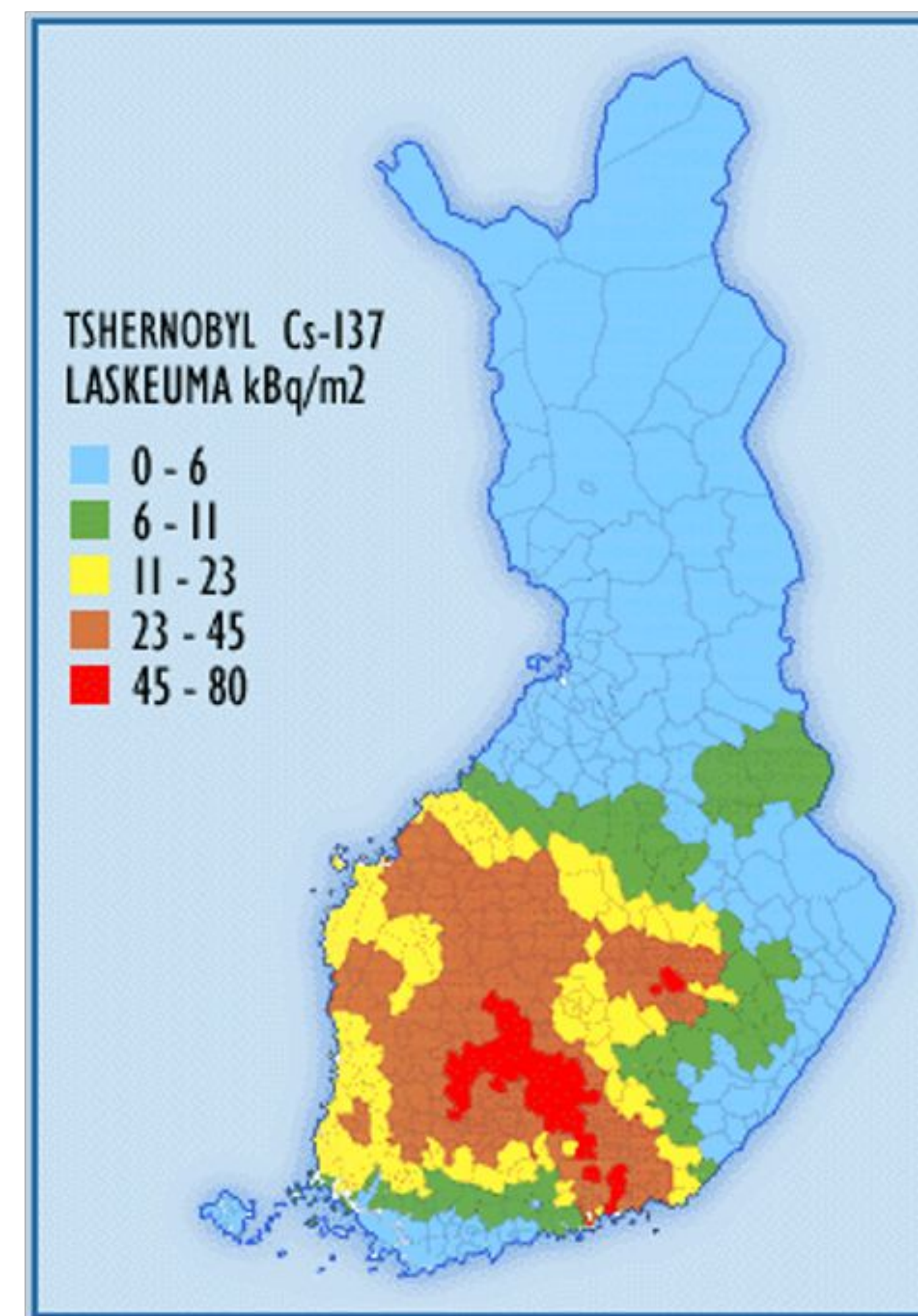
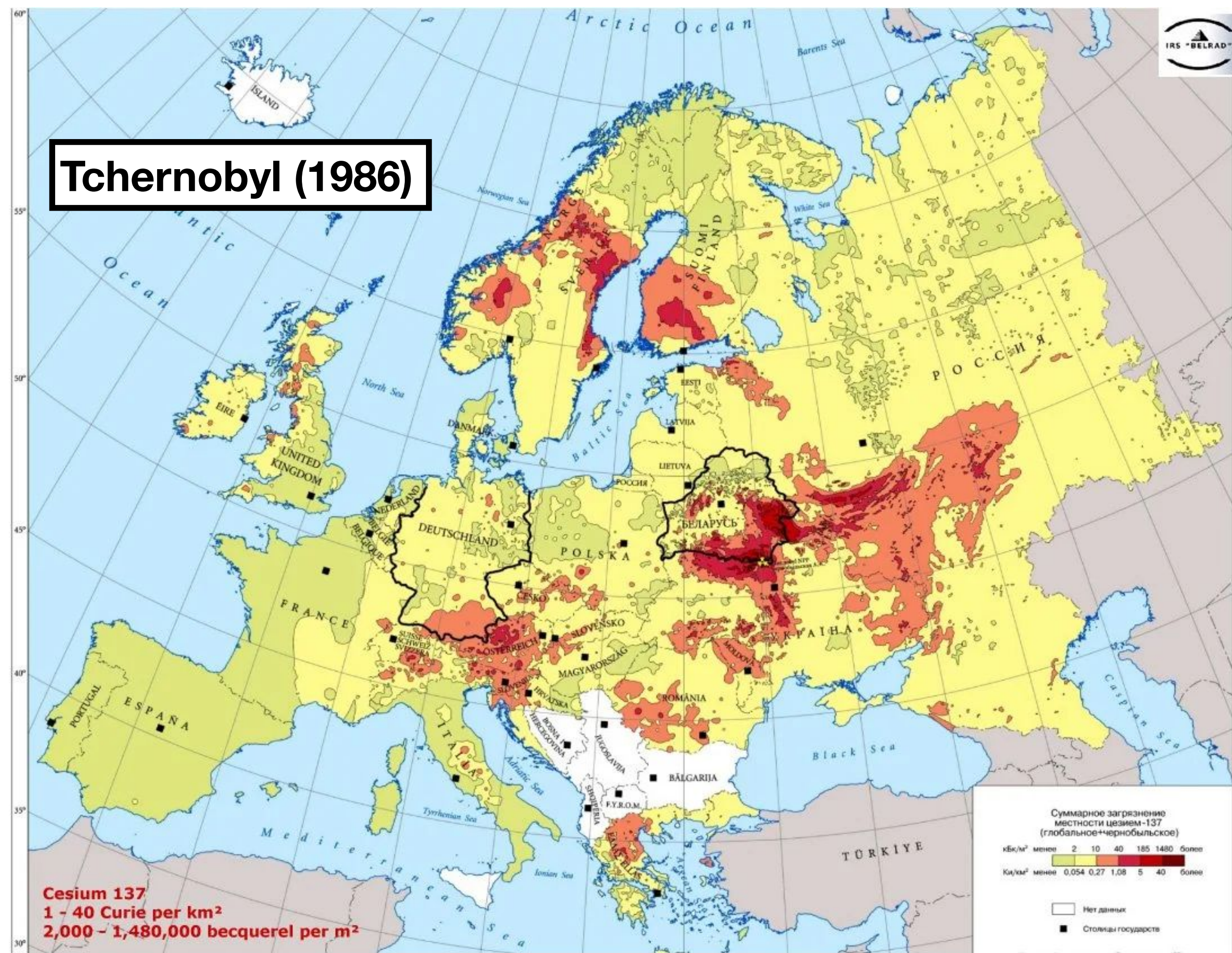


Cesium (^{137}Cs)



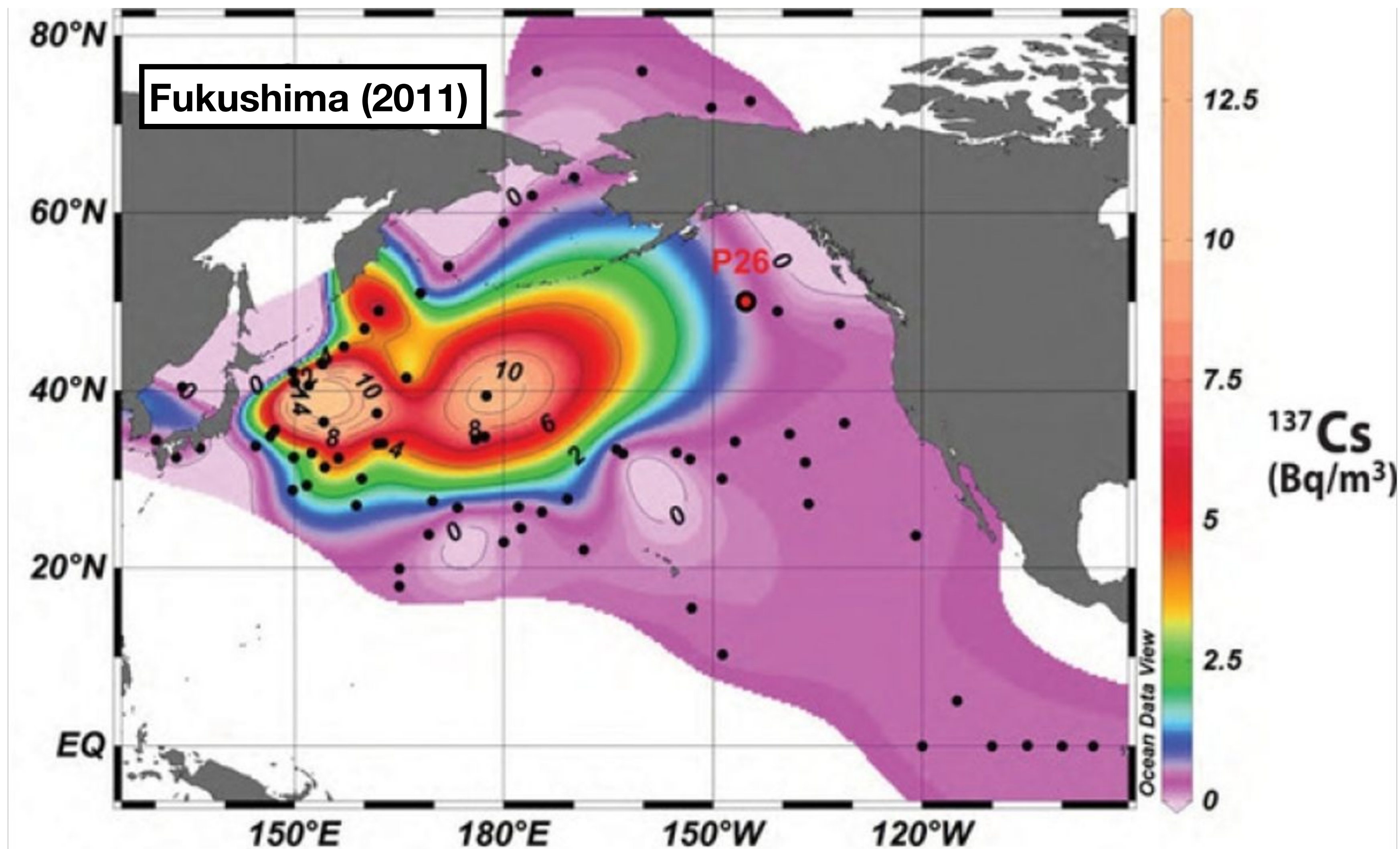


Cesium (^{137}Cs)

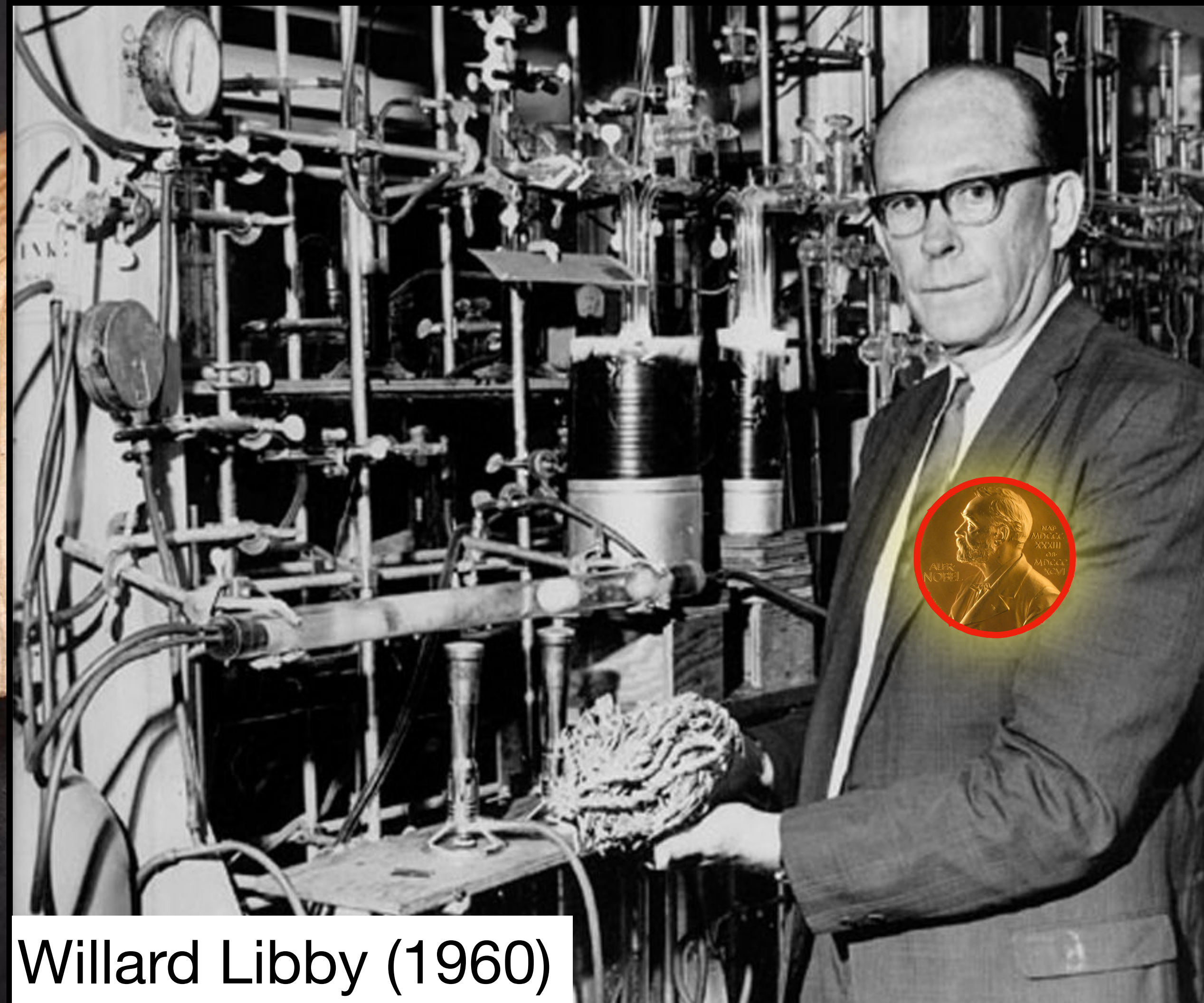




Cesium (^{137}Cs)



Radiocarbon (^{14}C)

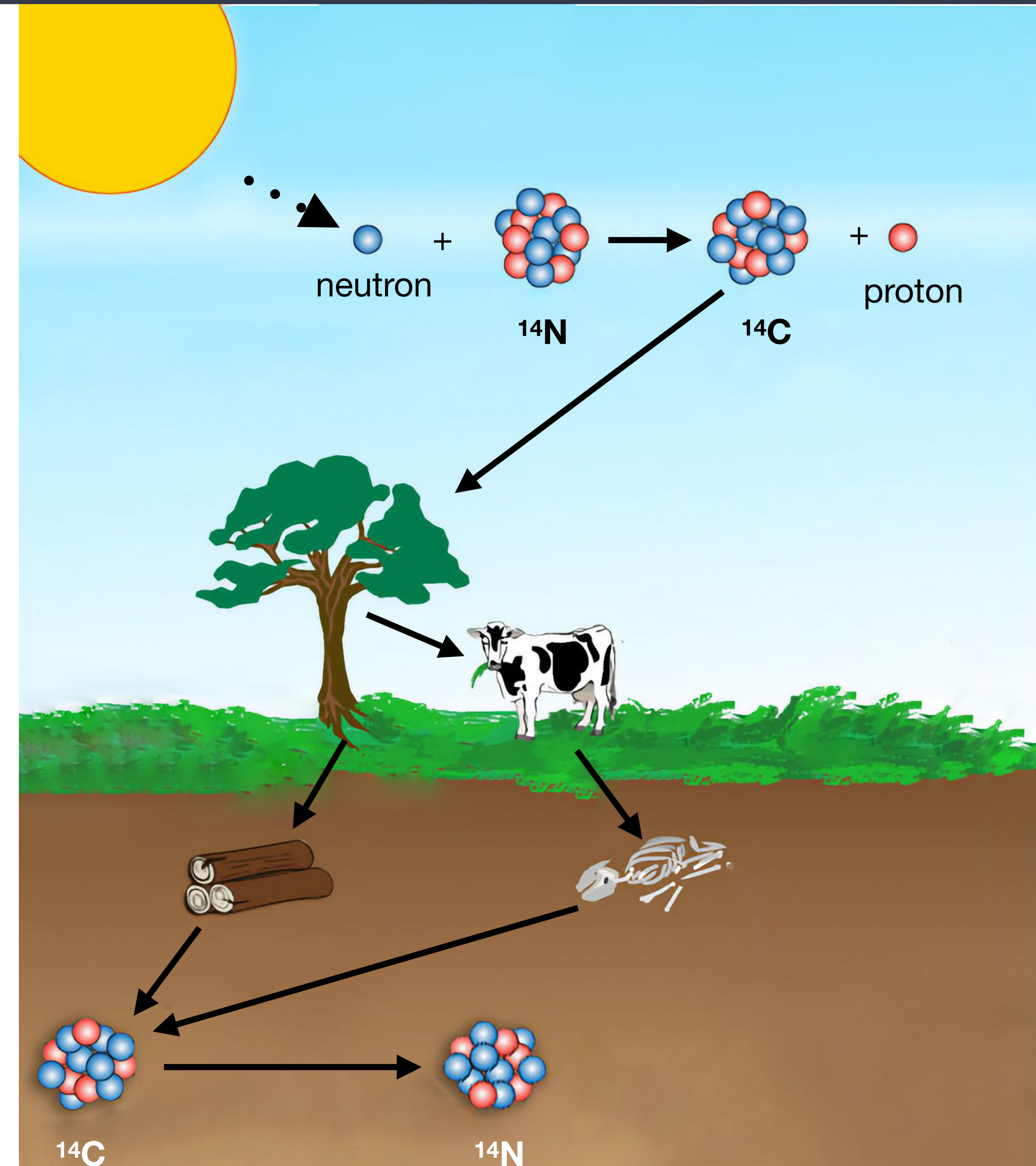


Willard Libby (1960)

Acer rubrum seed macrofossil

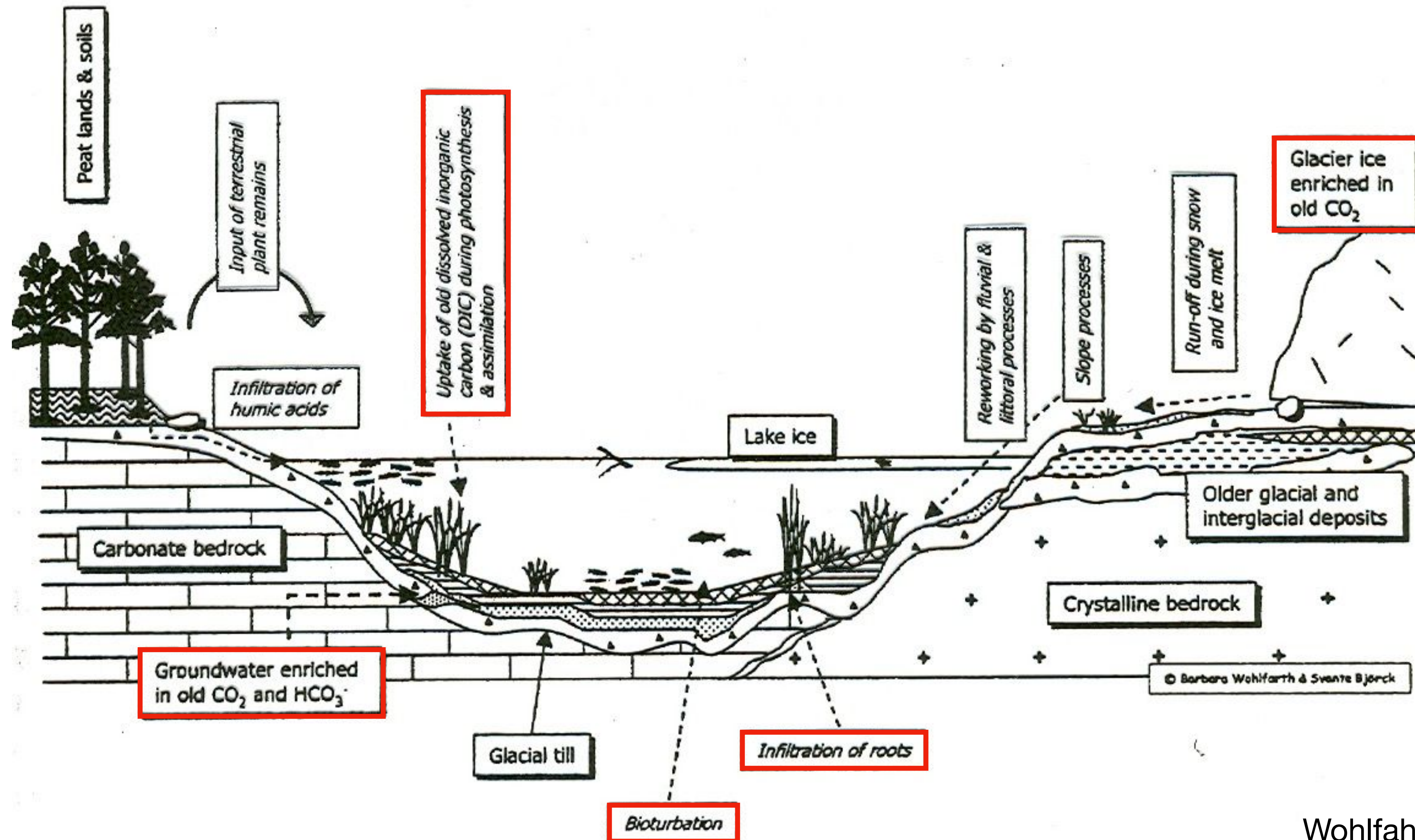
Radiocarbon (^{14}C)

- Cosmic rays: $^{14}\text{N} \rightarrow ^{14}\text{C}$
- ^{14}C in **equilibrium** in living organisms
- **After death:** no new overturn, only **decay** $^{14}\text{C} \rightarrow ^{14}\text{N}$





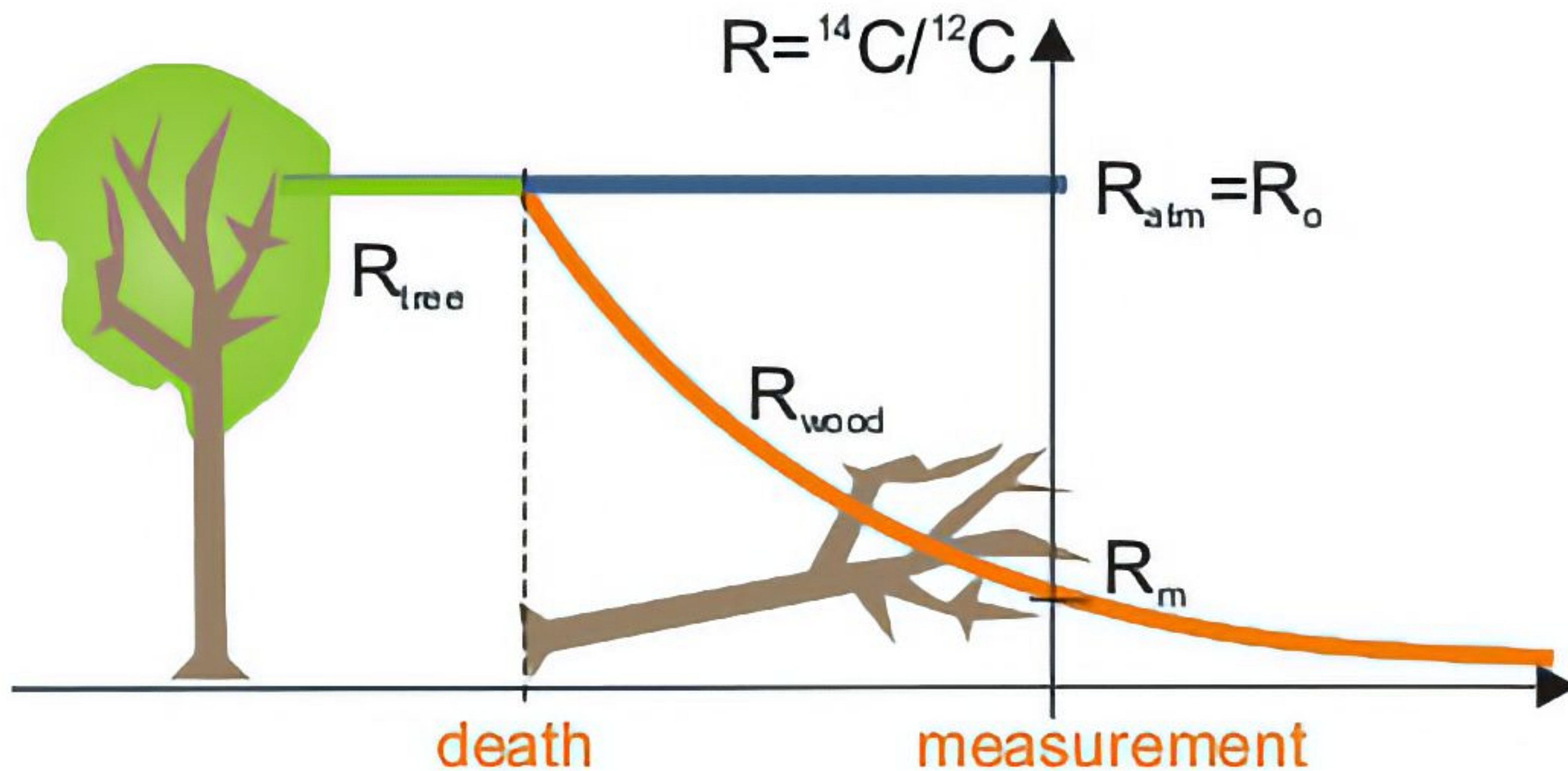
Radiocarbon (^{14}C)





Radiocarbon (^{14}C)

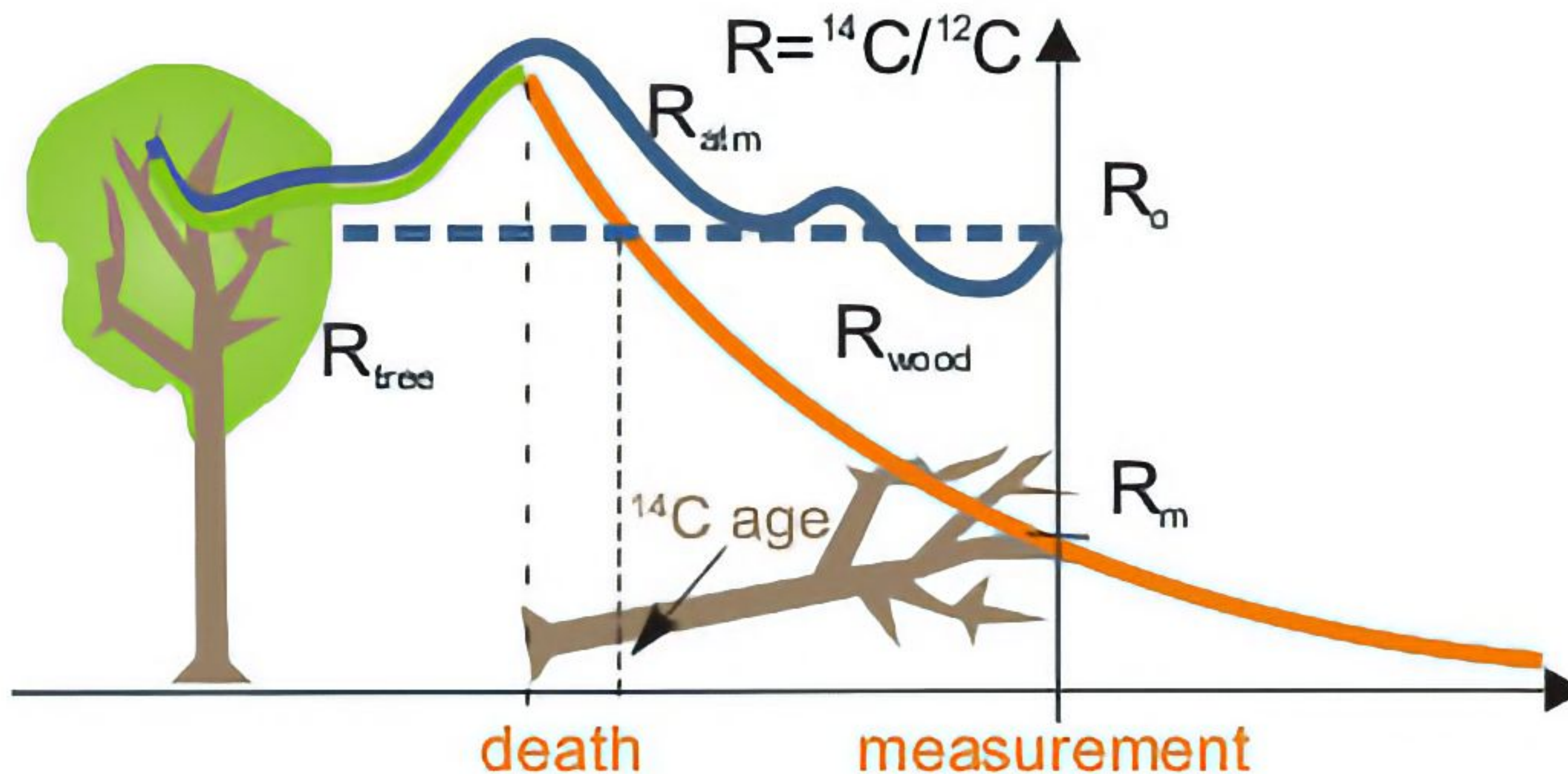
^{14}C half-life: 5730 ± 40 years

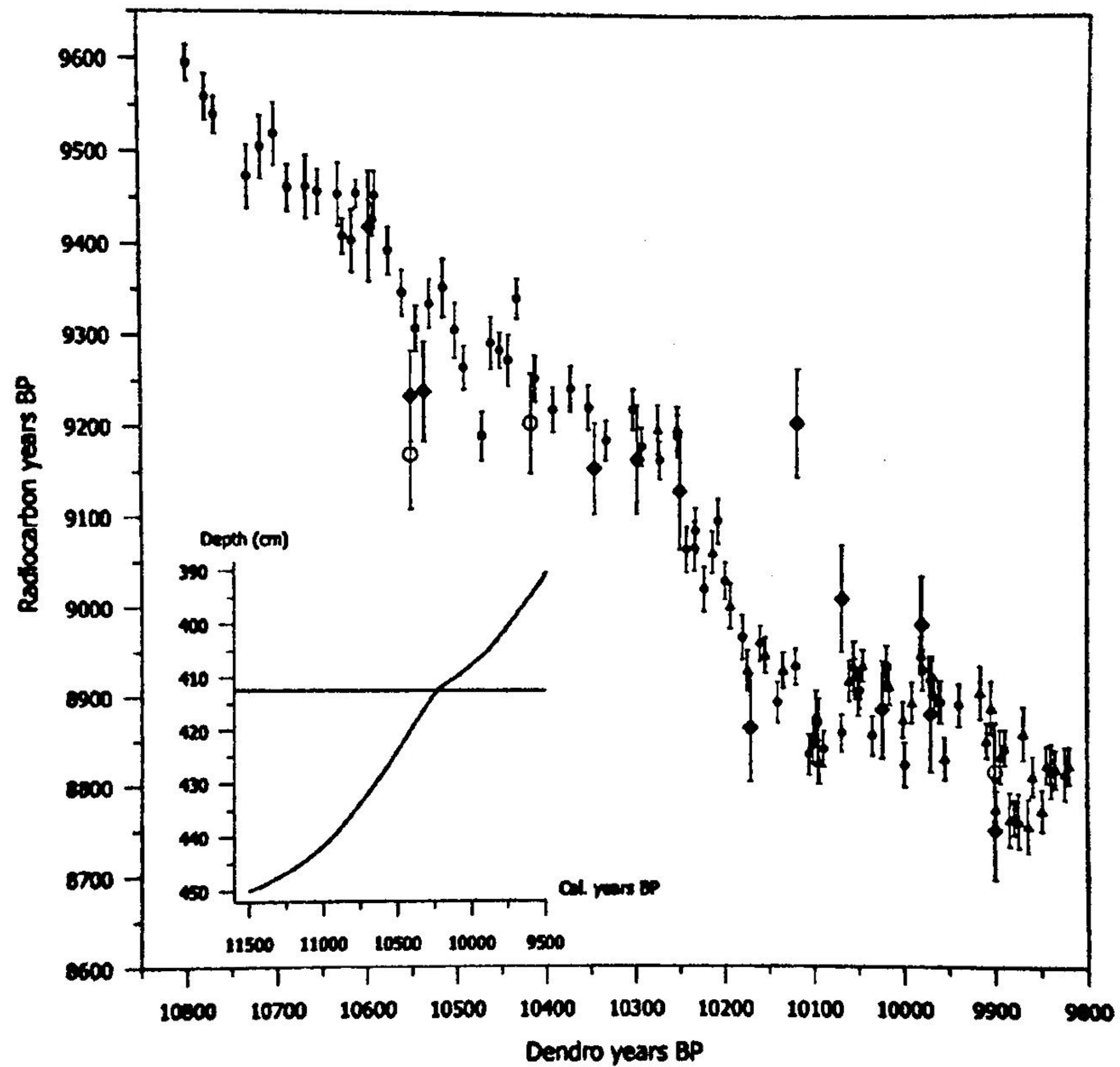




Radiocarbon (^{14}C)

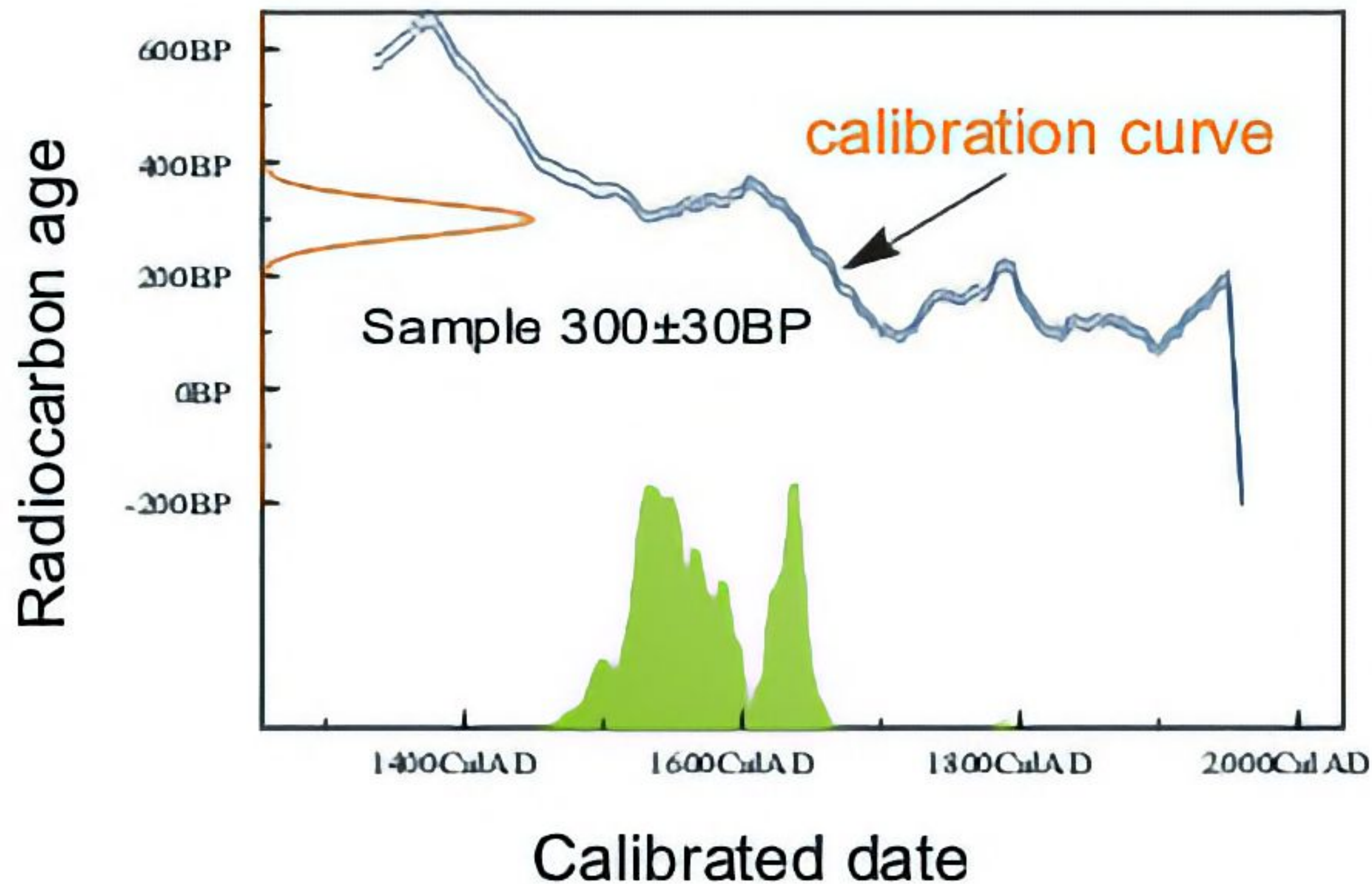
^{14}C half-life: 5730 ± 40 years







Radiocarbon (^{14}C)

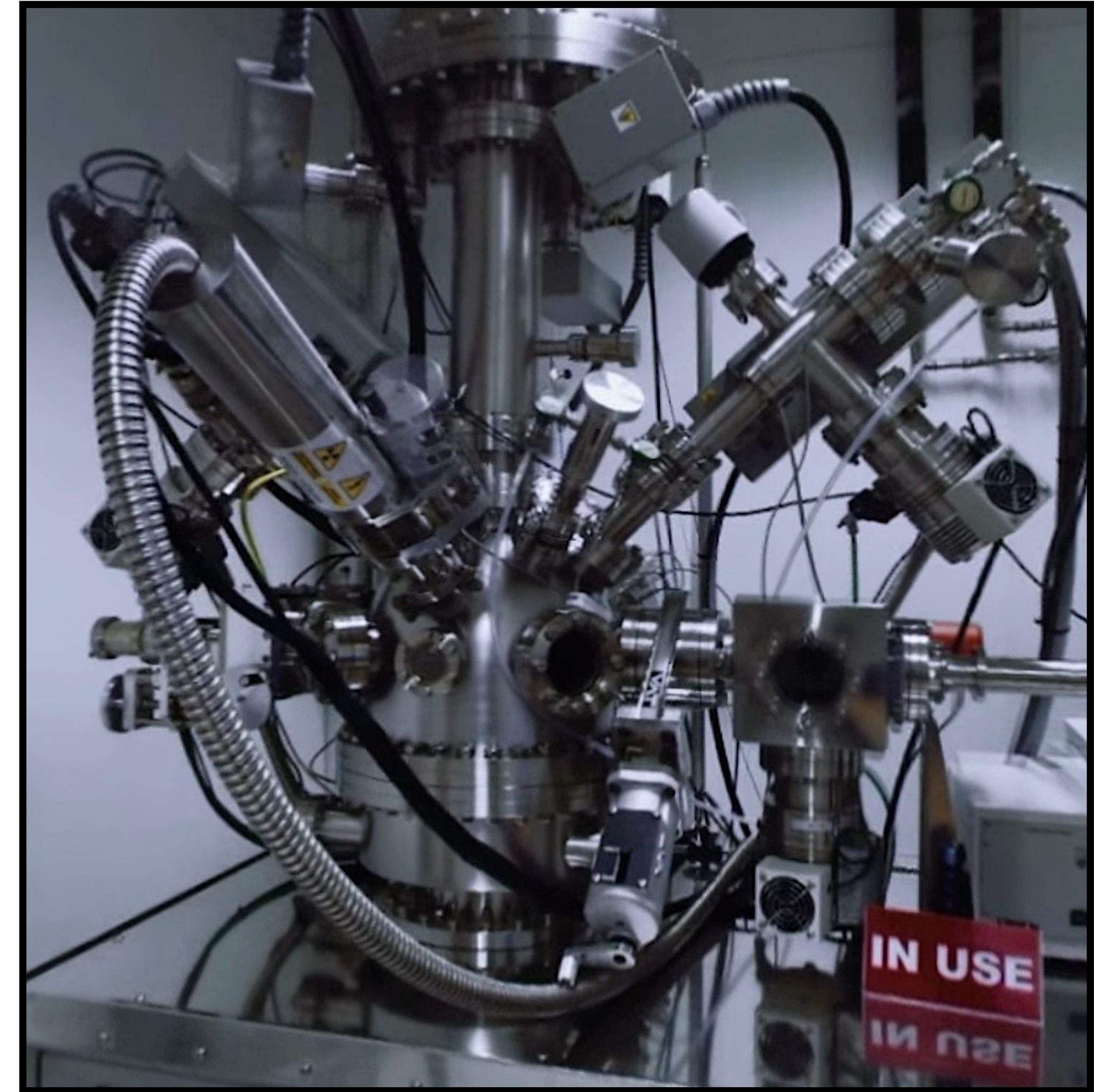




Radiocarbon (^{14}C)

Accelerator Mass Spectrometry (AMS)

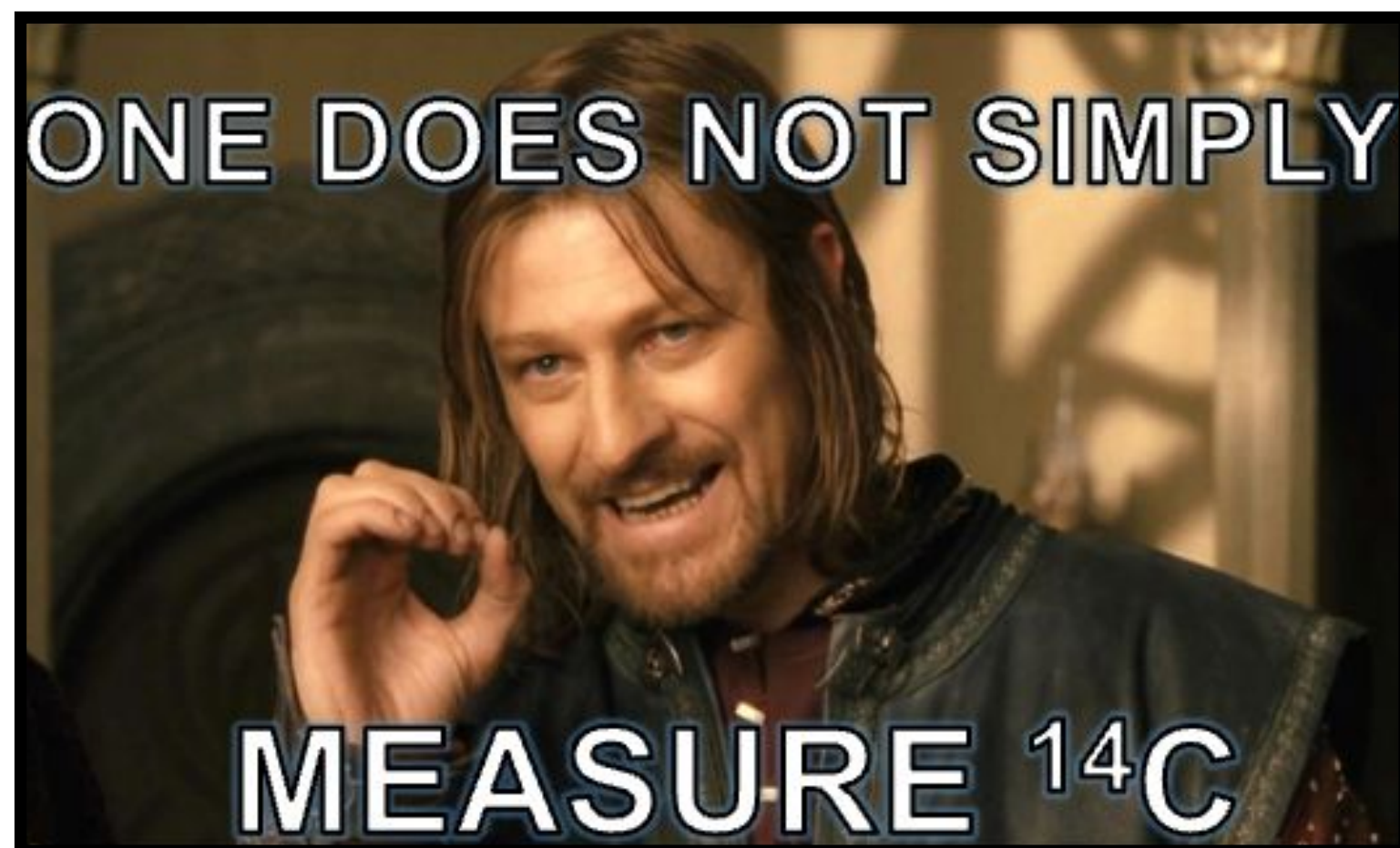
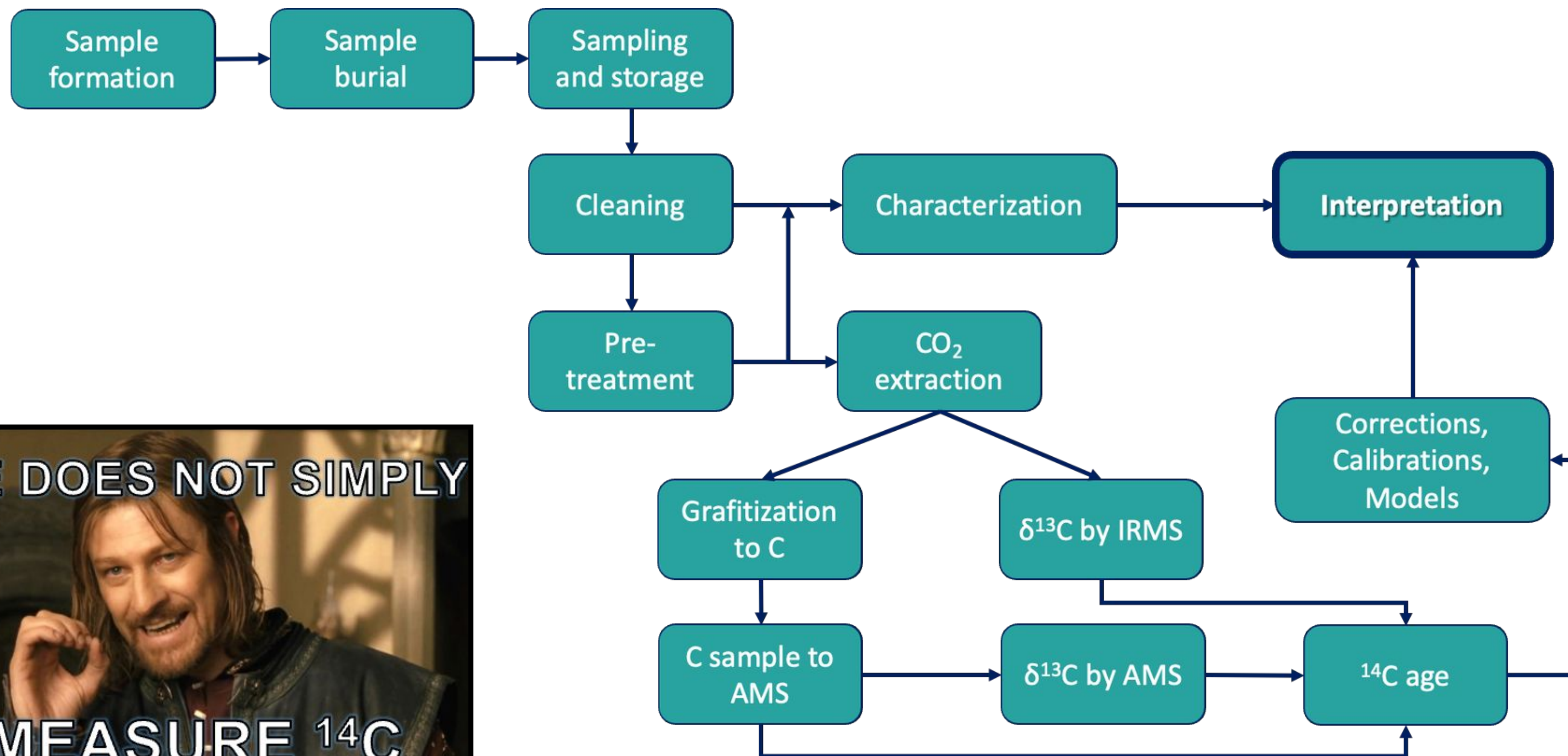
- **Small sample size** ($\leq 1 \text{ mg C}$)
→ dating material can be chosen
- **Lower errors** → older samples
(max. $\approx 60,000$ years)
- Pre-handling of the sample more careful



<https://www.youtube.com/watch?v=jDVWwZG7O1Y>



Radiocarbon (^{14}C)





Varves

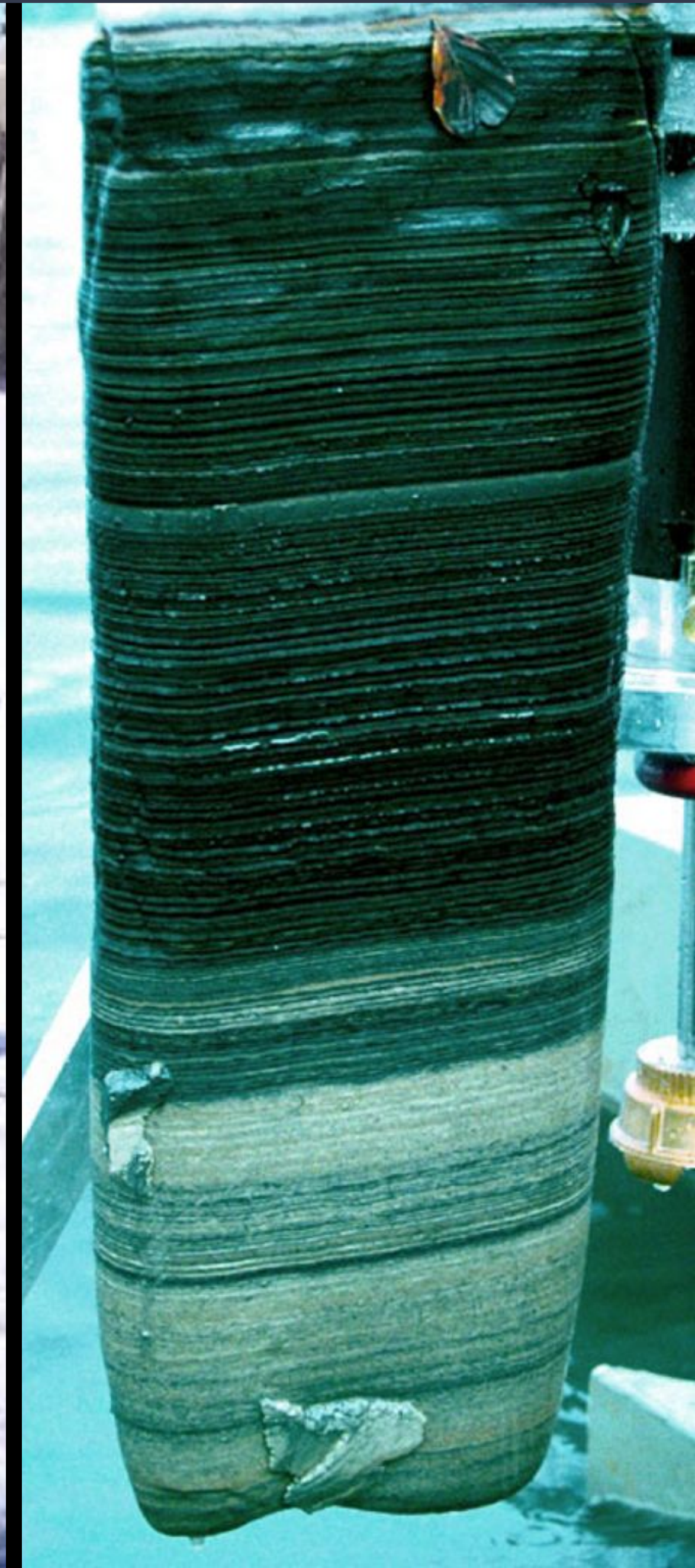


Geopark Indalsälven, Sweden



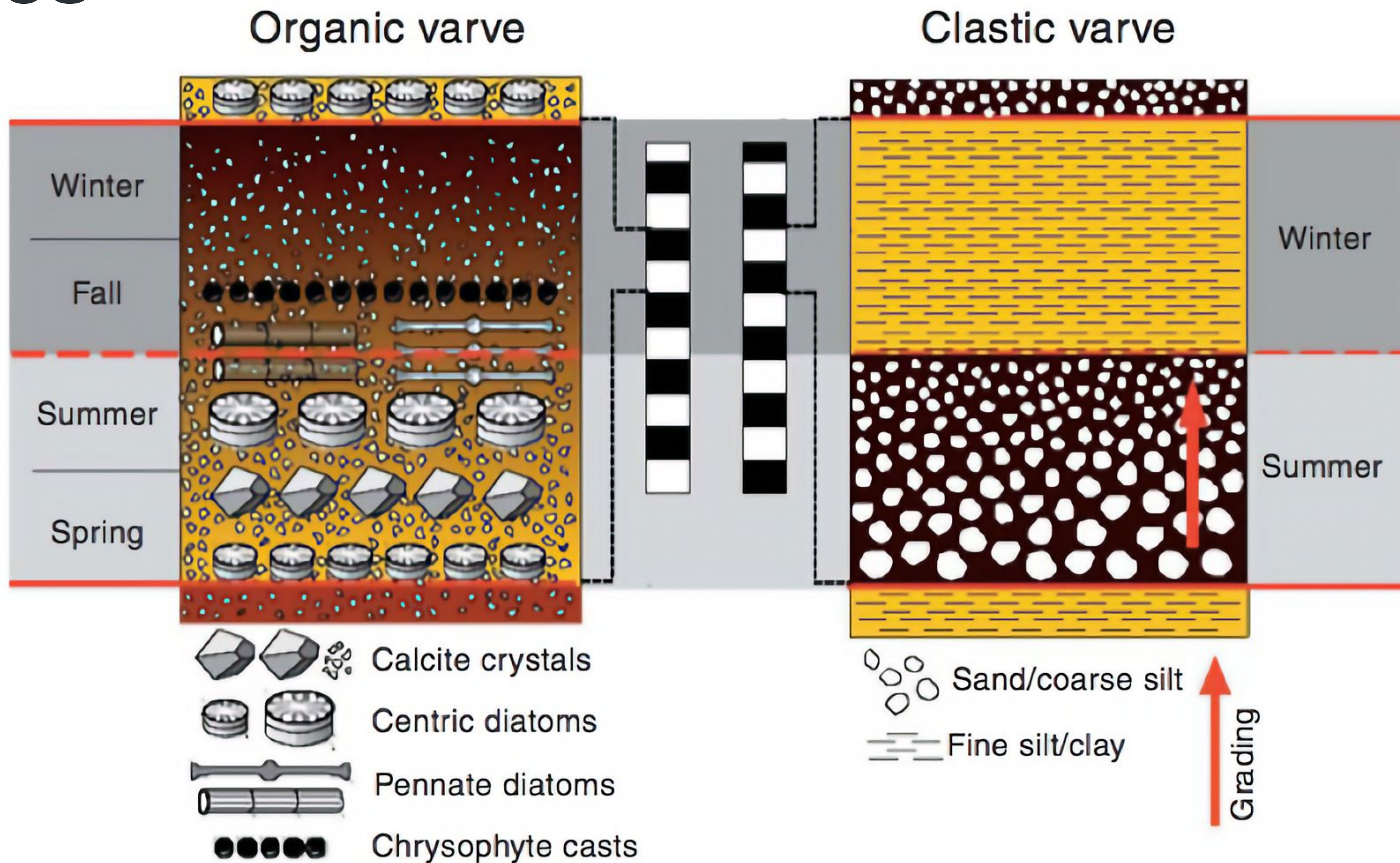
Varves

- Undisturbed conditions
- **Annual / seasonal** laminations
 - **Clastic**
 - **Biogenic**
- Laminae $\neq$ Varves!



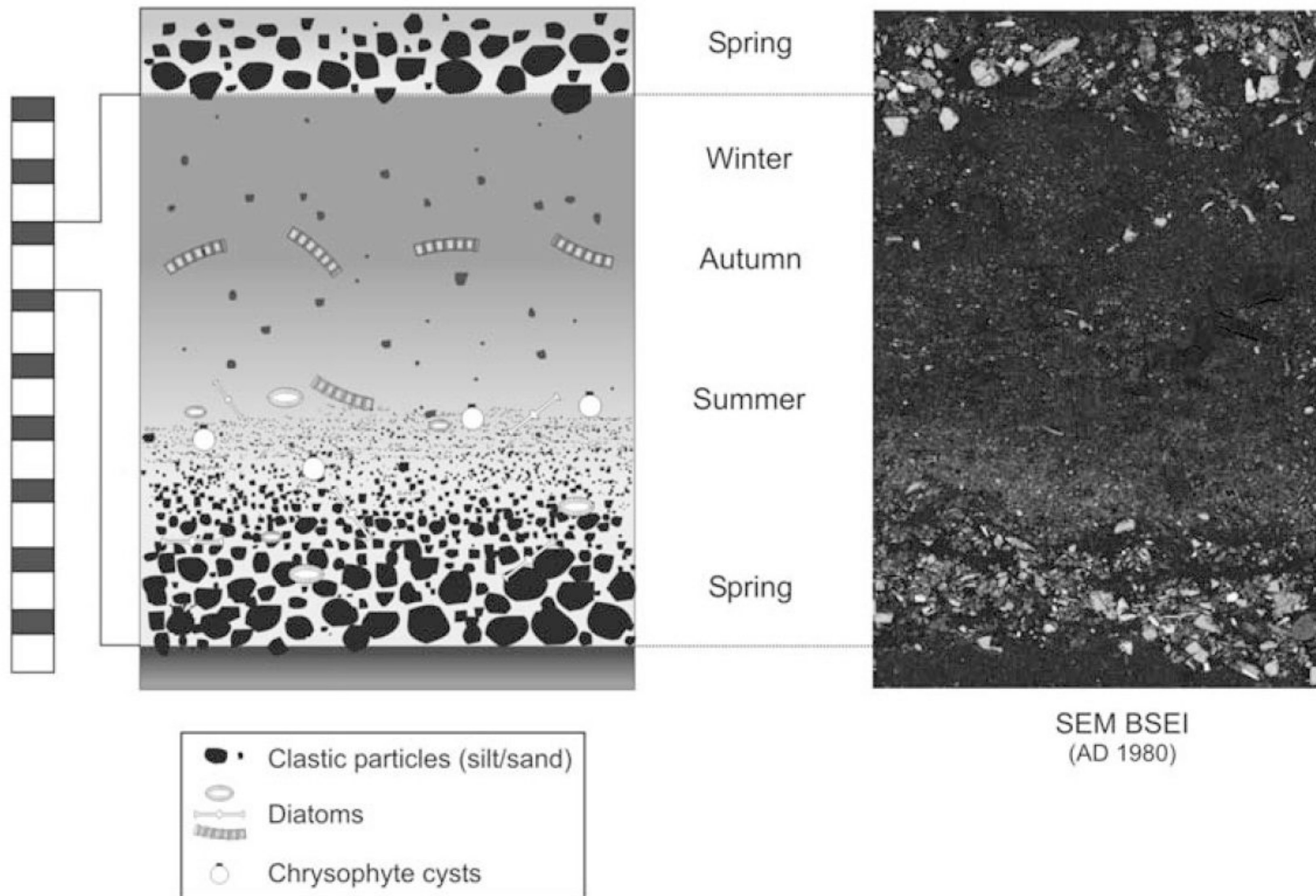


Varves

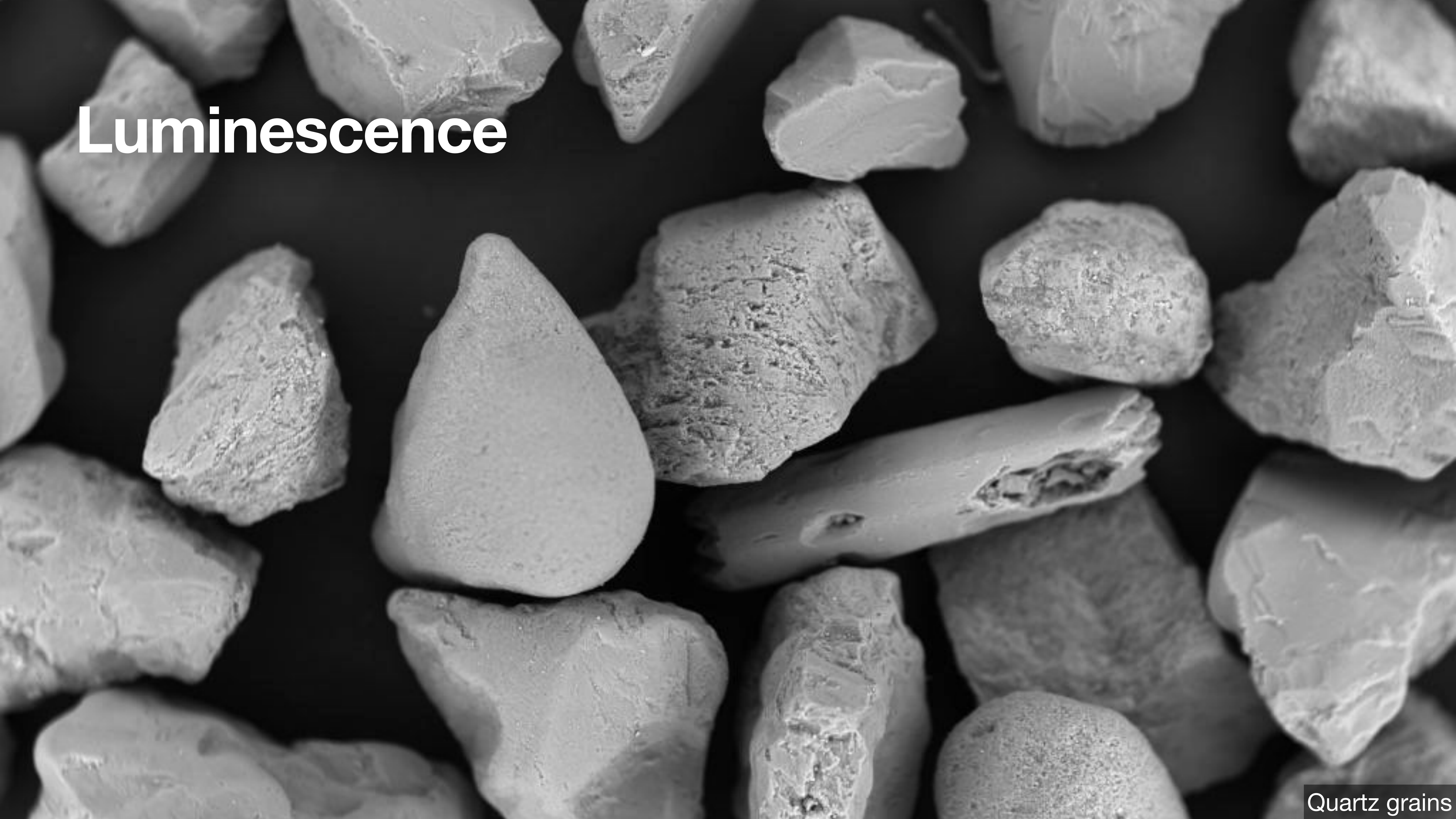




Varves



Luminescence

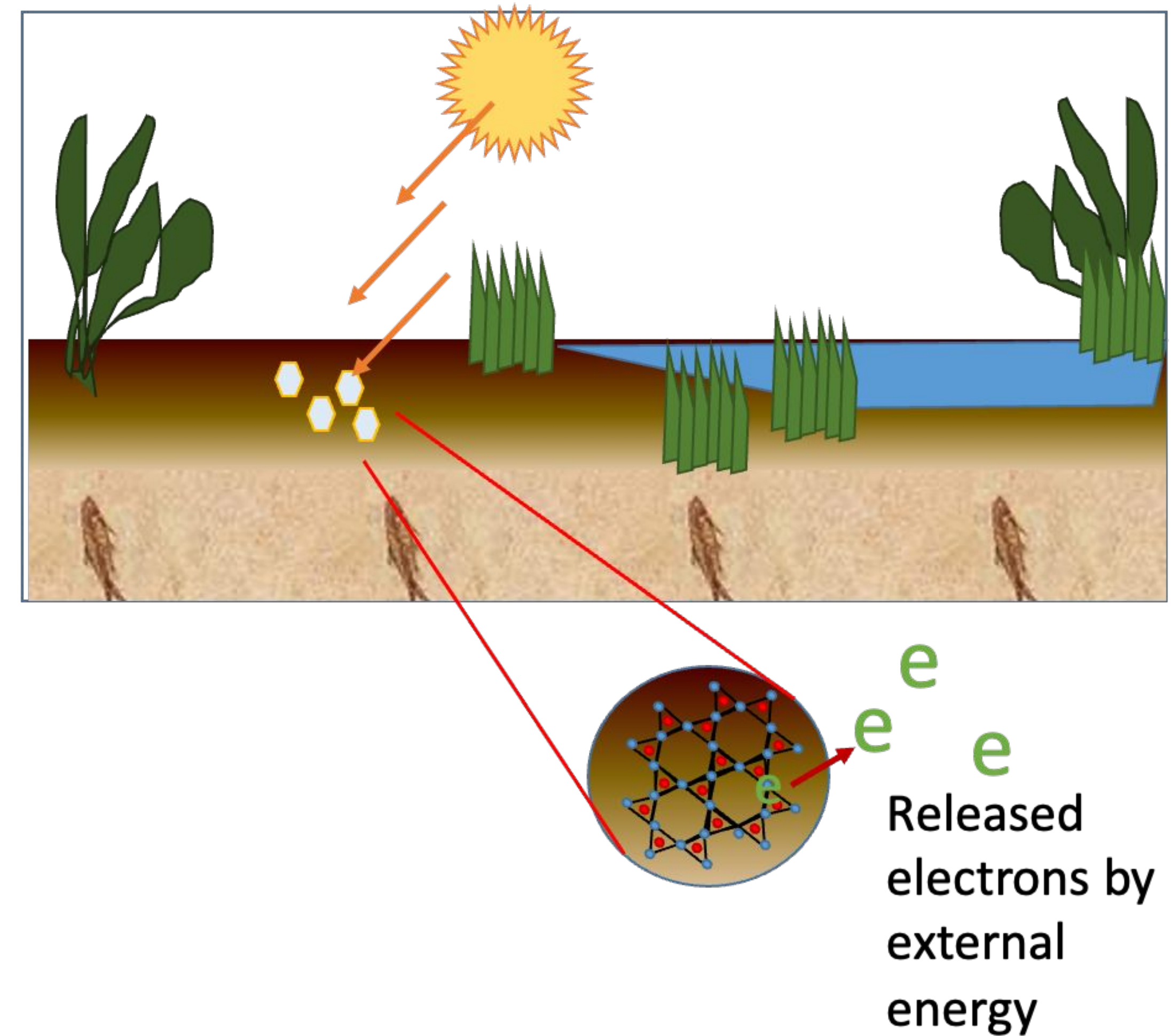


Quartz grains

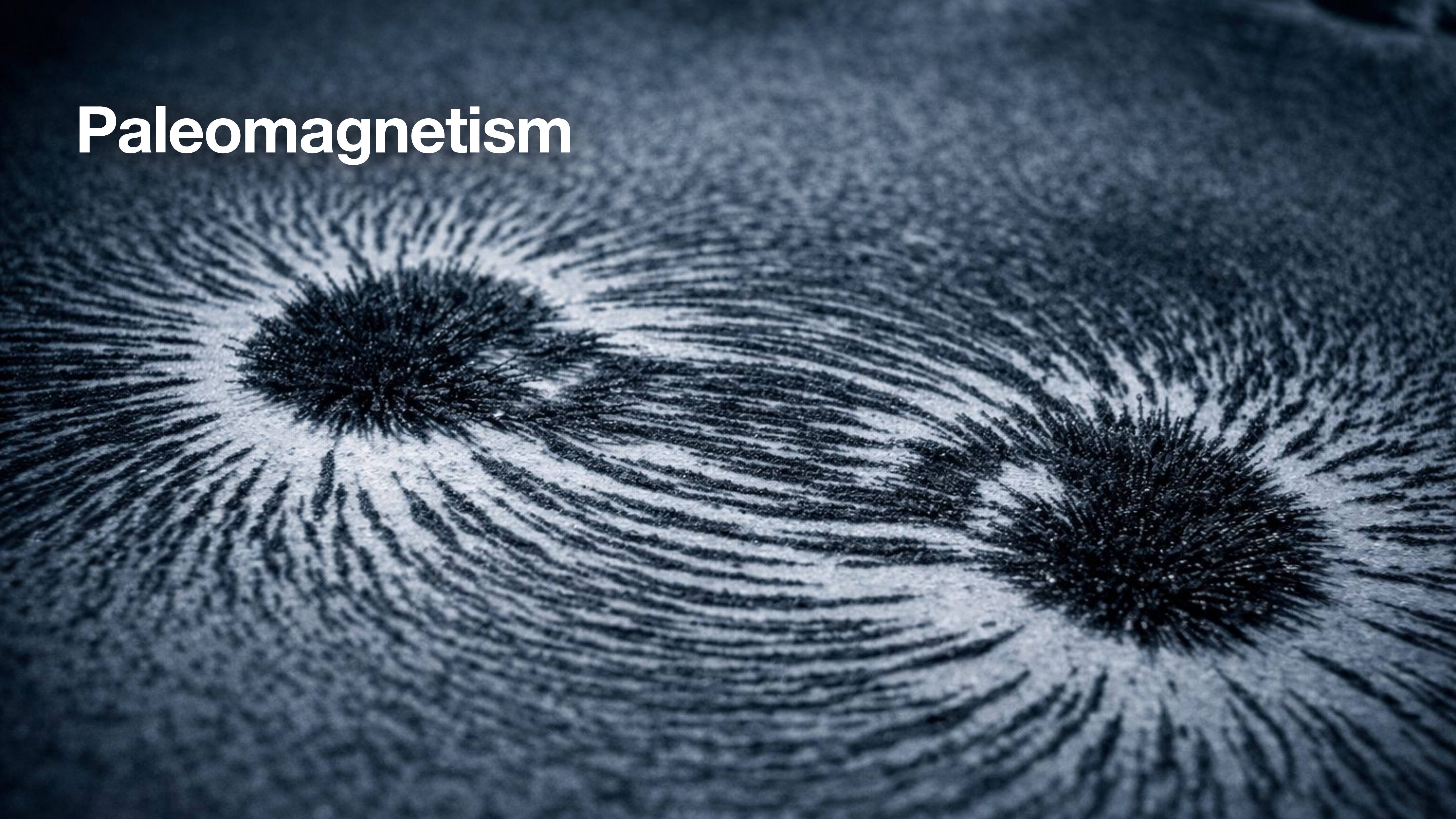


Luminescence

- Mineral **grains** store energy from **natural radiation**
- **Sunlight / heat** reset storage
- After burial, storage fills up over time
- Luminescence / **energy stored** indicates when grains were **last exposed to light**



Paleomagnetism





CANADA

Magnetic north pole,
1900

70° N

80° N

GREENLAND

2020
2015

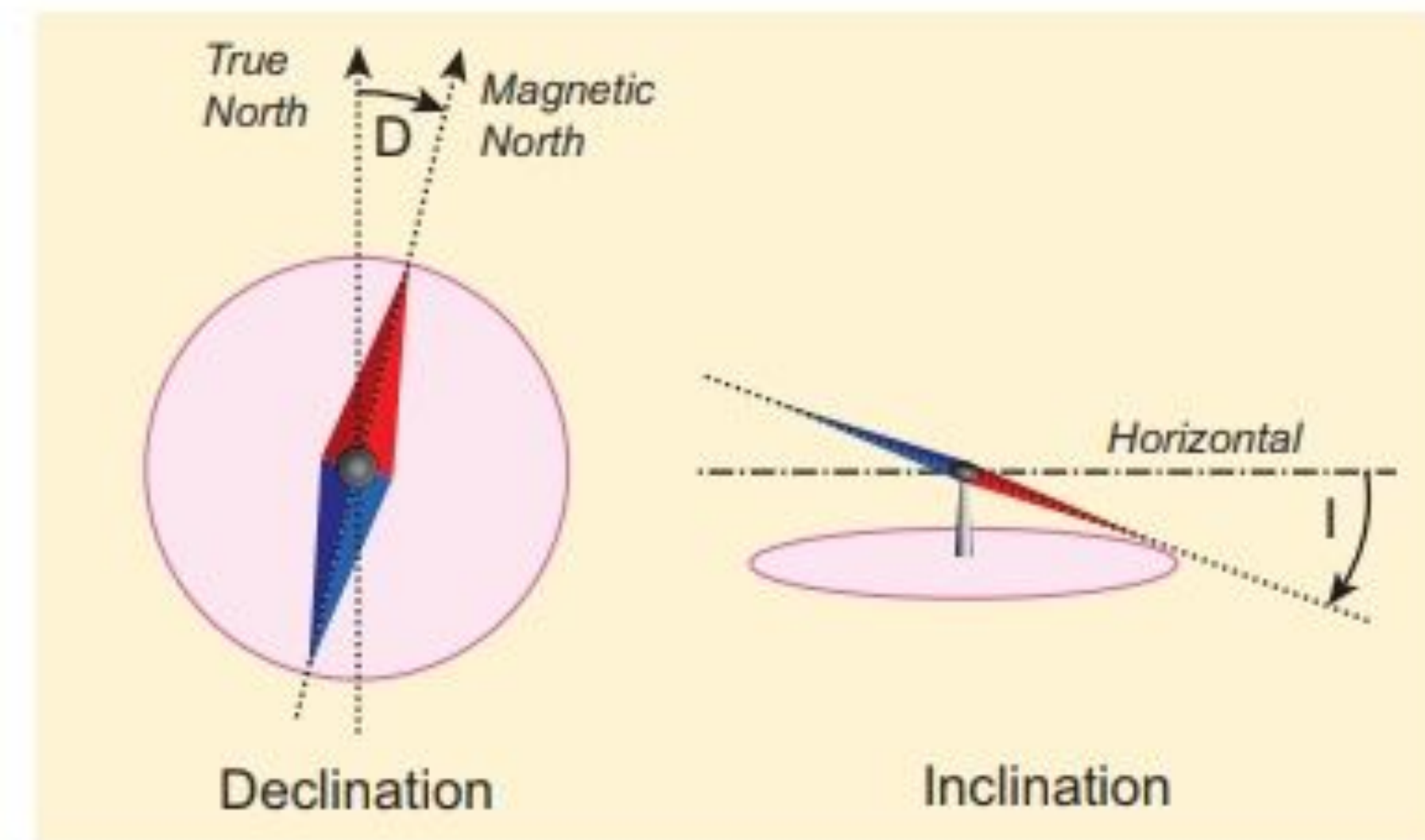
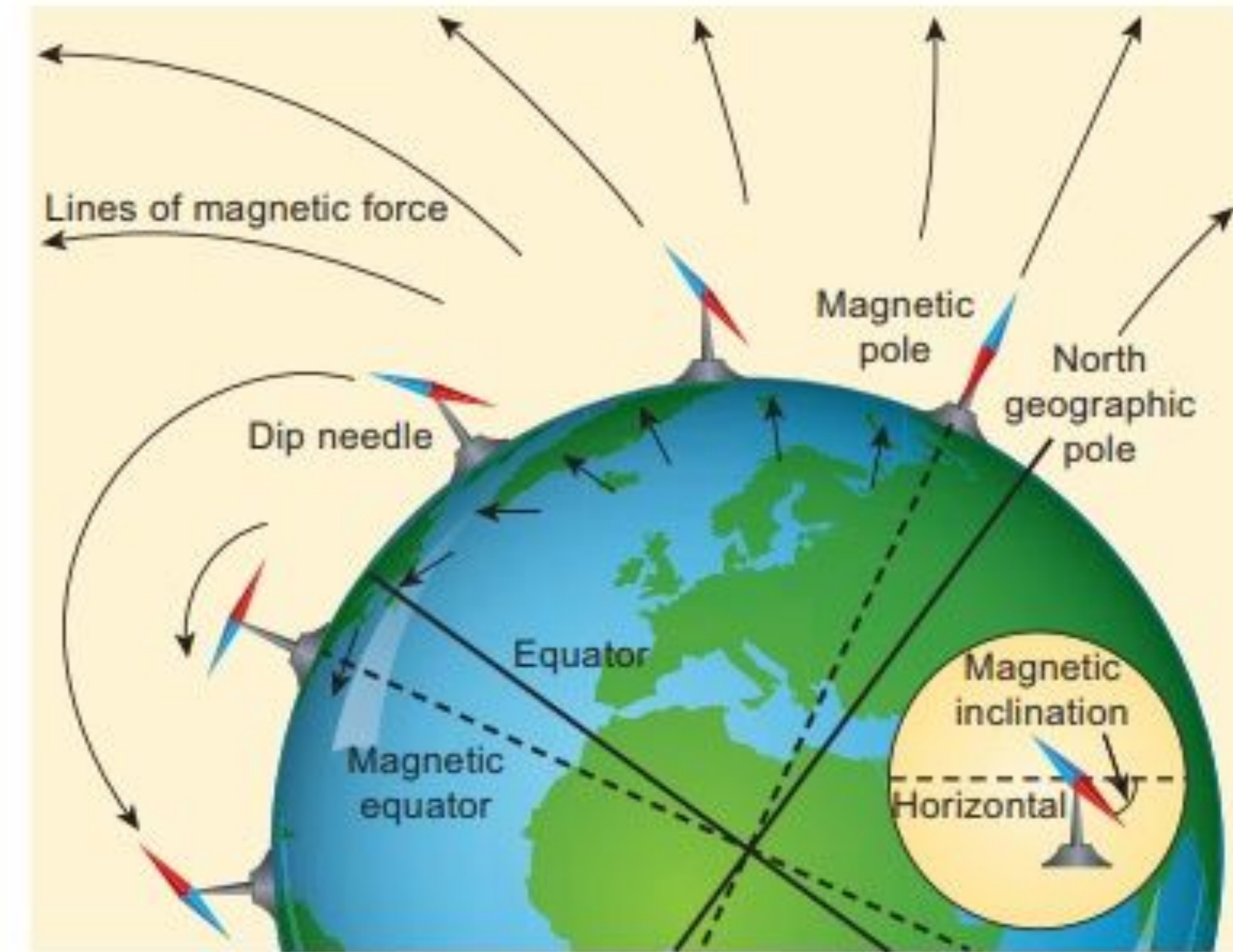
Geographic
North Pole

*Arctic
Ocean*



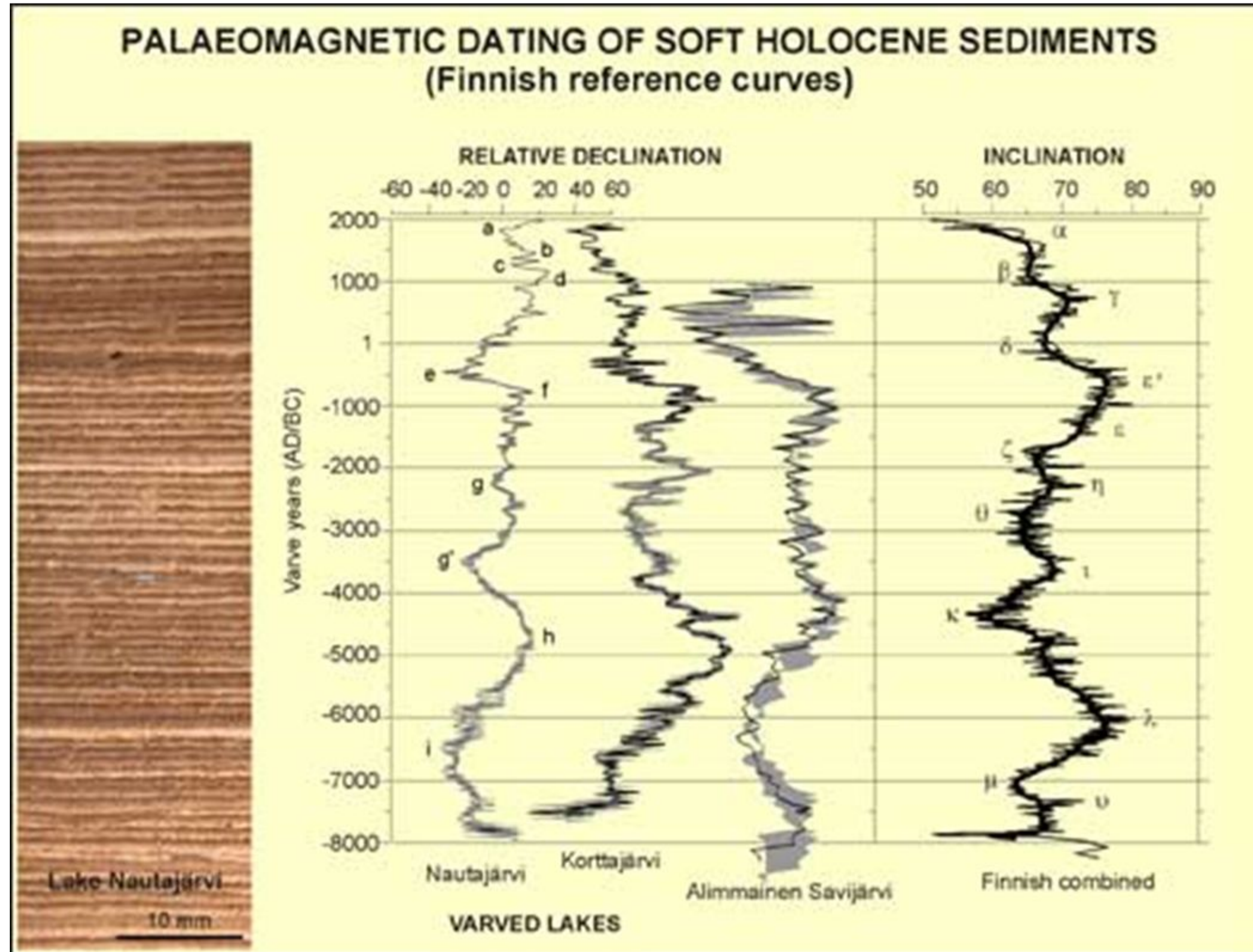
Paleomagnetism

- The Earth's **magnetic field** changes over time
- Magnetic **minerals align with Earth's field**
- Magnetization locks in after deposition (declination + inclination)





Paleomagnetism



Tephra



Grimsvötn, Iceland



Tephras

- Pyroclastic **material from volcanic eruptions**
- Ash forms “time stamp” layer
- **Ash** identified by chemical fingerprint, **matched to known eruption date**





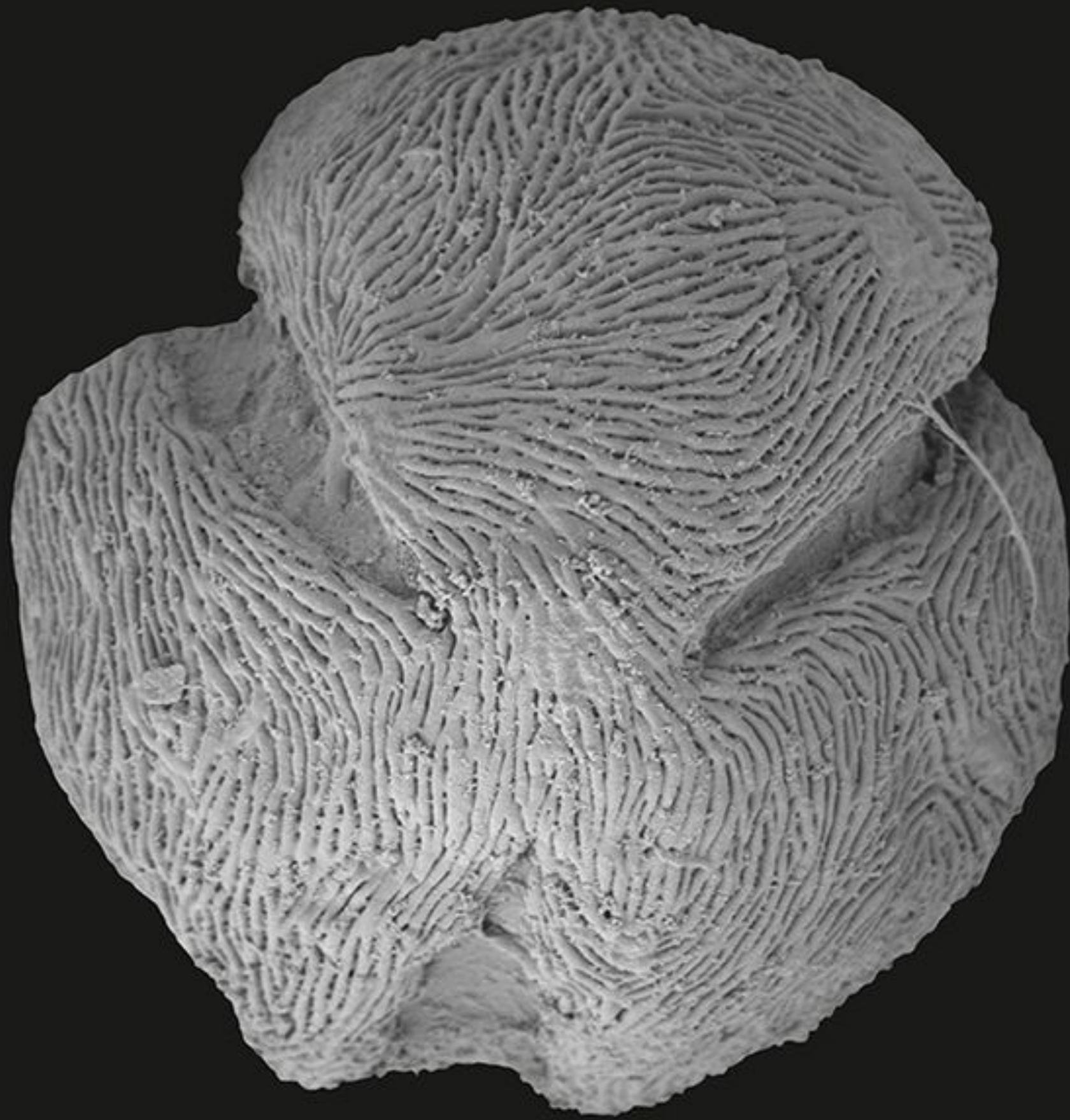
Tephra

Examples:

- Lacher-tephra Late-Glacial, 10 800 BP
- Vasset/Killian tephra in France 8230 BP
- Mazama ash 6600 BP
- Hekla, Vesuvius, Santorini eruptions



Fossils

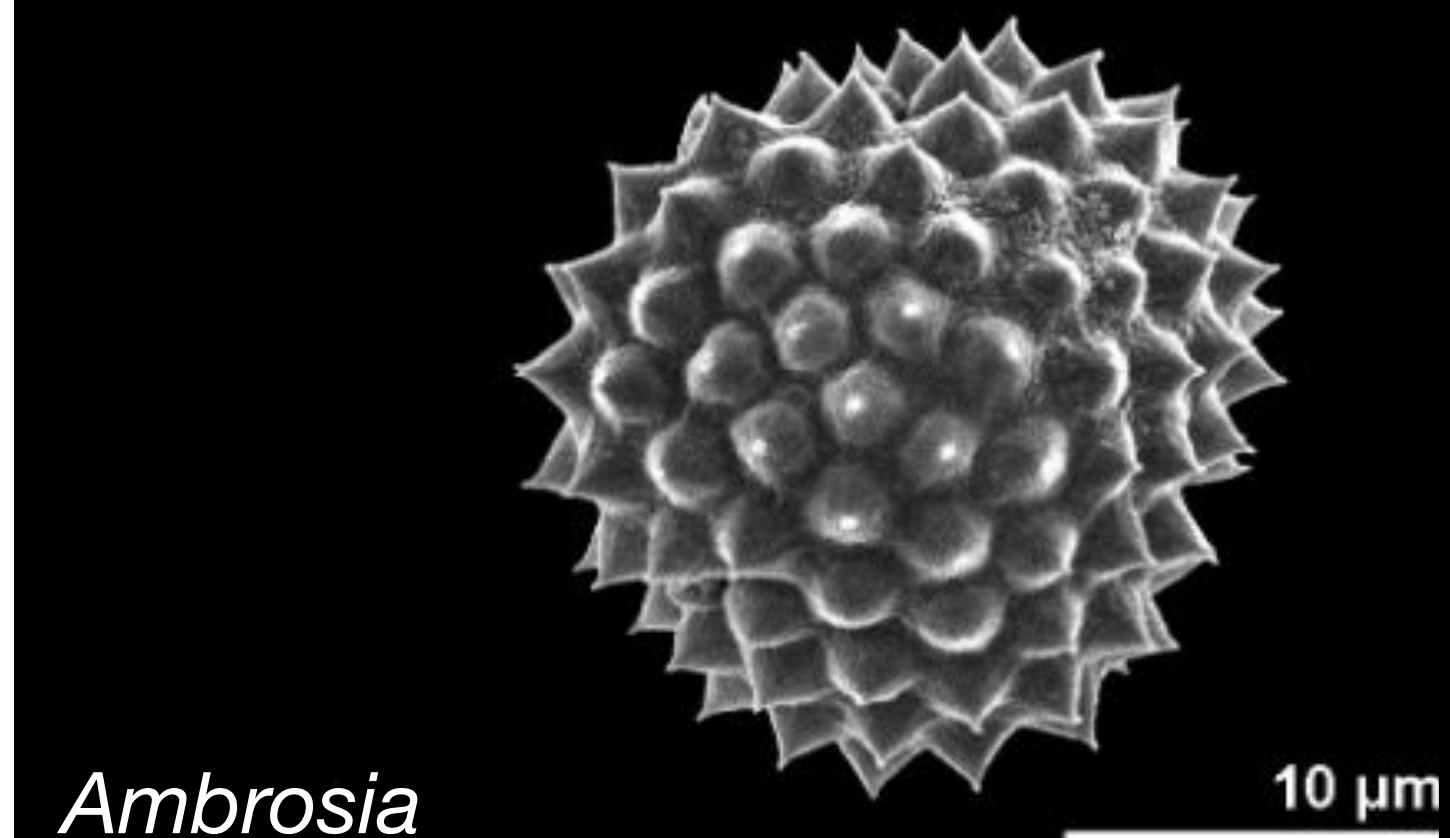
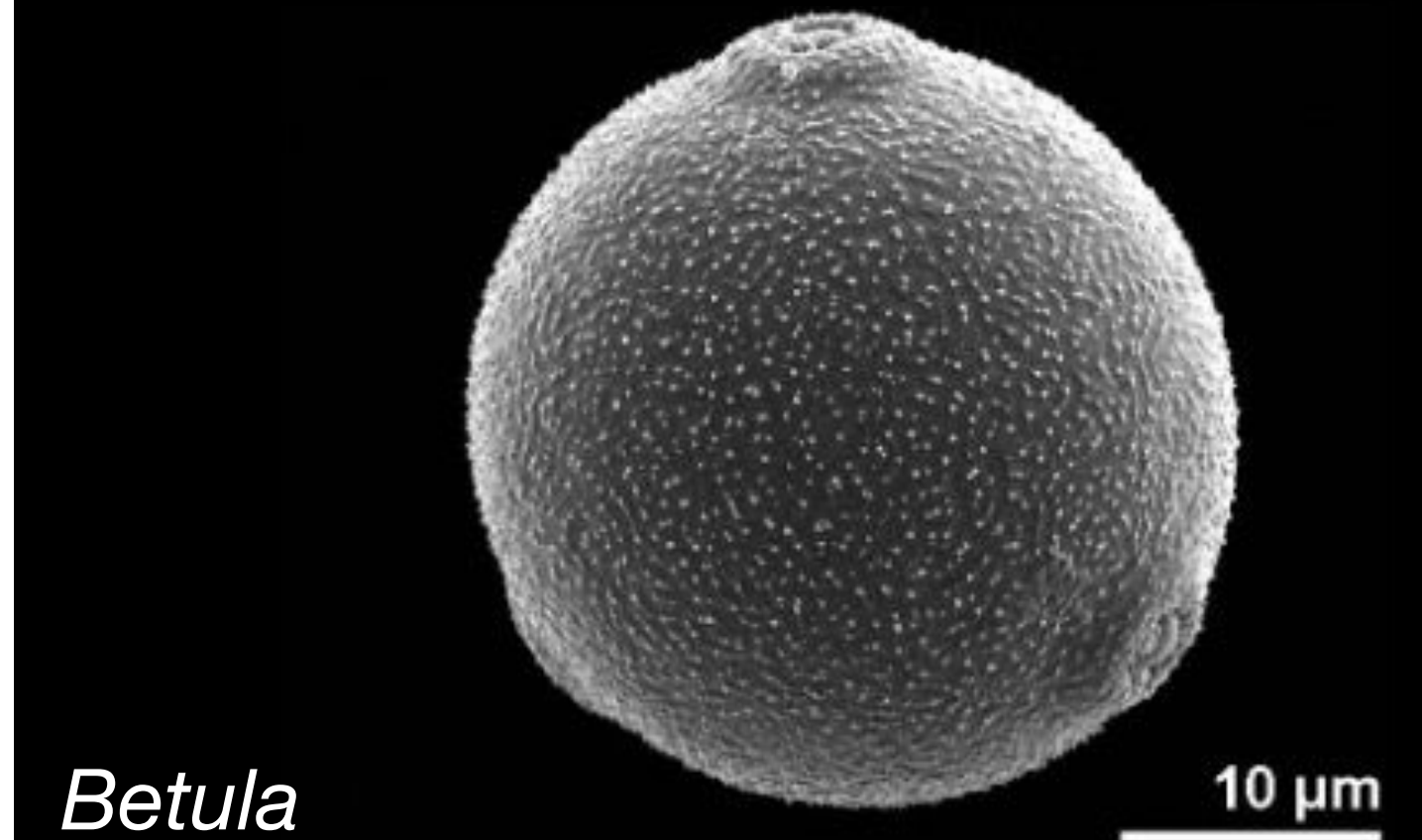
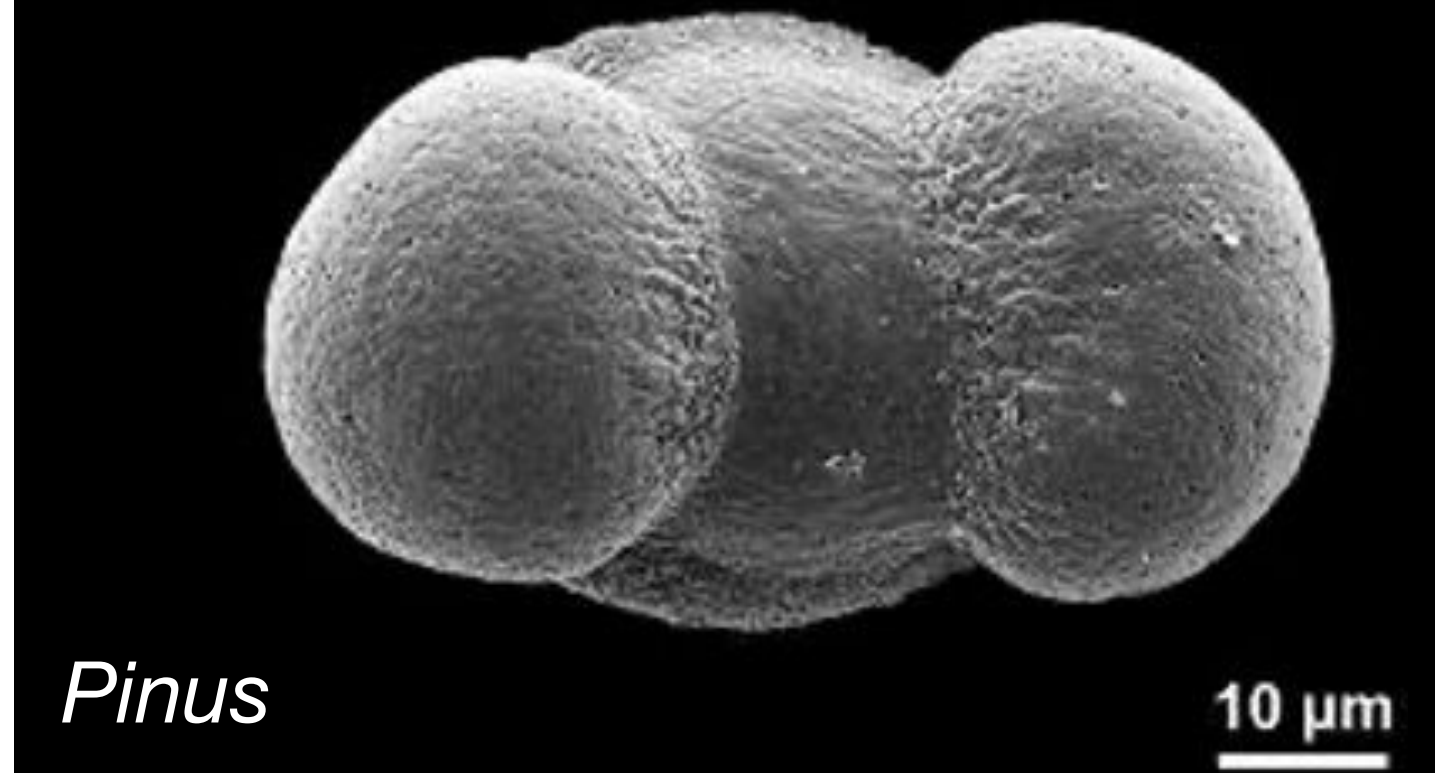


Fossil pollen from central Asia



Fossils

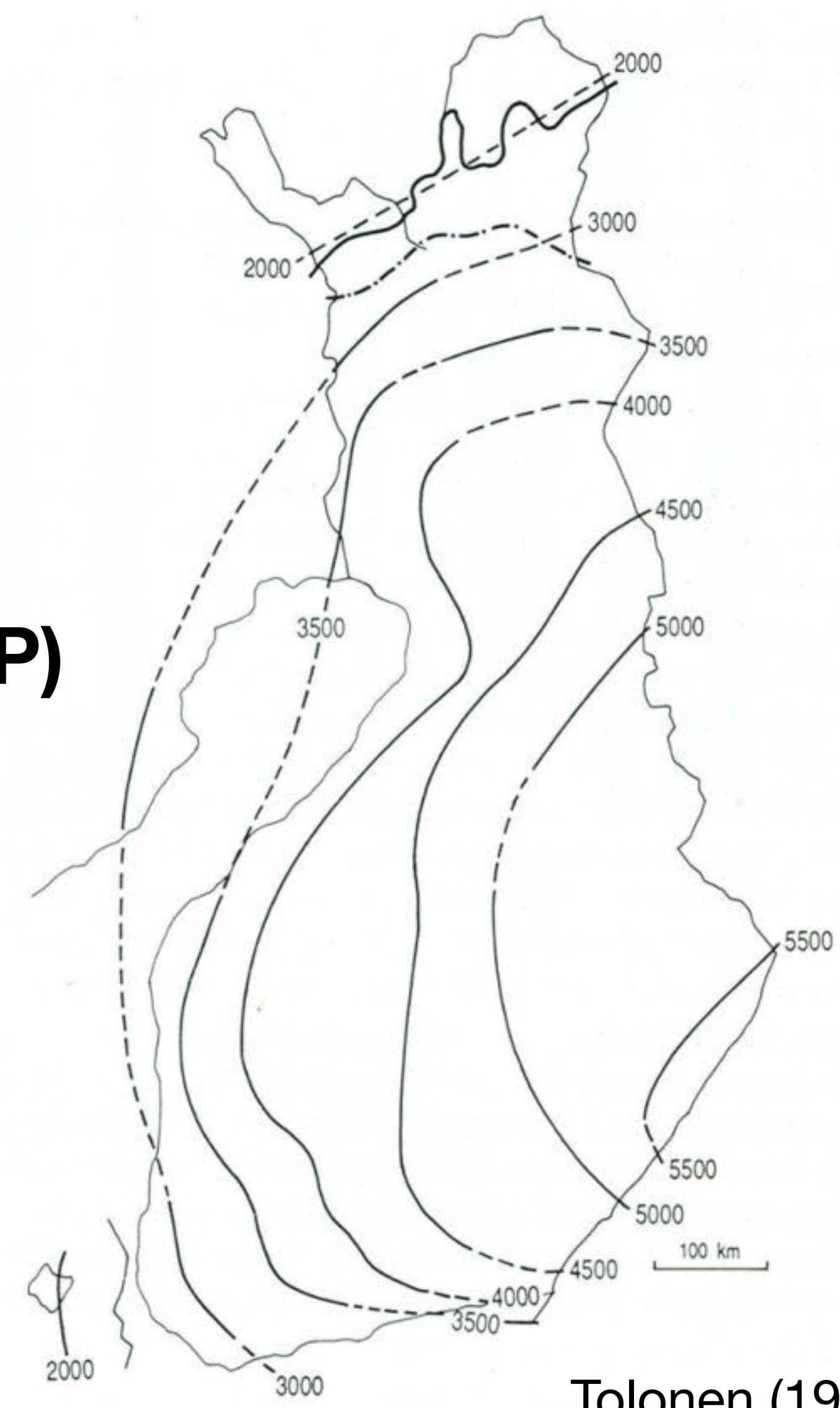
- Some **species (dis-)appear at known times** / events
- Changes vegetation mark temporal episodes / sediment layers (esp. trees)
- Caveat: species timing varies by region, transport can blur signals



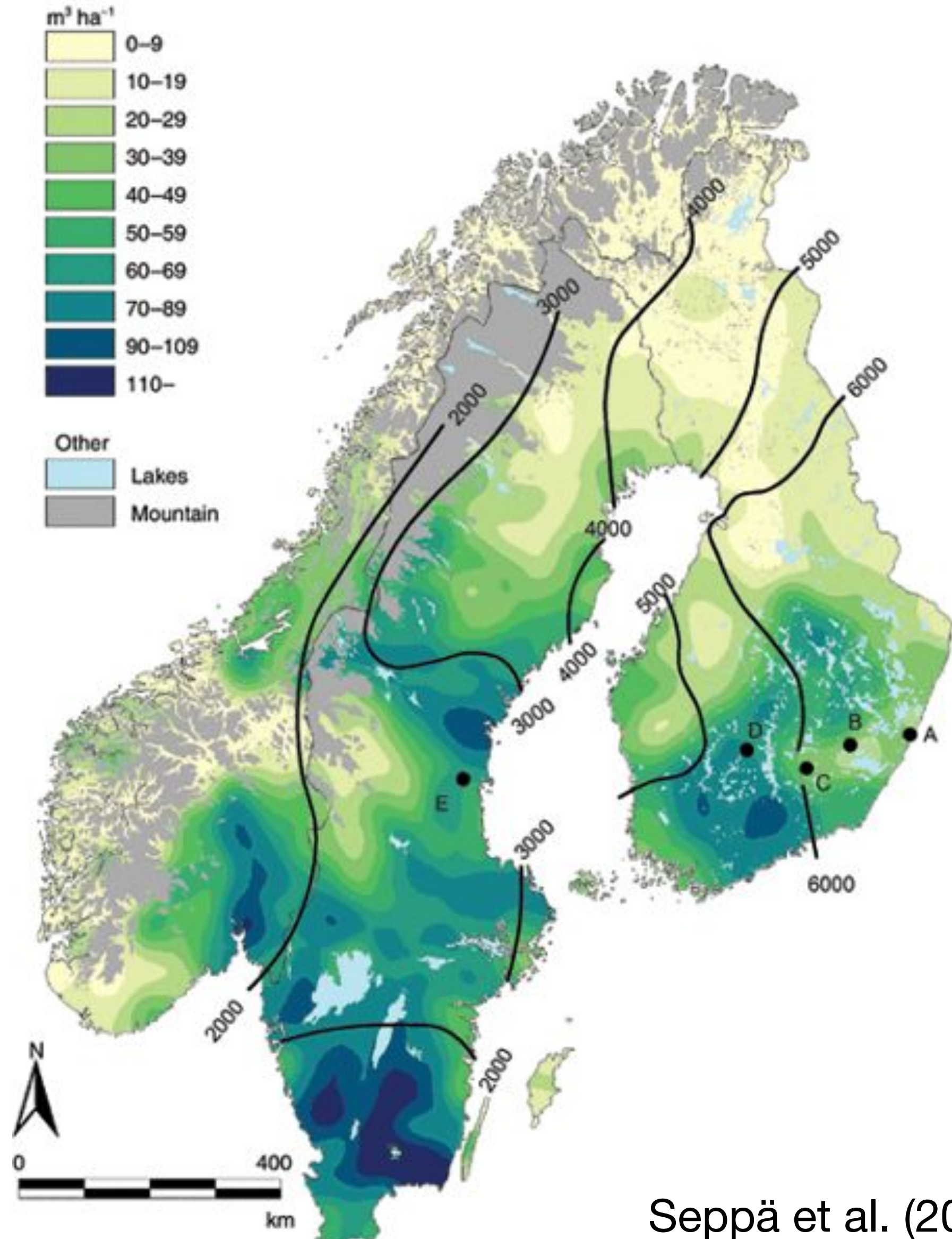


Fossils

Spruce expansion (yBP)



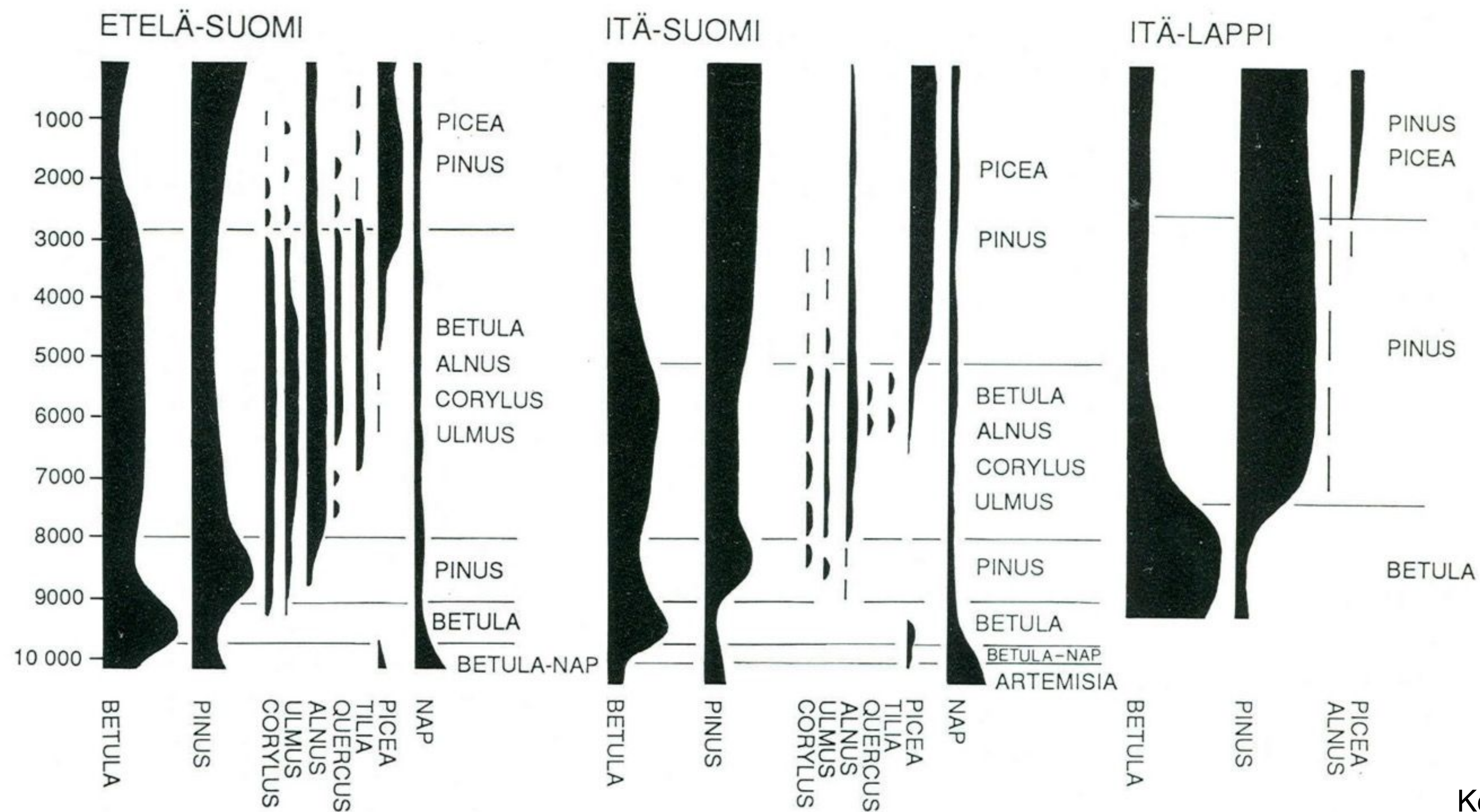
Tolonen (1983)



Seppä et al. (2009)

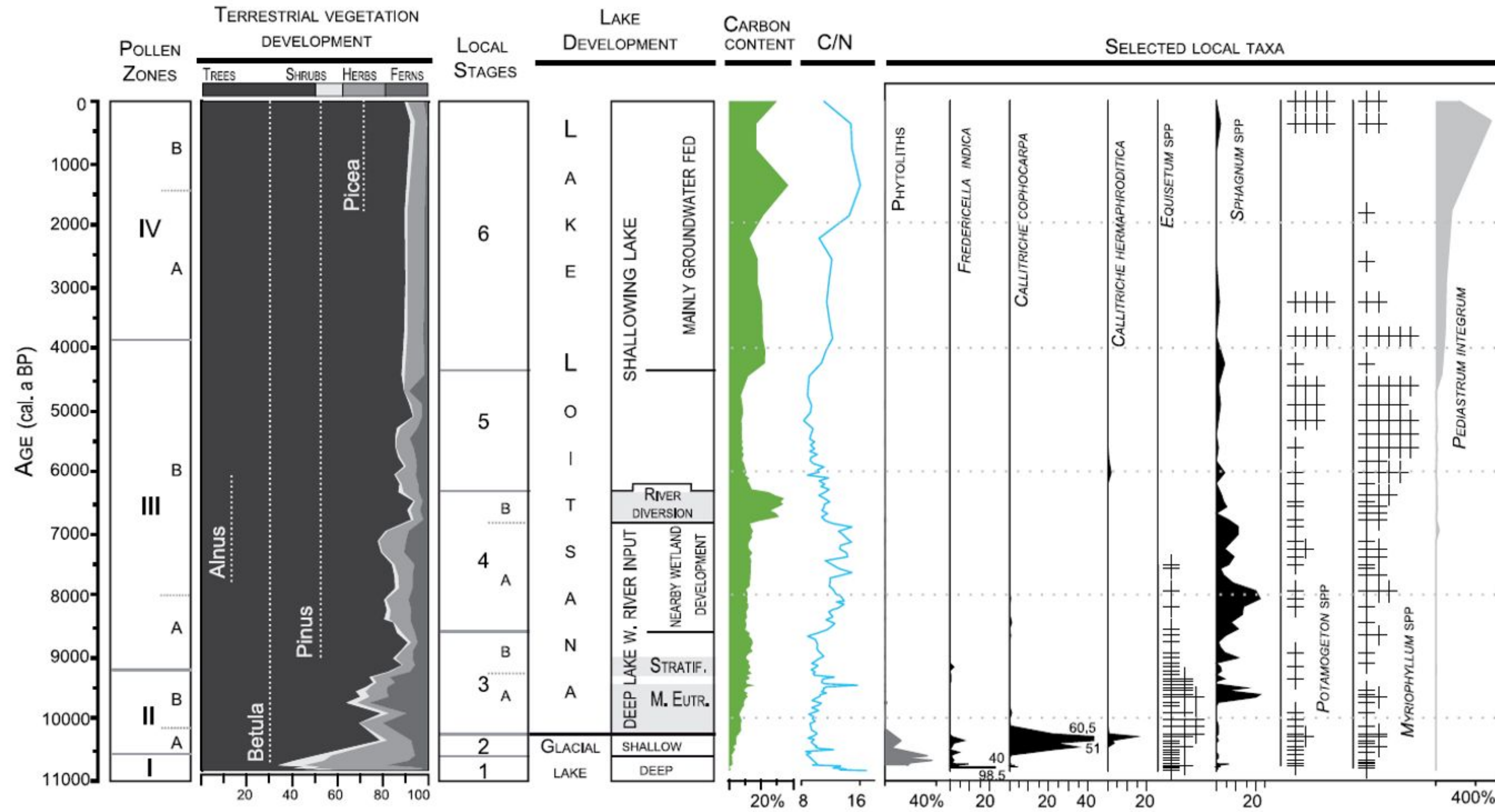


Fossils





Fossils

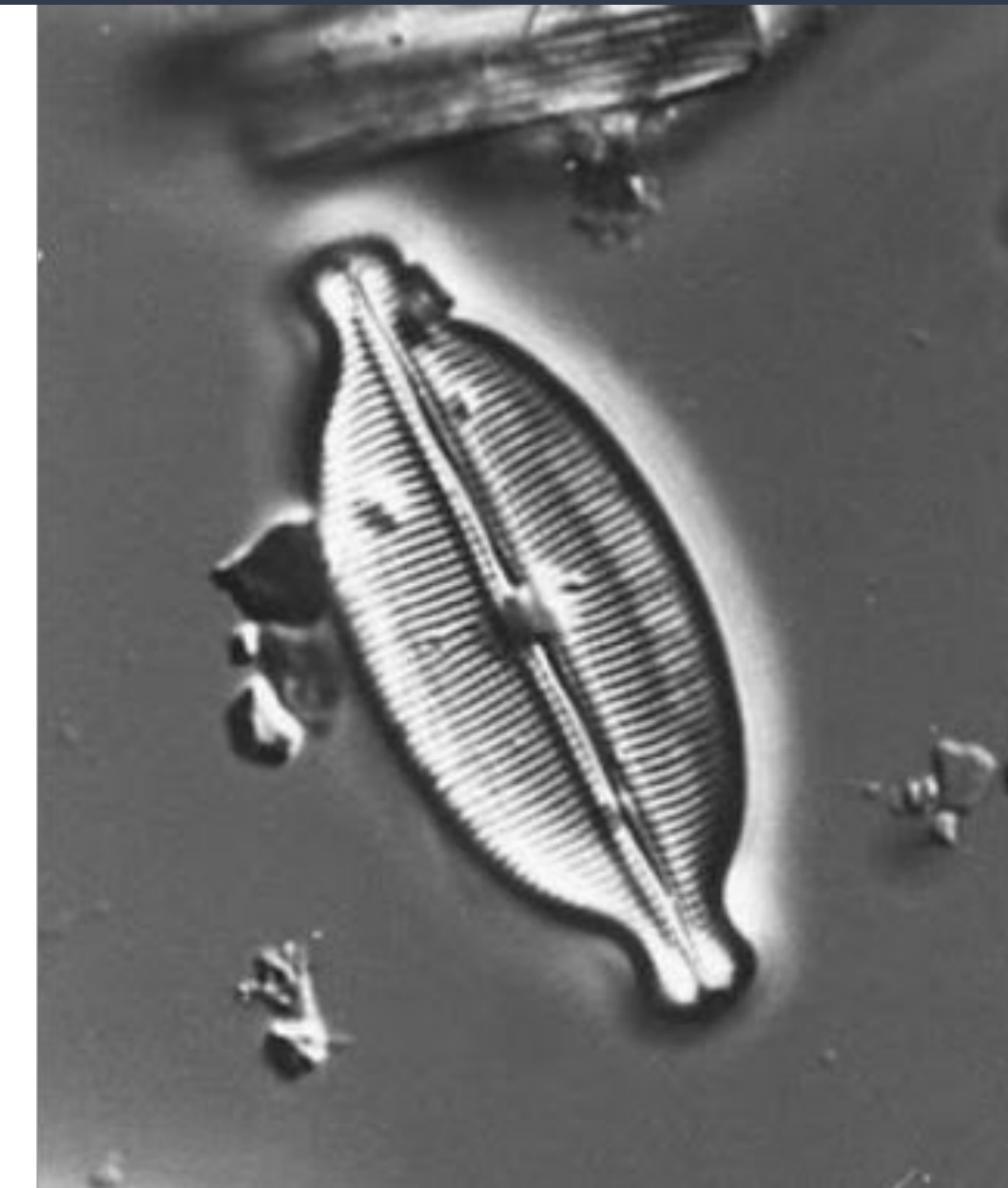




Fossils

Diatoms (silica algae)

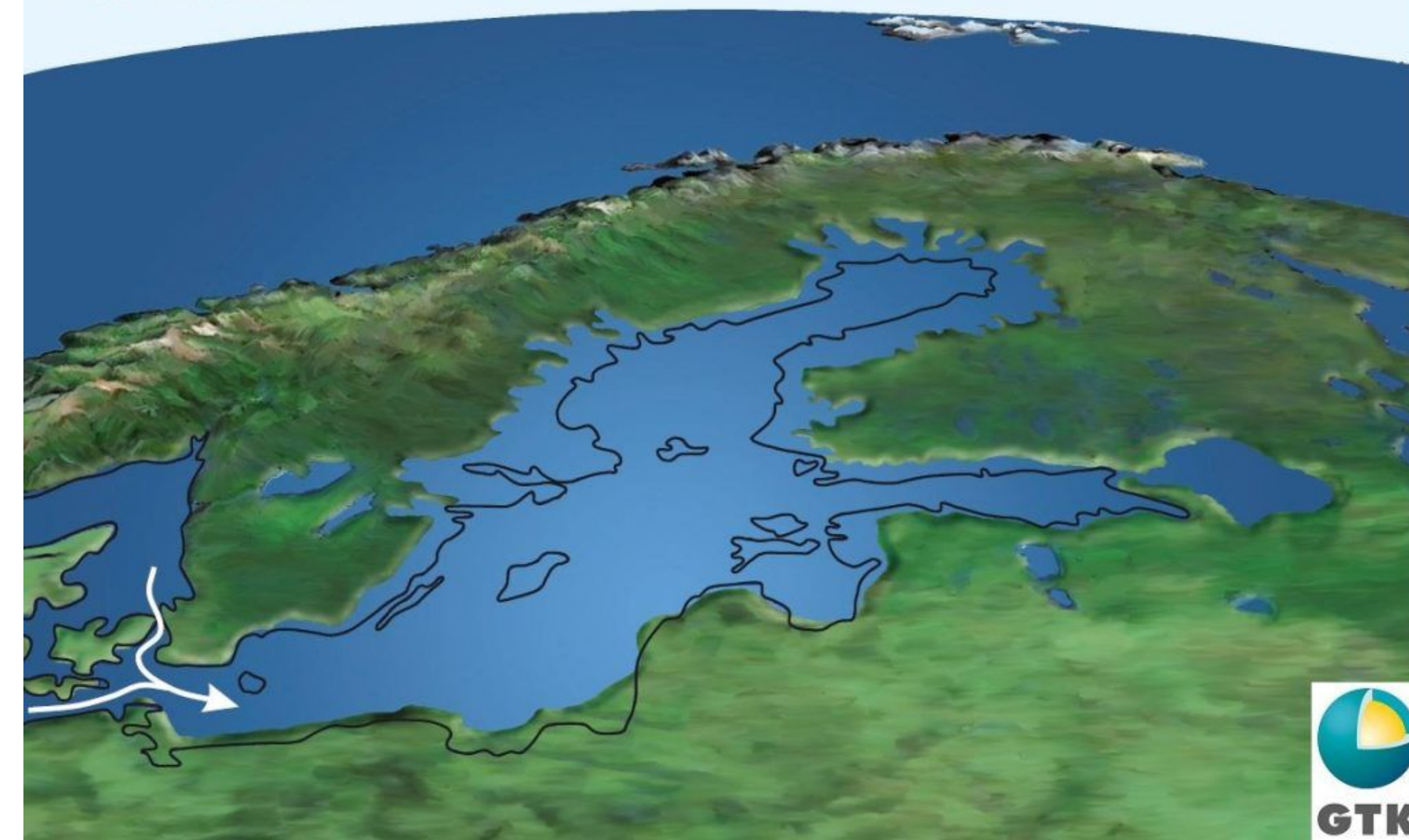
- Preserve well in sediments
- **Different species prefer different conditions** (salinity, nutrients, pH)
- Big environmental shifts can cause diatom “turnovers”



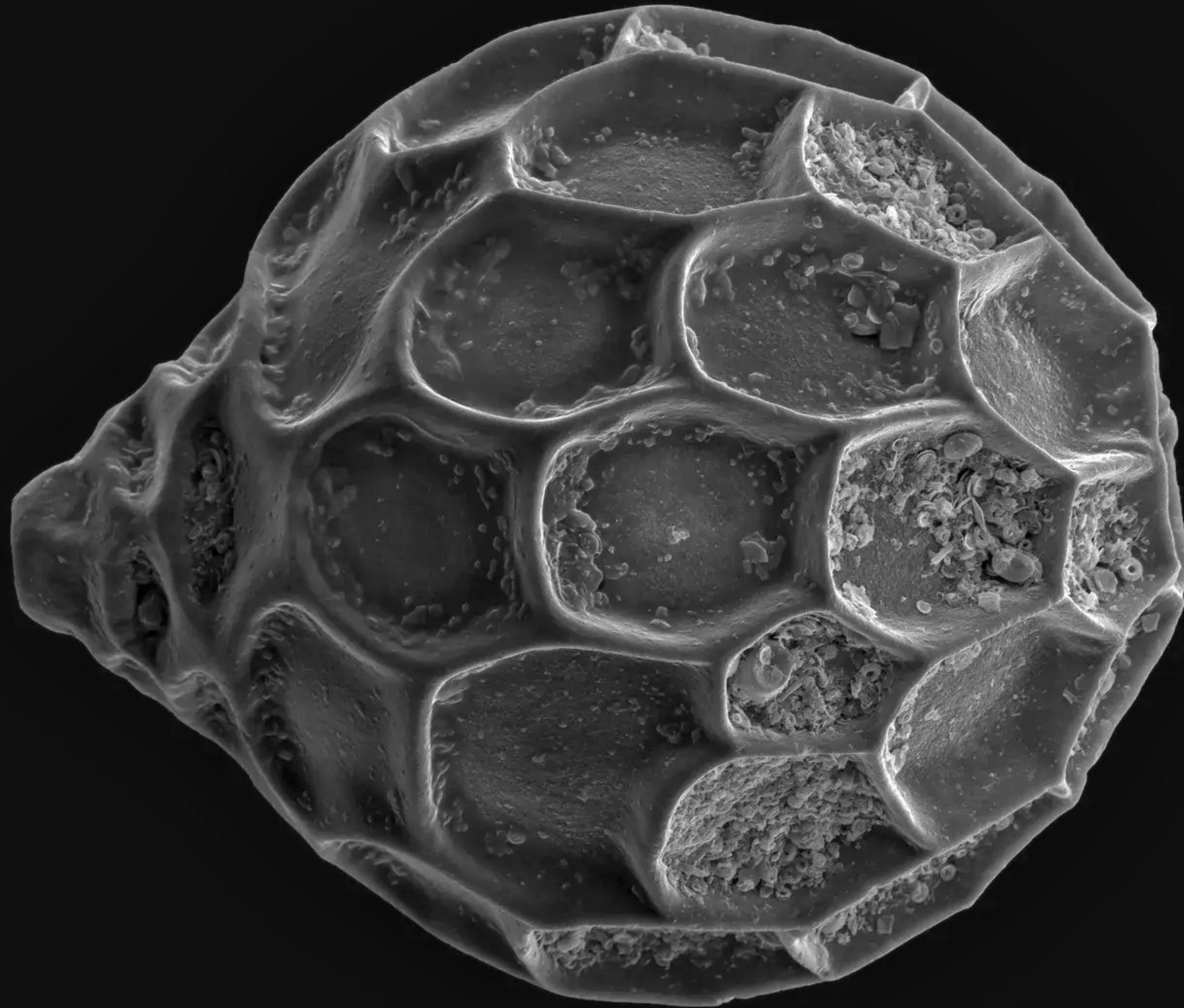
Litorinameri



Mastogloia Baltica



$\delta^{18}\text{O}$



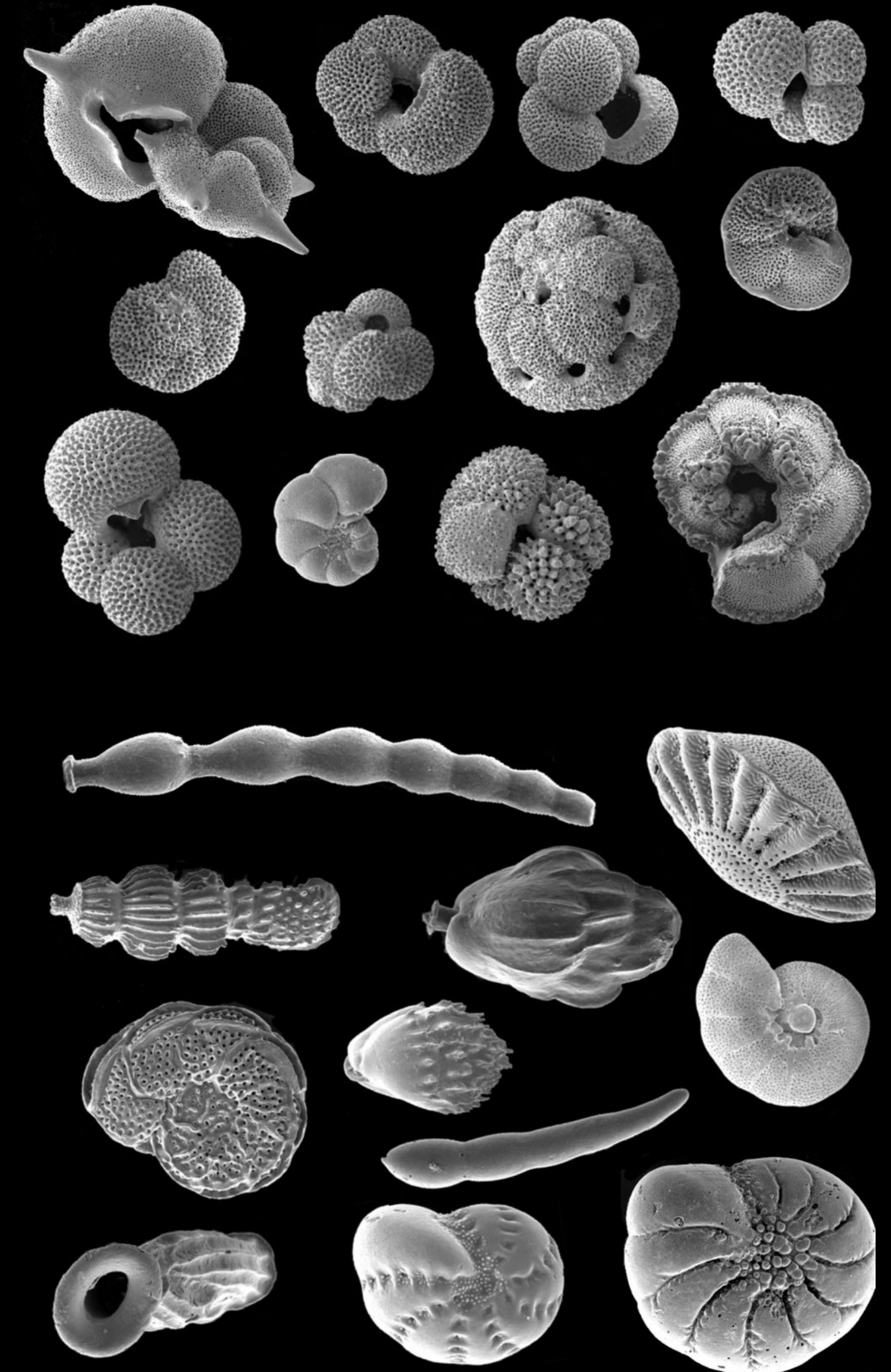
Foraminifera species *Favulina hexagona*



$\delta^{18}\text{O}$

Foraminifera:

- Single-celled marine protists with **external shell**
- First fossils date to the Cambrian ($541 \pm 1\text{Ma BP}$)
- **Chemistry of shell reflects seawater chemistry**

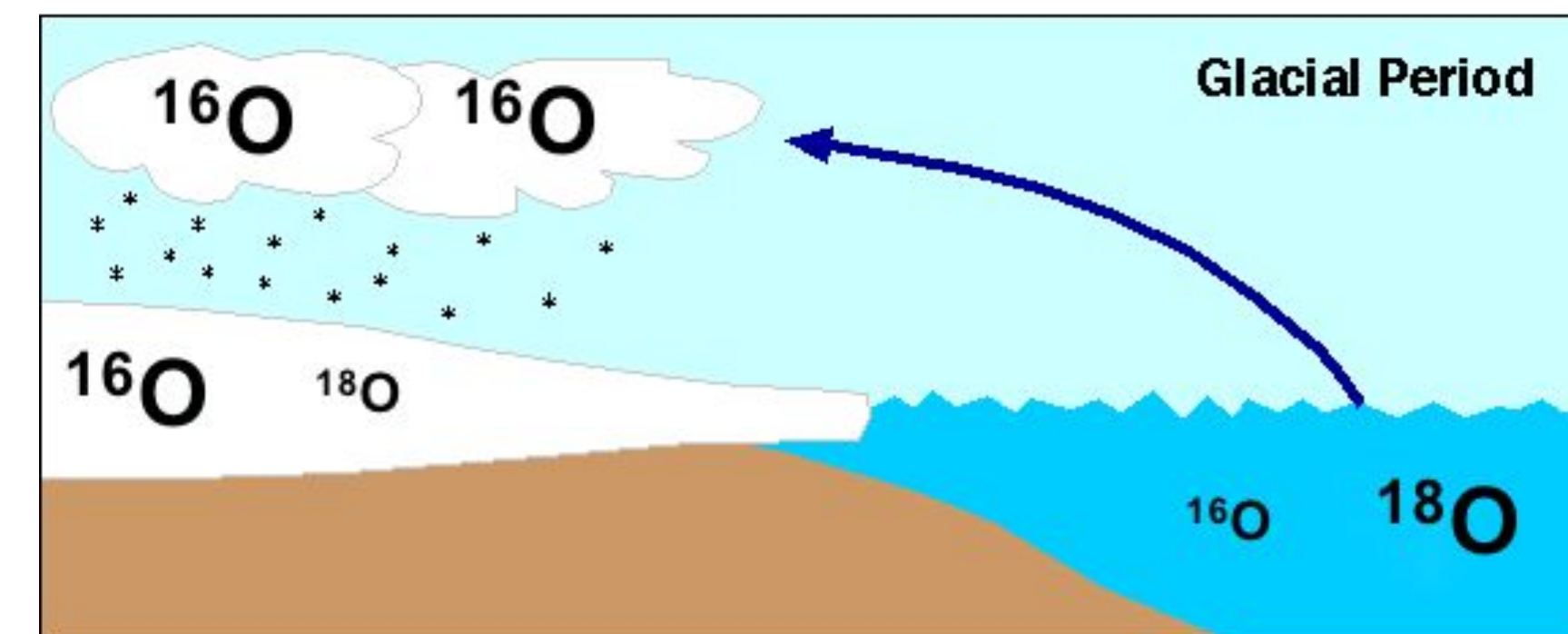
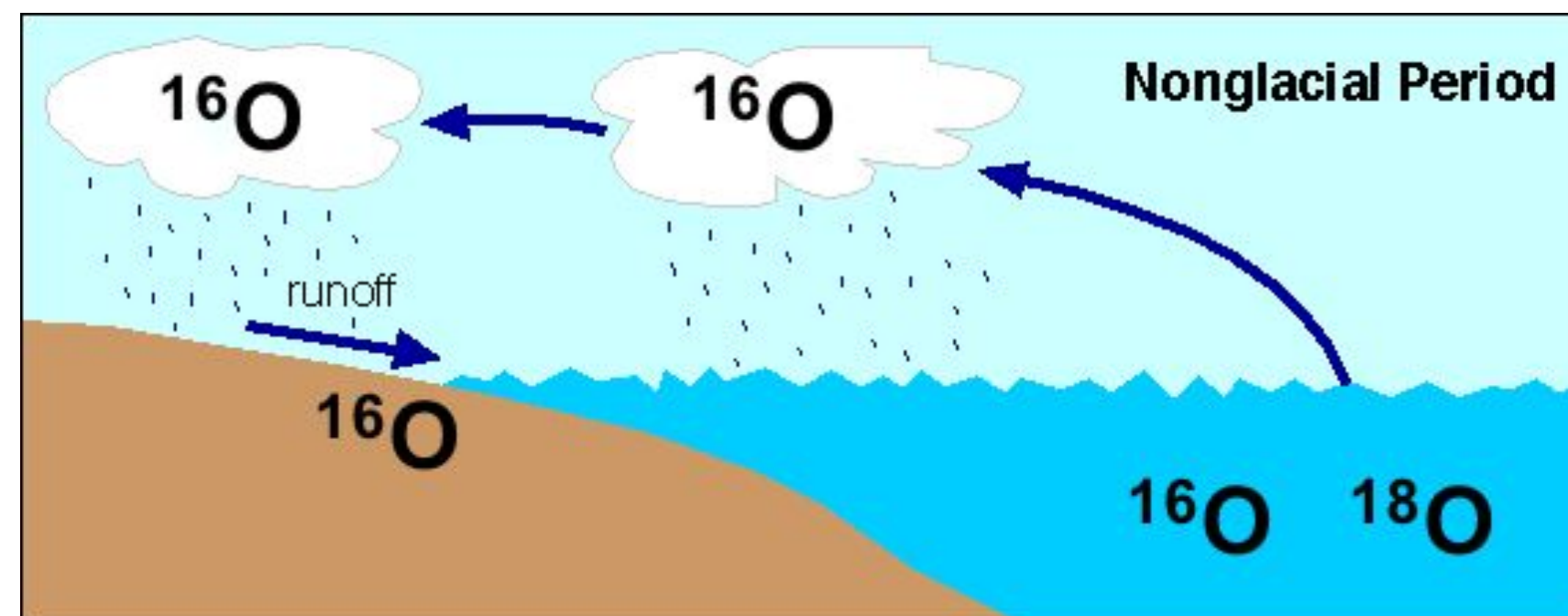
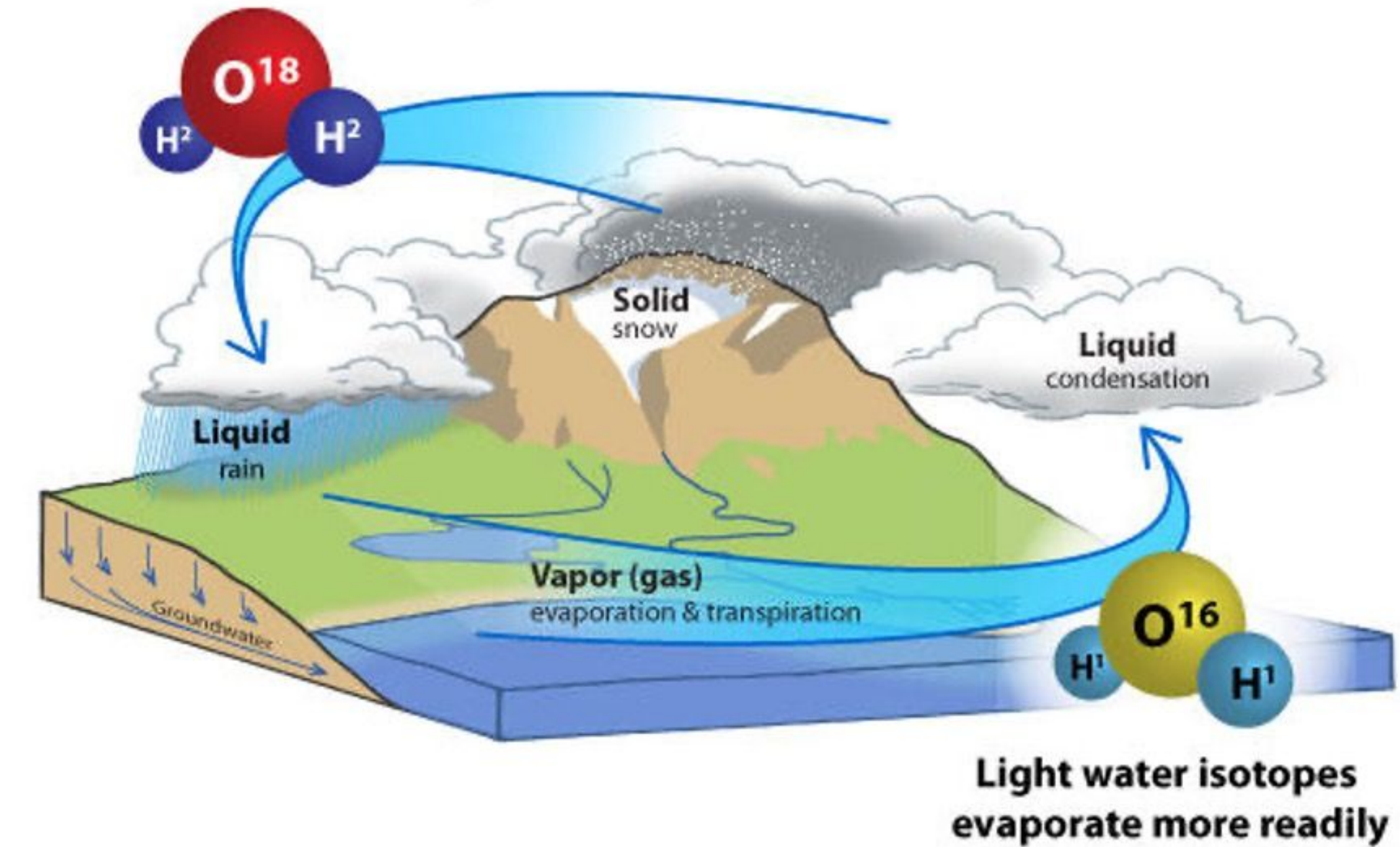




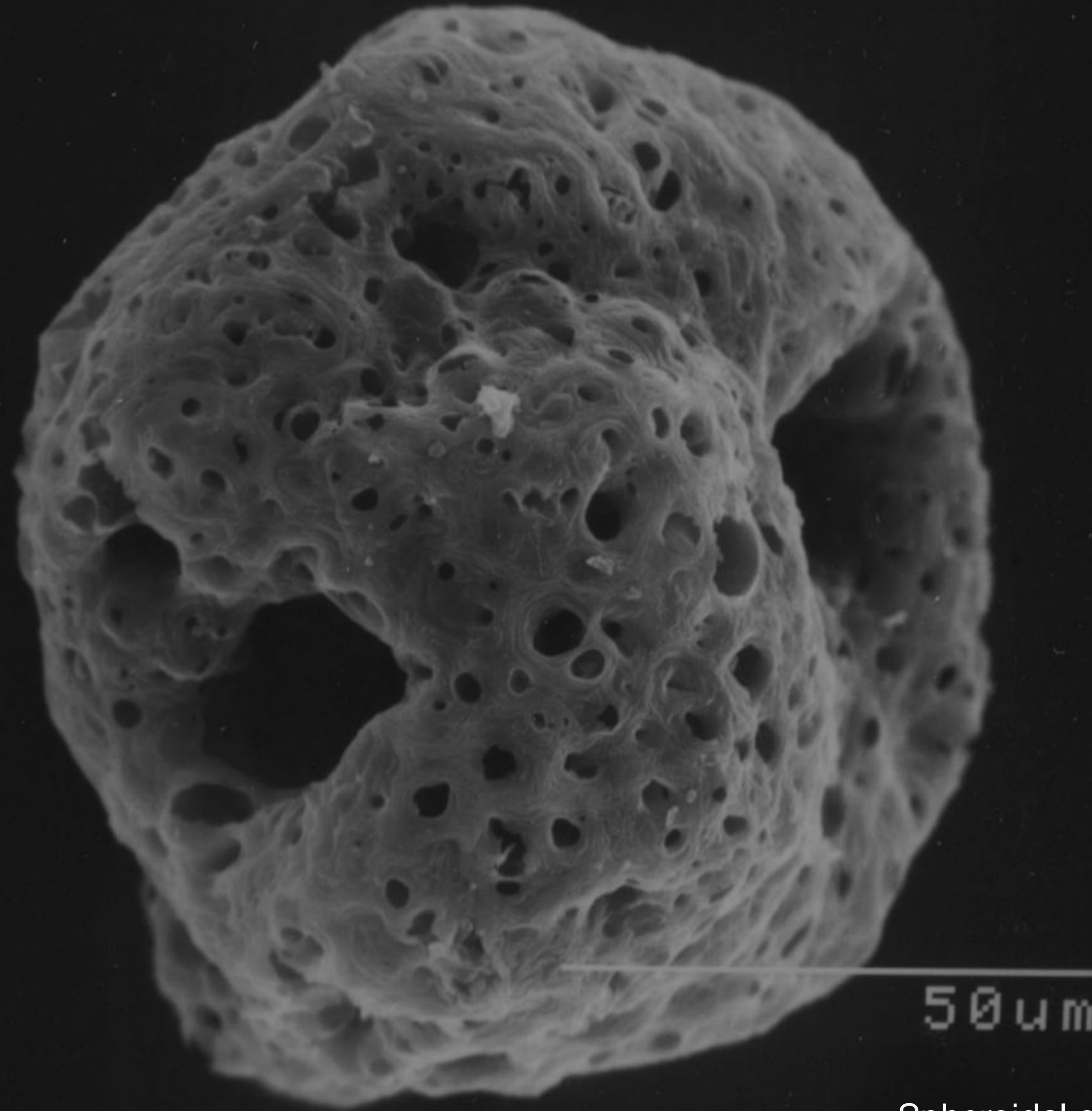
$\delta^{18}\text{O}$

- Shell $\delta^{18}\text{O}$ records seawater + temperature
- **higher $\delta^{18}\text{O}$: more global ice / colder**
- Produces **glacial-interglacial pattern** in marine sediments

Heavy water isotopes rain and snow more readily



SCPs

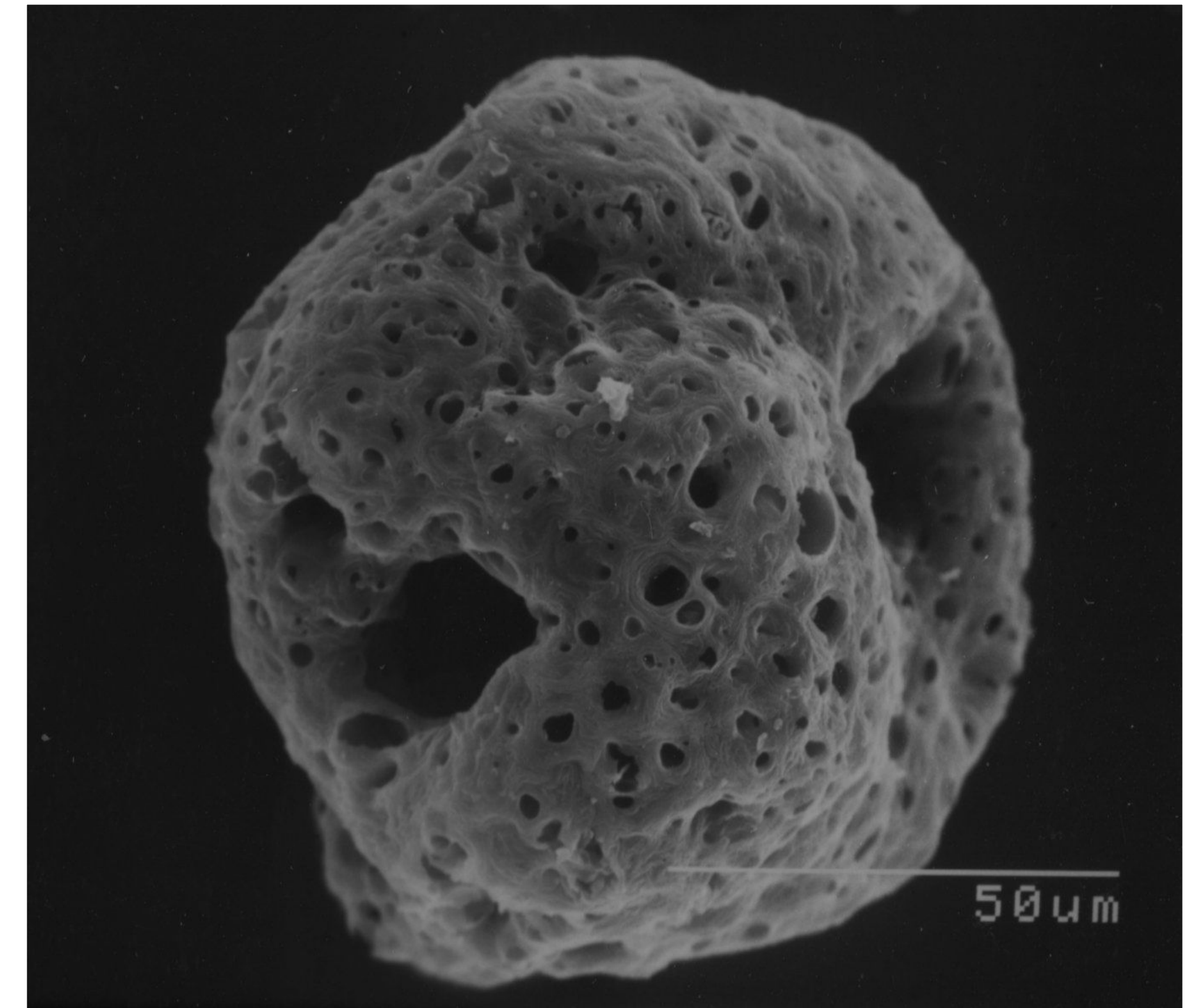


Spheroidal carbonaceous fly-ash particle



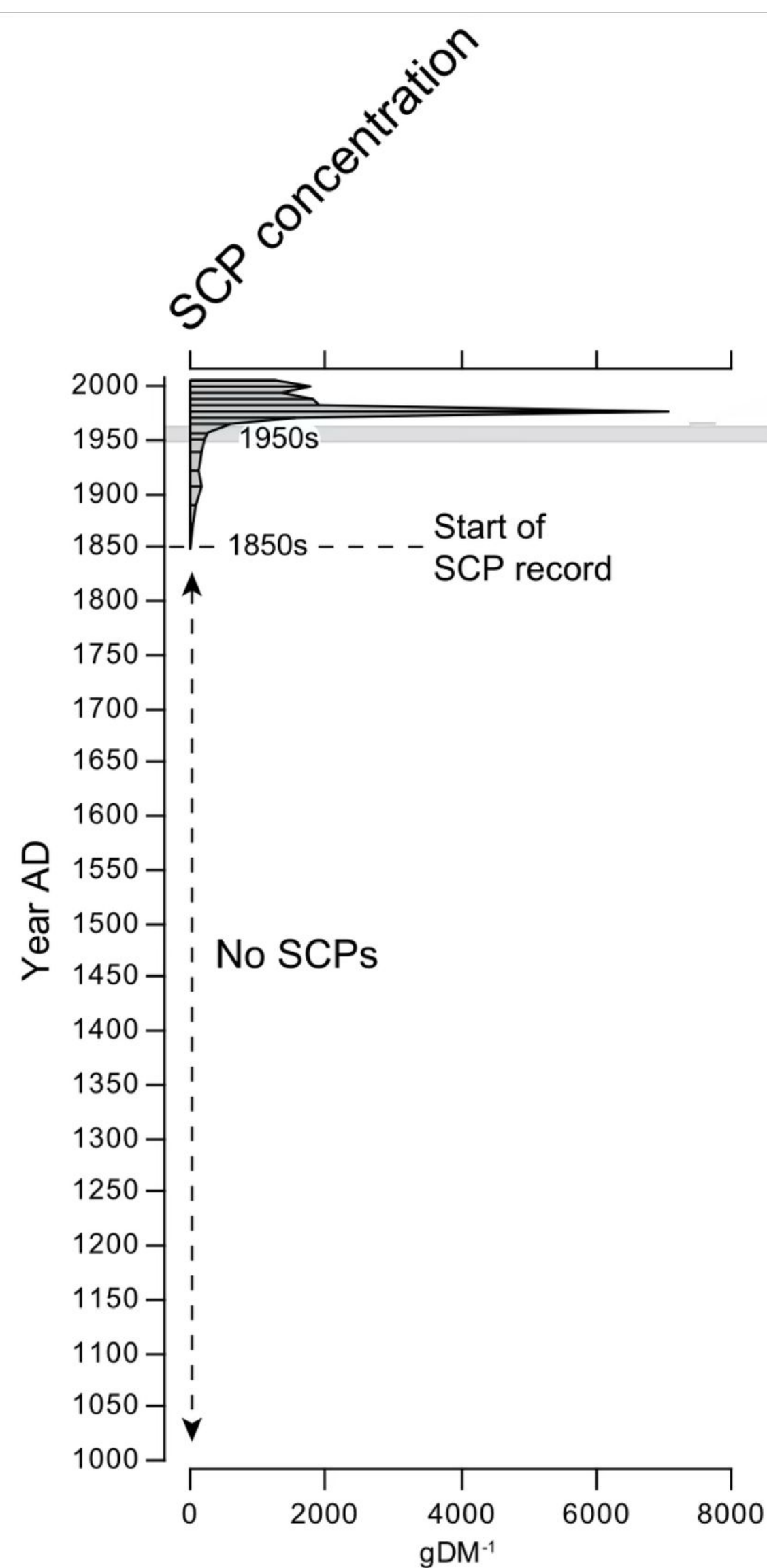
SCPs

- Spheroidal Carbonaceous Particles
- **Formed during** high-temperature **fossil-fuel burning**
- Virtually **absent before industrialization** → “human-era” marker

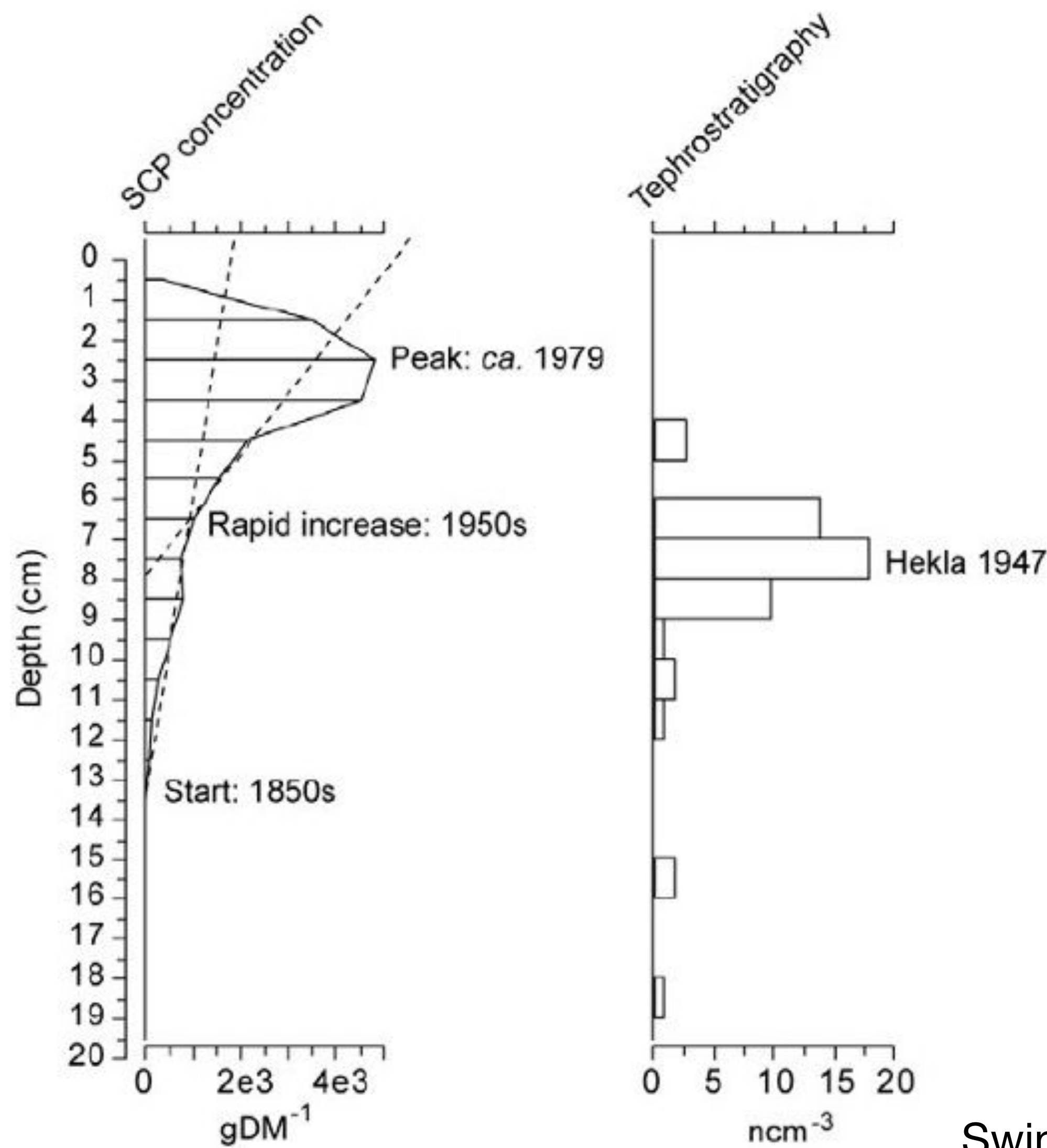




SCPs



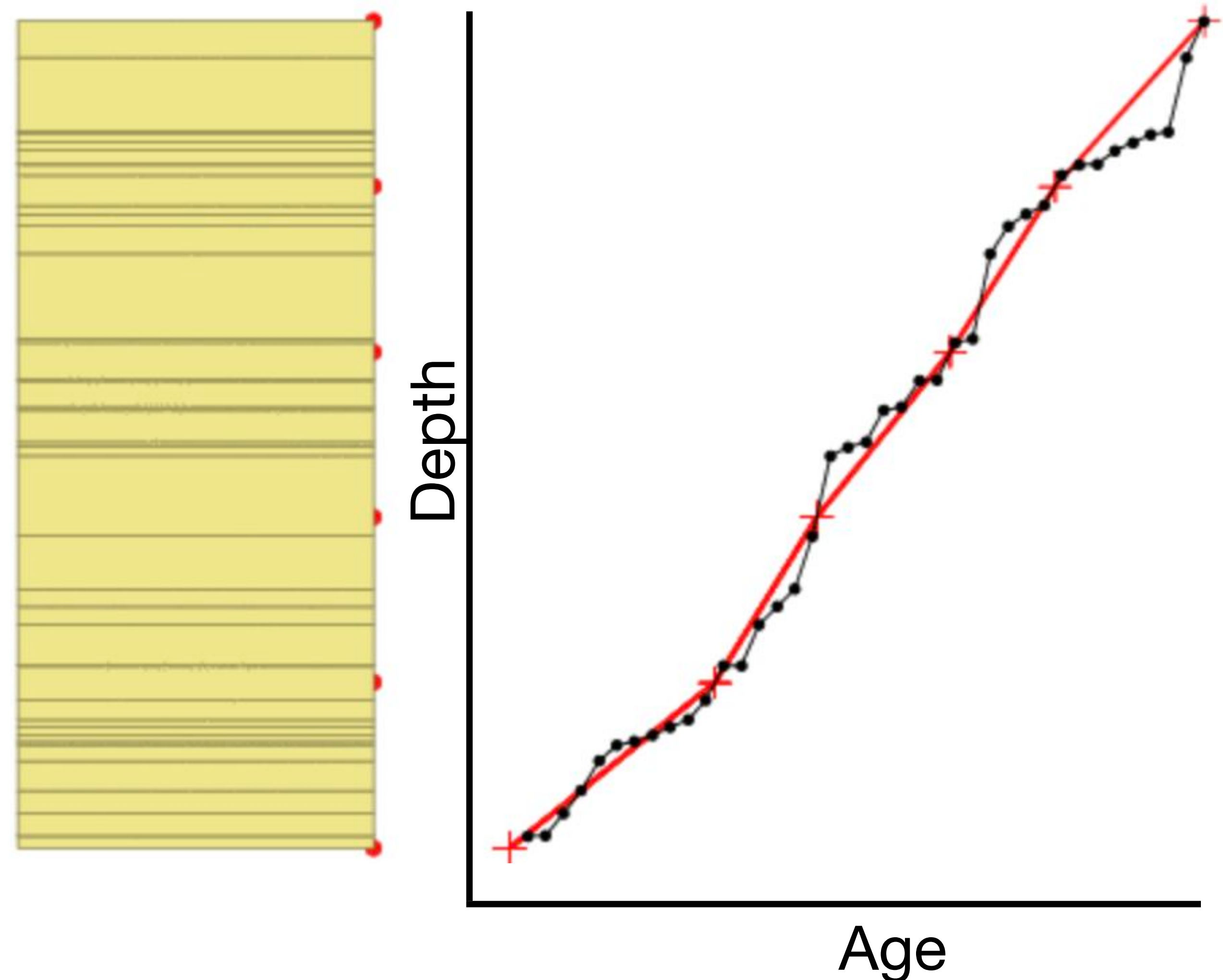
Swindles et al. (2015)



Swindles (2010)

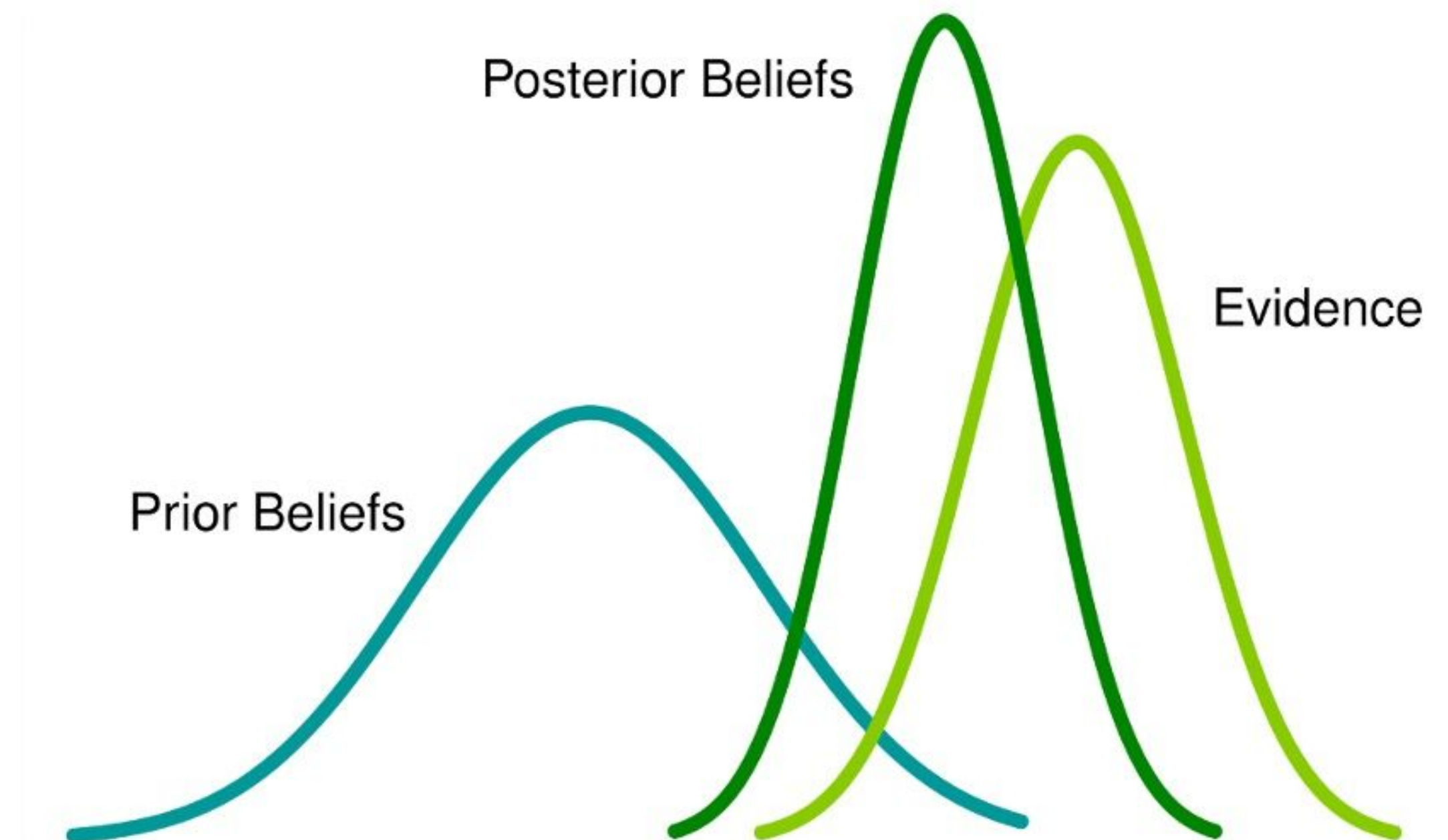
Age-Depth Models

- Dating every slice is impractical
→ **only a few depths are dated**
- Ages and uncertainties needed for every depth
- Sedimentation rate changes
→ can't assume straight lines

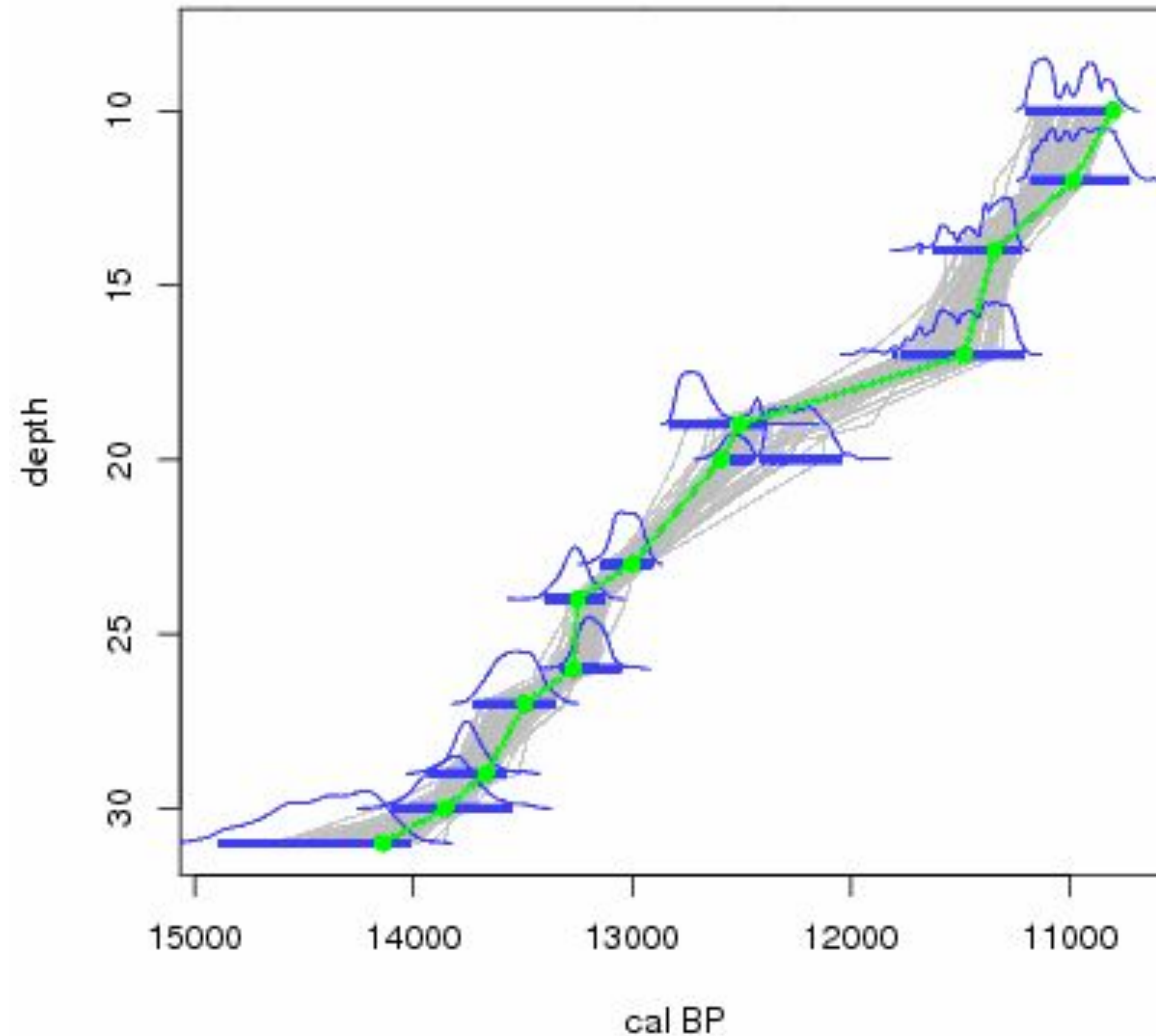


Age-Depth Models - Bayesian Statistics

- **Bayes' law:** The probability of an event is given by the existing evidence
- **Bayesian models:** Estimate most probable order of events, given the measurements and prior knowledge
- How likely is age-depth relationship based on:
 - ^{14}C dates (radiocarbon age)
 - Prior knowledge of atmospheric ^{14}C variations, sediment accumulation rate, ranges and variation...



Age-Depth Models - Bayesian Statistics



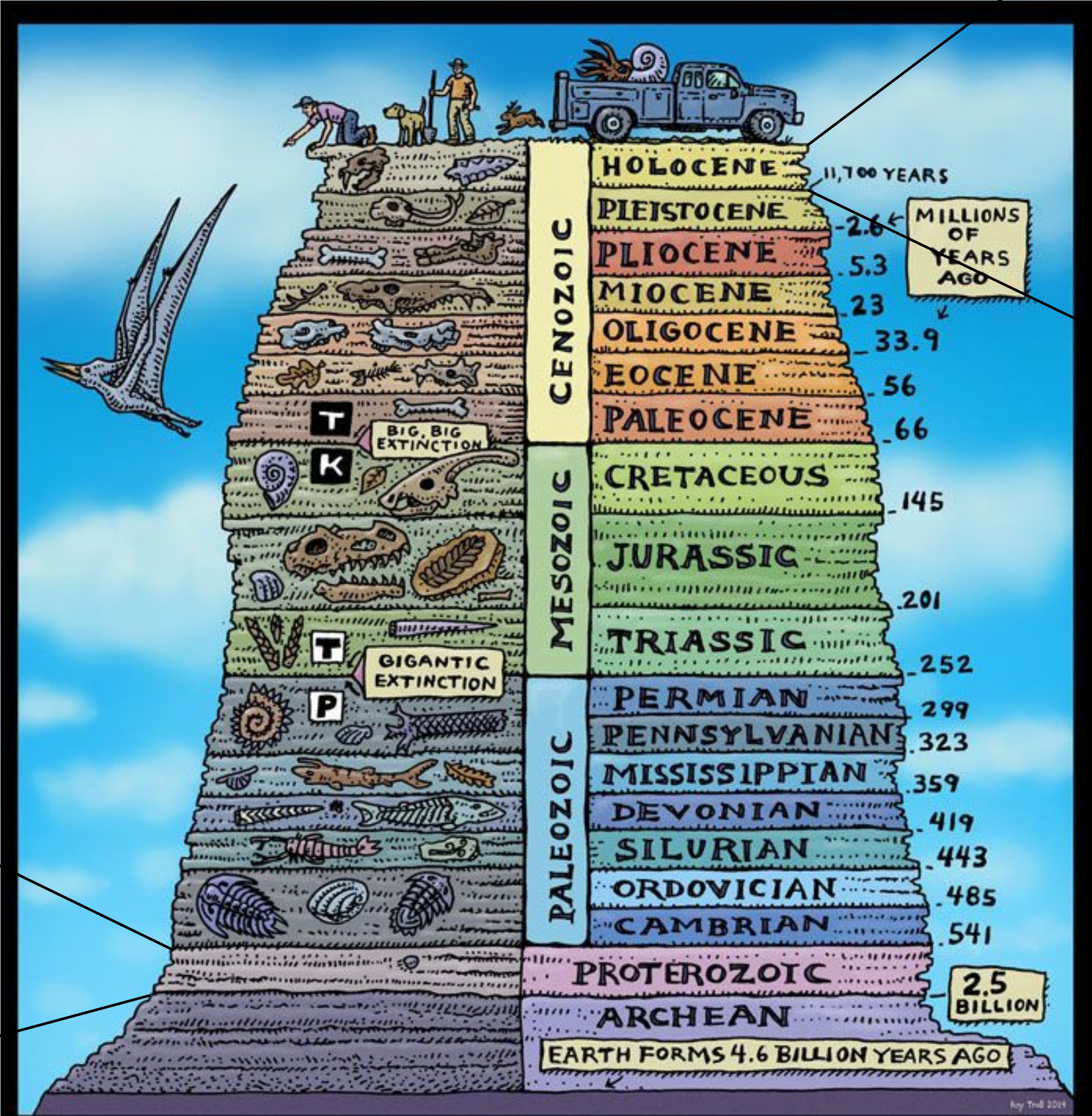
Which Method To Choose?

Peatland

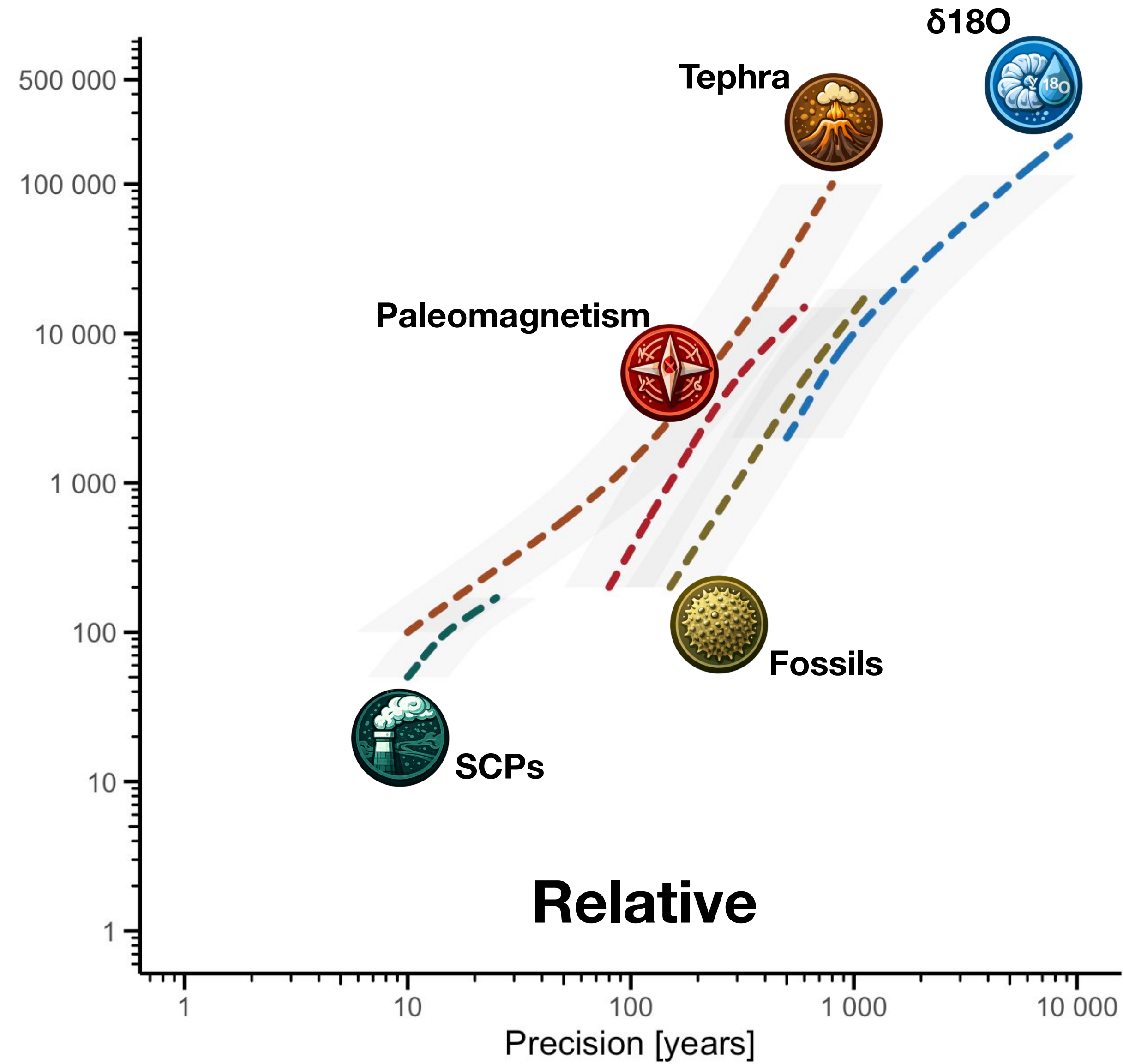
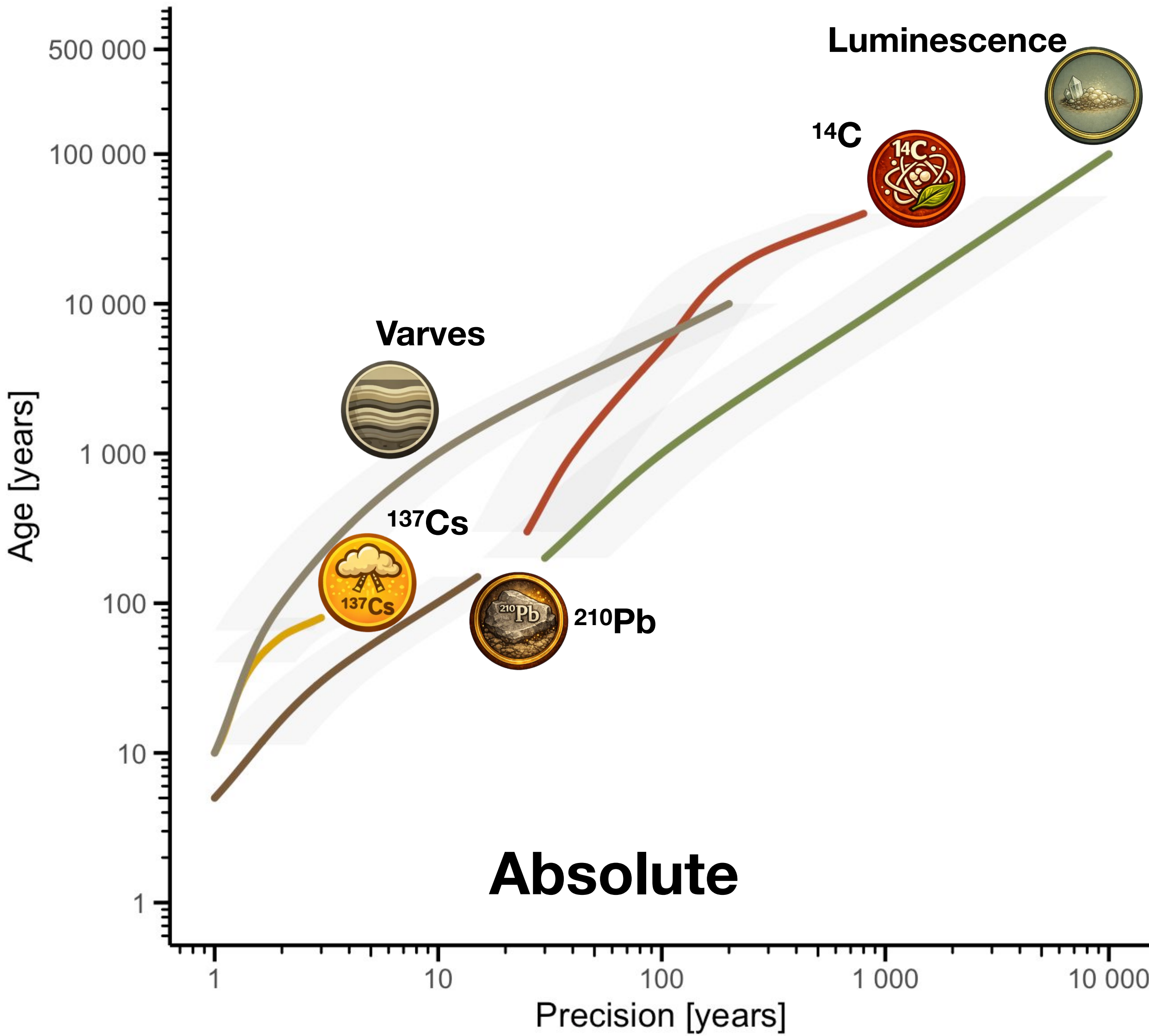
Lacustrine

Marine

Which Method To Choose?



Which Method To Choose?





^{210}Pb



^{14}C



Fossils



Tephra



Varves

Which method(s) would you use to date the Kruunuvuorenlampi sediments?



SCPs



Paleomagnetism



Luminescence



$\delta^{18}\text{O}$



^{137}Cs